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**Engineering and  
Physical Sciences  
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# In Situ Biomineralisation for Groundwater Radionuclide Remediation

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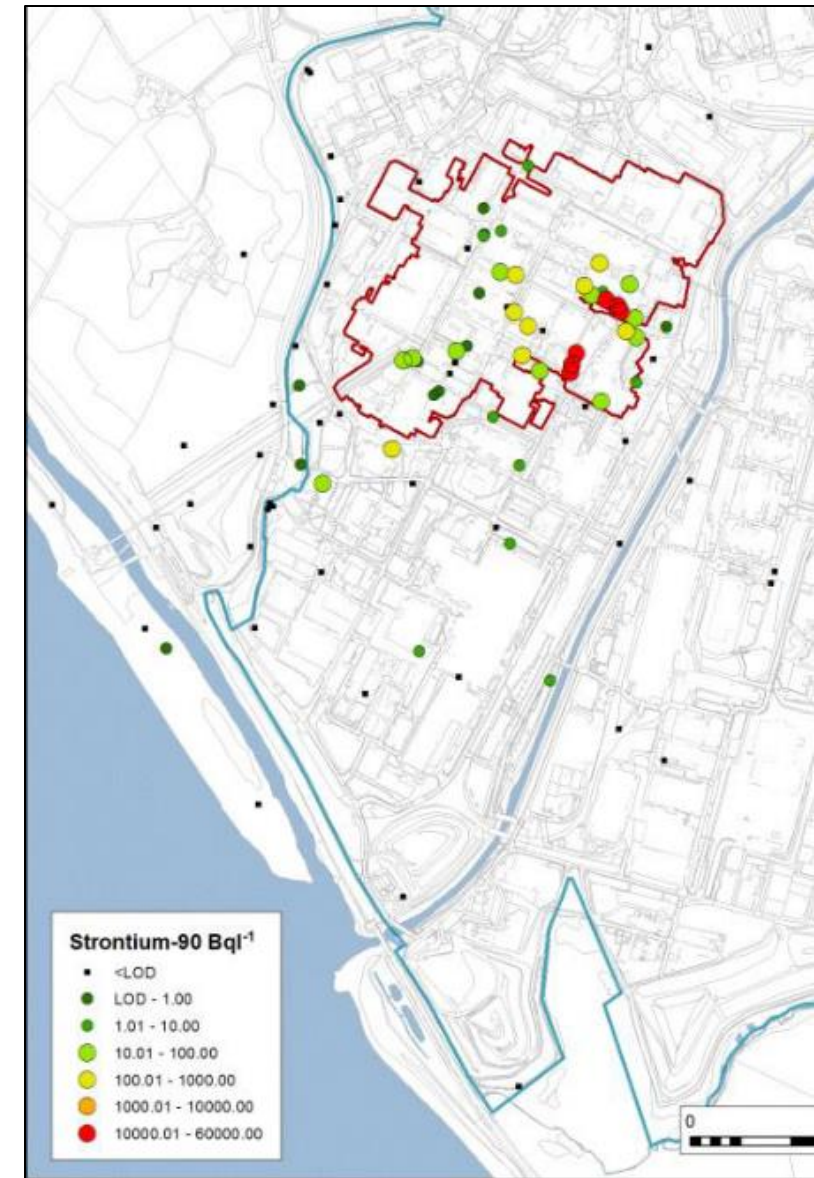
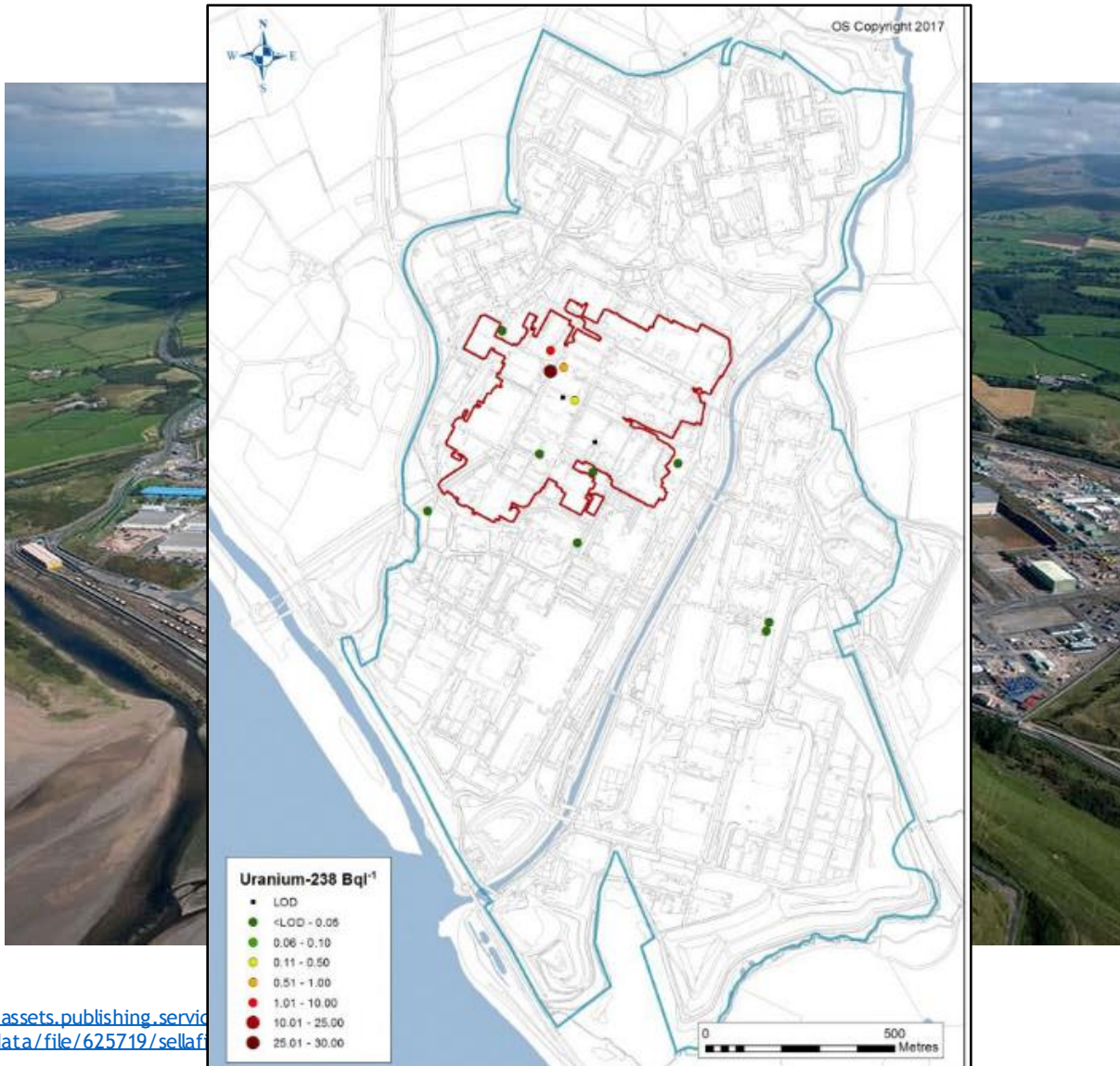
# Radionuclide Contamination at Sellafield



- ▶ Sellafield has legacy storage facilities and silos from historical operations
- ▶ Significant radionuclide contamination of the subsurface
- ▶ Focus of this work  $^{90}\text{Sr}$  and U



# Radionuclide Contamination at Sellafield

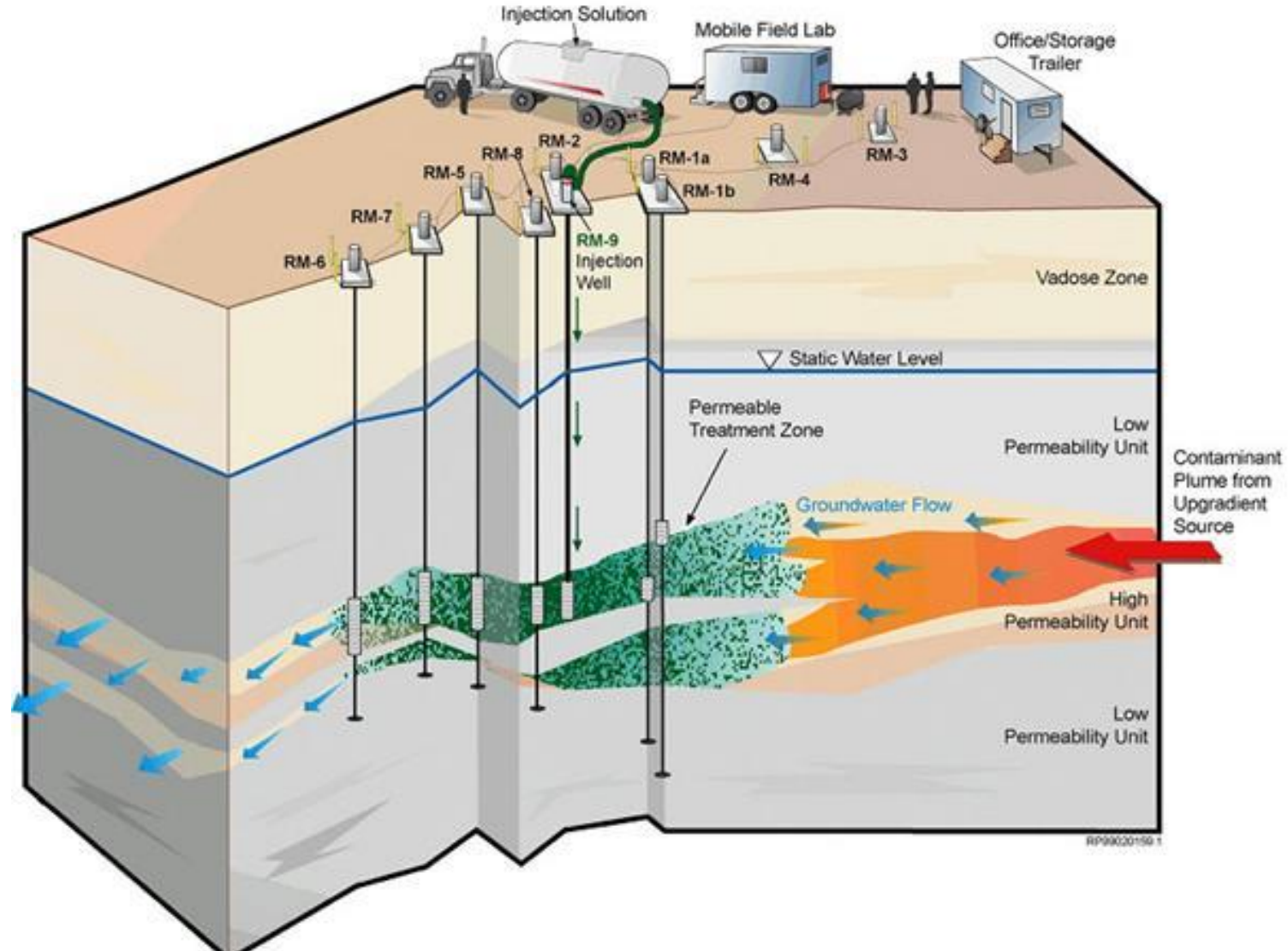


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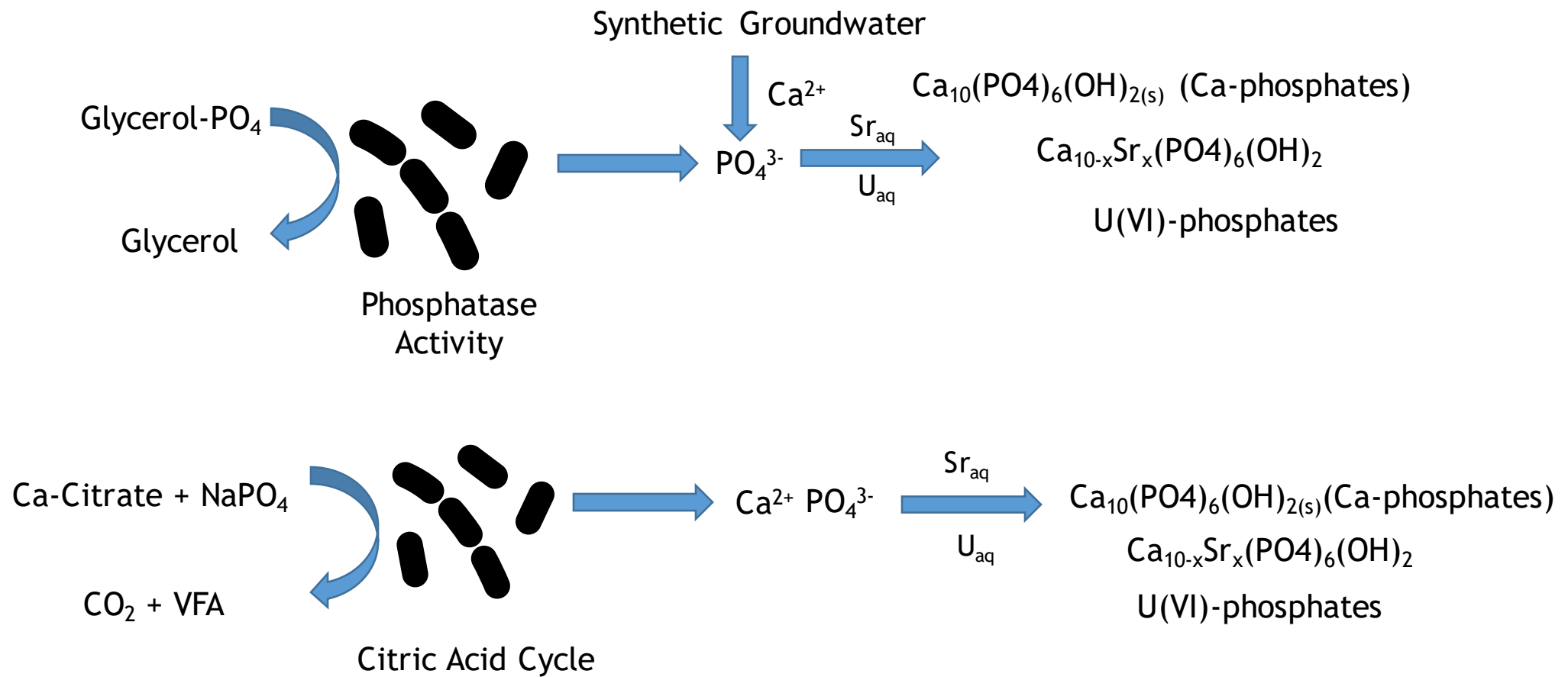
Groundwater monitoring at Sellafield: Annual Data review 2016 Sellafield Ltd, 2016

# Phosphate Mineralisation

- Phosphate containing minerals such as hydroxyapatite (HAp) have been shown to incorporate  $^{90}\text{Sr}$  within their structure
- Uranium phosphates are highly insoluble mineral phases
- Direct injection of phosphate containing solutions into the subsurface can lead to injection well clogging
- In situ, slow release phosphate delivery is promising for co-treatment of  $\text{Sr}^{2+}$  and  $\text{U}^{\text{VI}}\text{O}_2^{2+}$



# Biomineralisation





# Sediment Characterisation and Synthetic Groundwater

Montage from centre of Active Face of Peel Place Quarry (NW BAY)



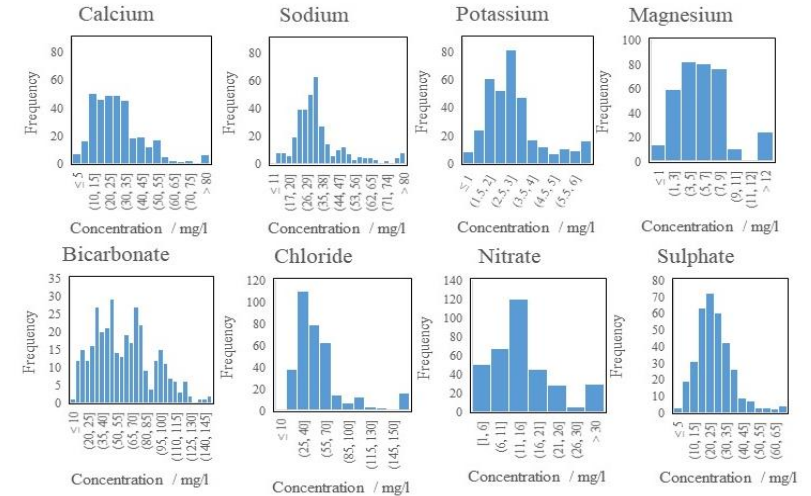
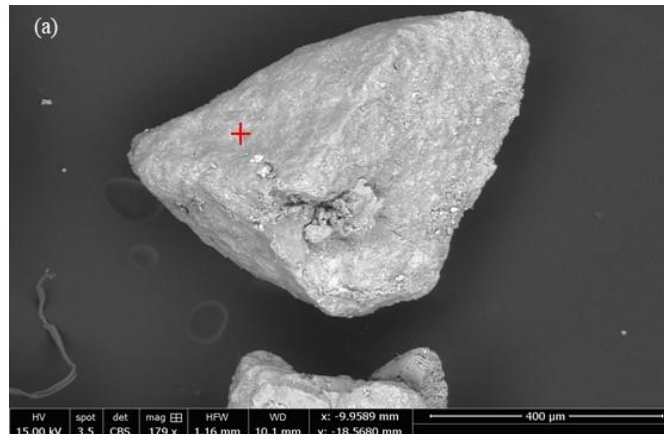
Location 1

Gravel/Topsoil/Spoil

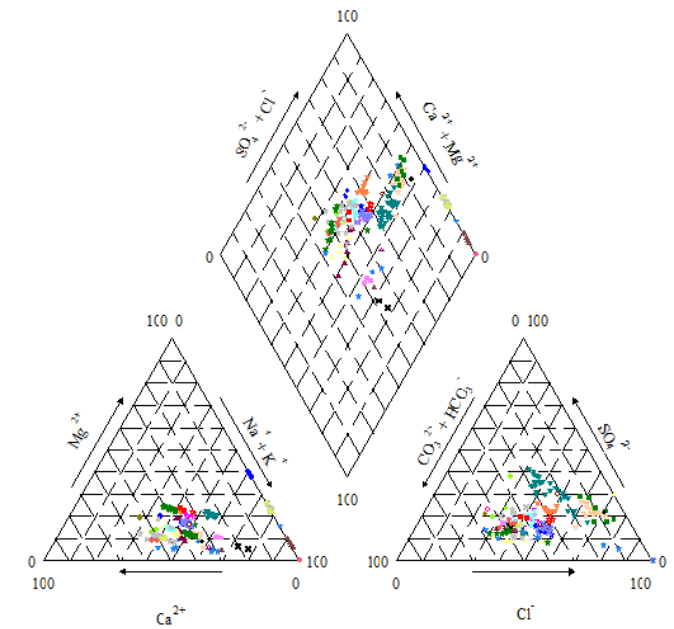
Erosional contact

Coarse heterogeneous sand

Homogeneous brown sand (Material characterised and used within microcosms experiments)



Sr Plume Groundwater Piper Plot



# Biom mineralisation of Sr

- Peel Place Quarry (PPQ) Sediment
- Calder River (CR) Sediment
- Oxidic Experiments
- Synthetic Groundwater
  - 1mM (88ppm) Sr Spiked
  - Calcium citrate/sodium phosphate
  - Glycerol phosphate



## Phosphate (Bio)mineralization Remediation of $^{90}\text{Sr}$ -Contaminated Groundwaters

Callum Robinson, Samuel Shaw, Jonathan R. Lloyd, James Graham, and Katherine Morris\*

Cite This: <https://doi.org/10.1021/acsestwater.3c00159>

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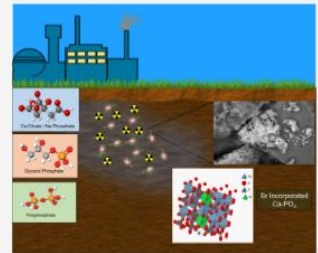
Metrics & More

Article Recommendations

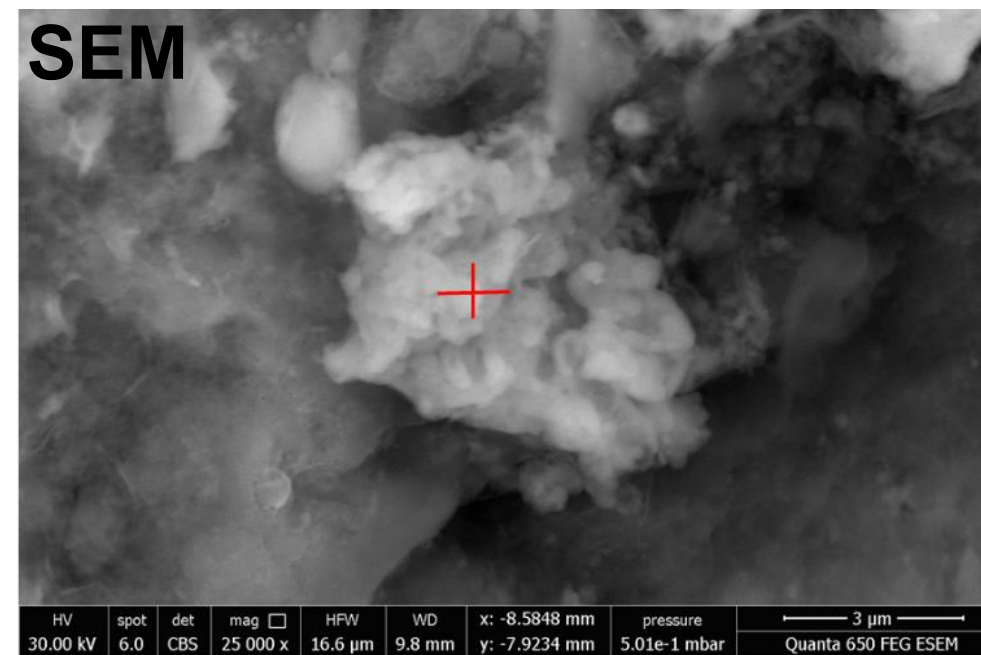
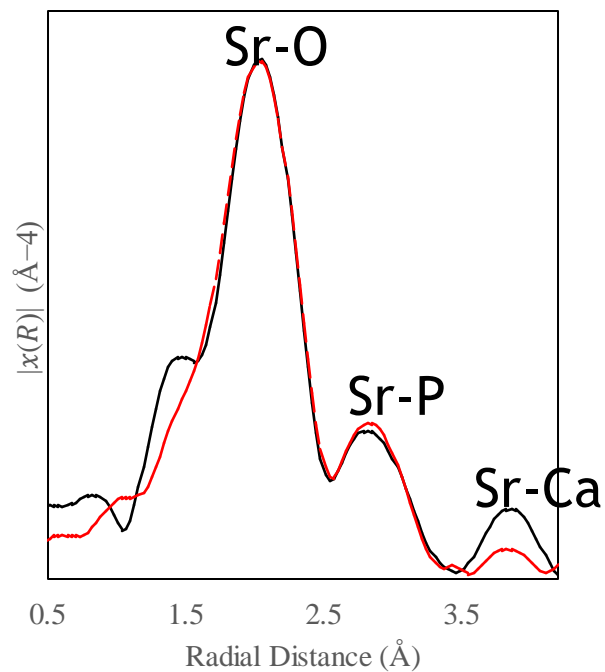
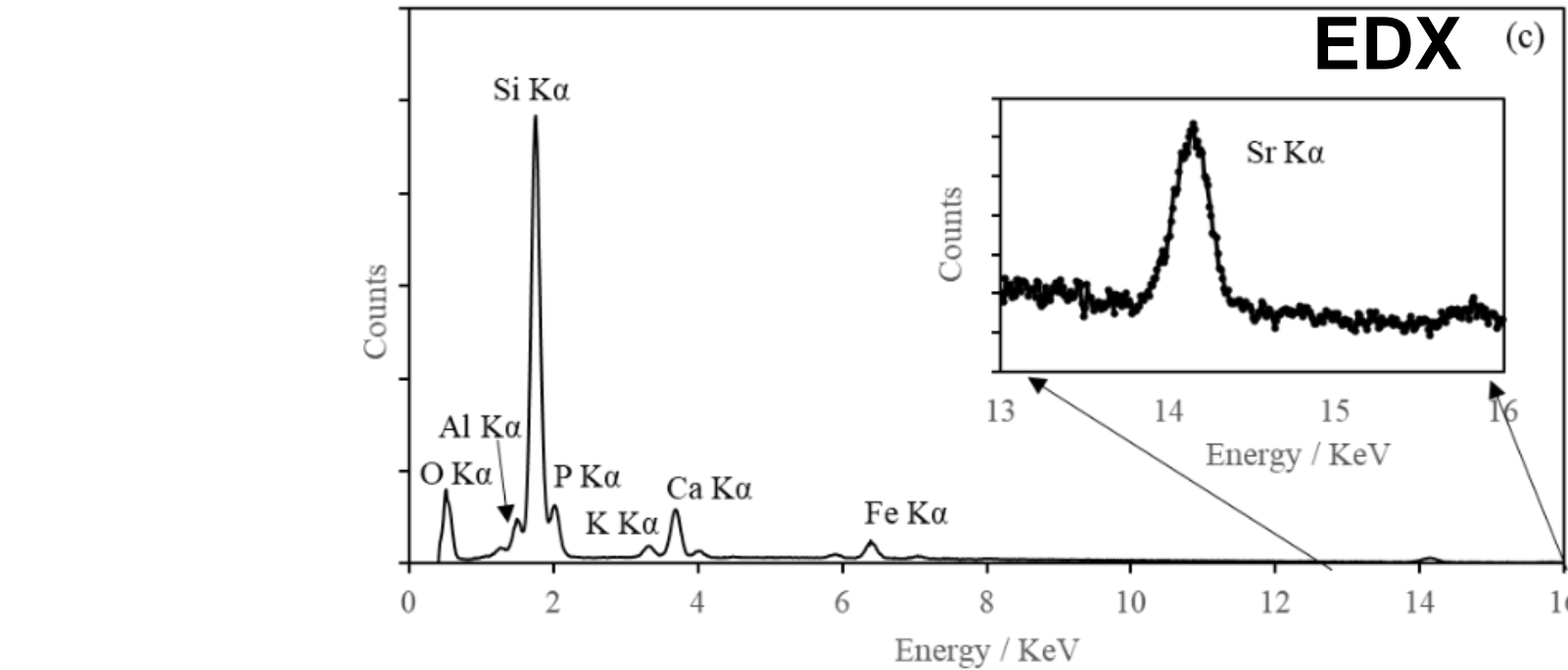
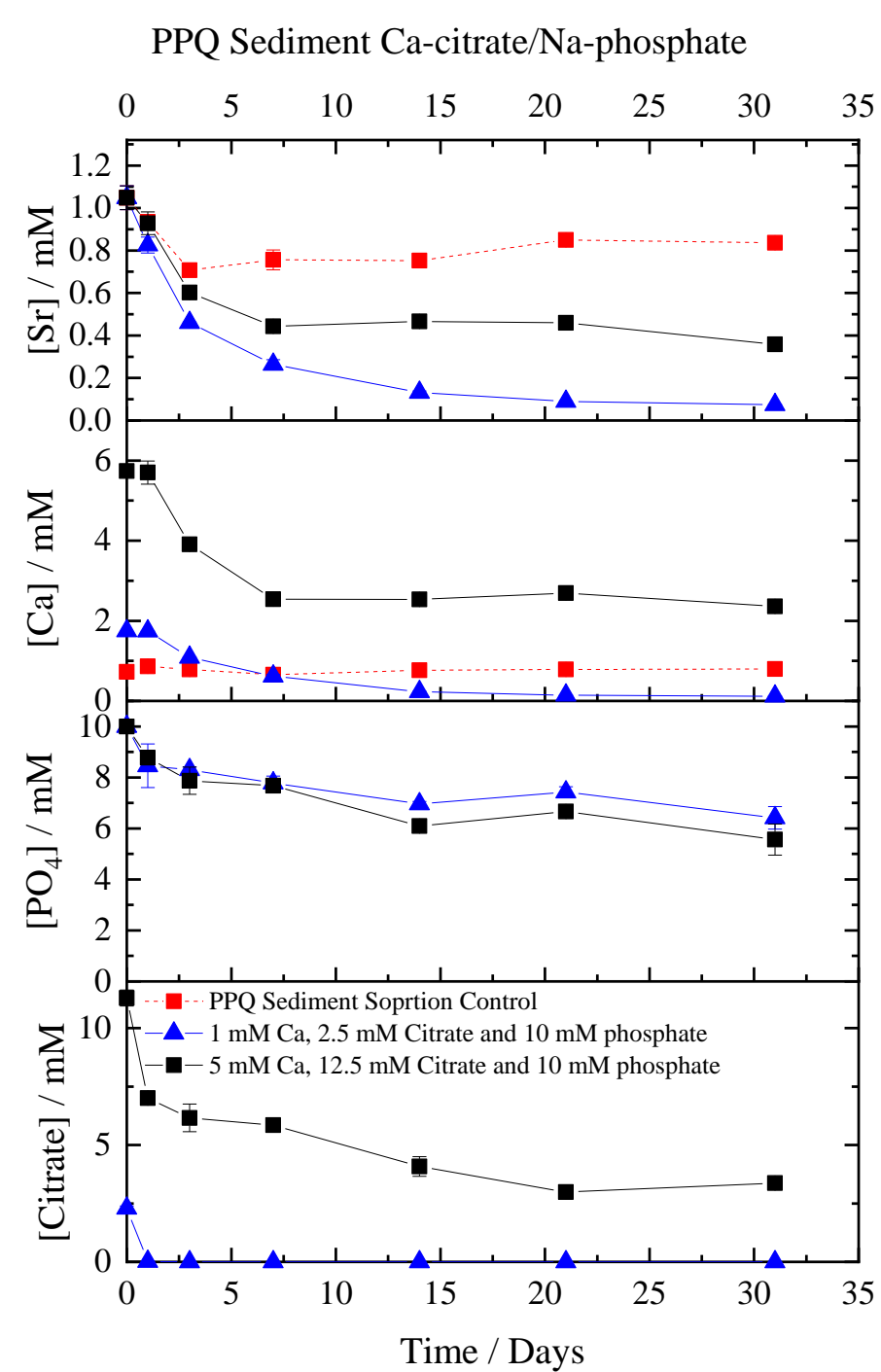
Supporting Information

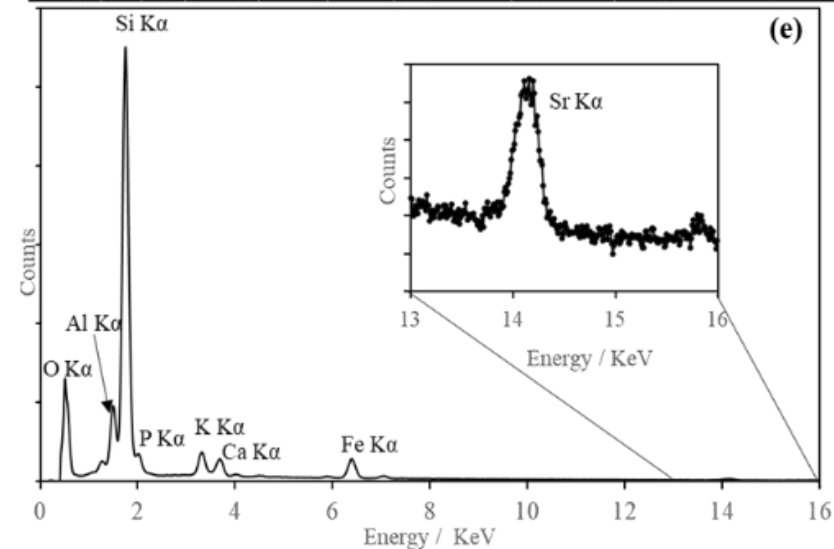
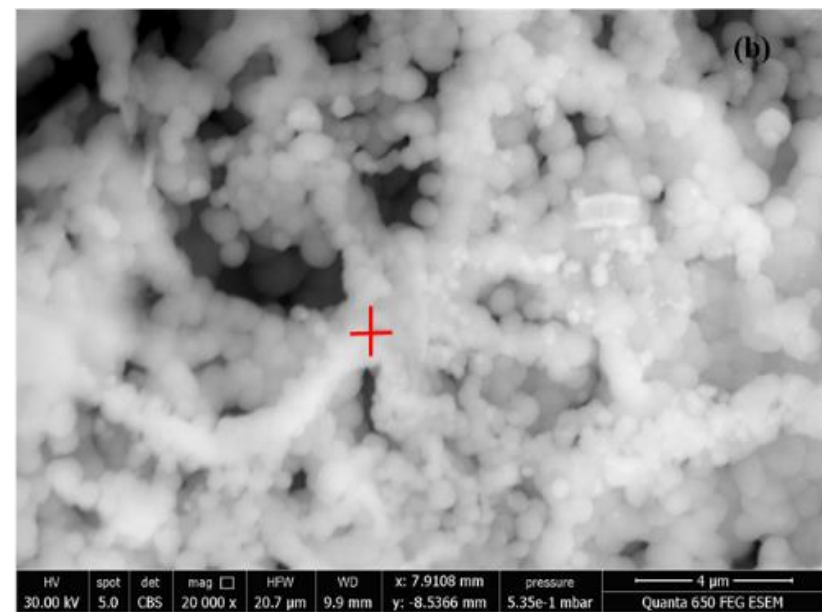
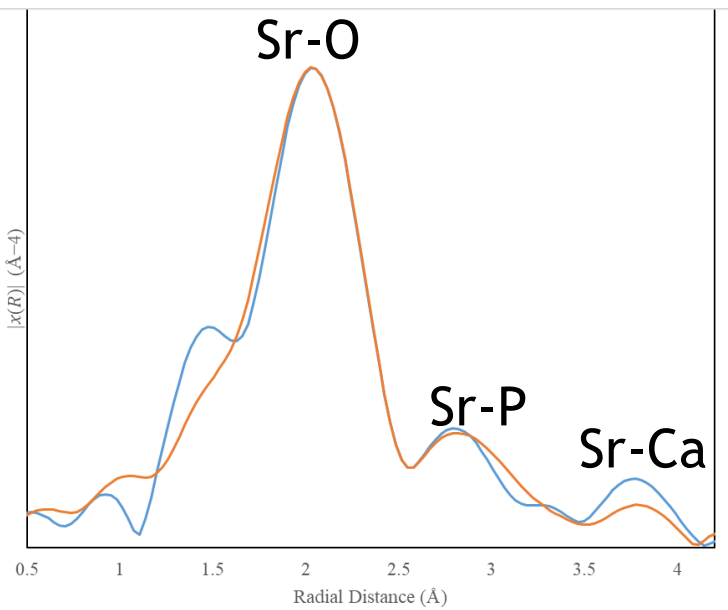
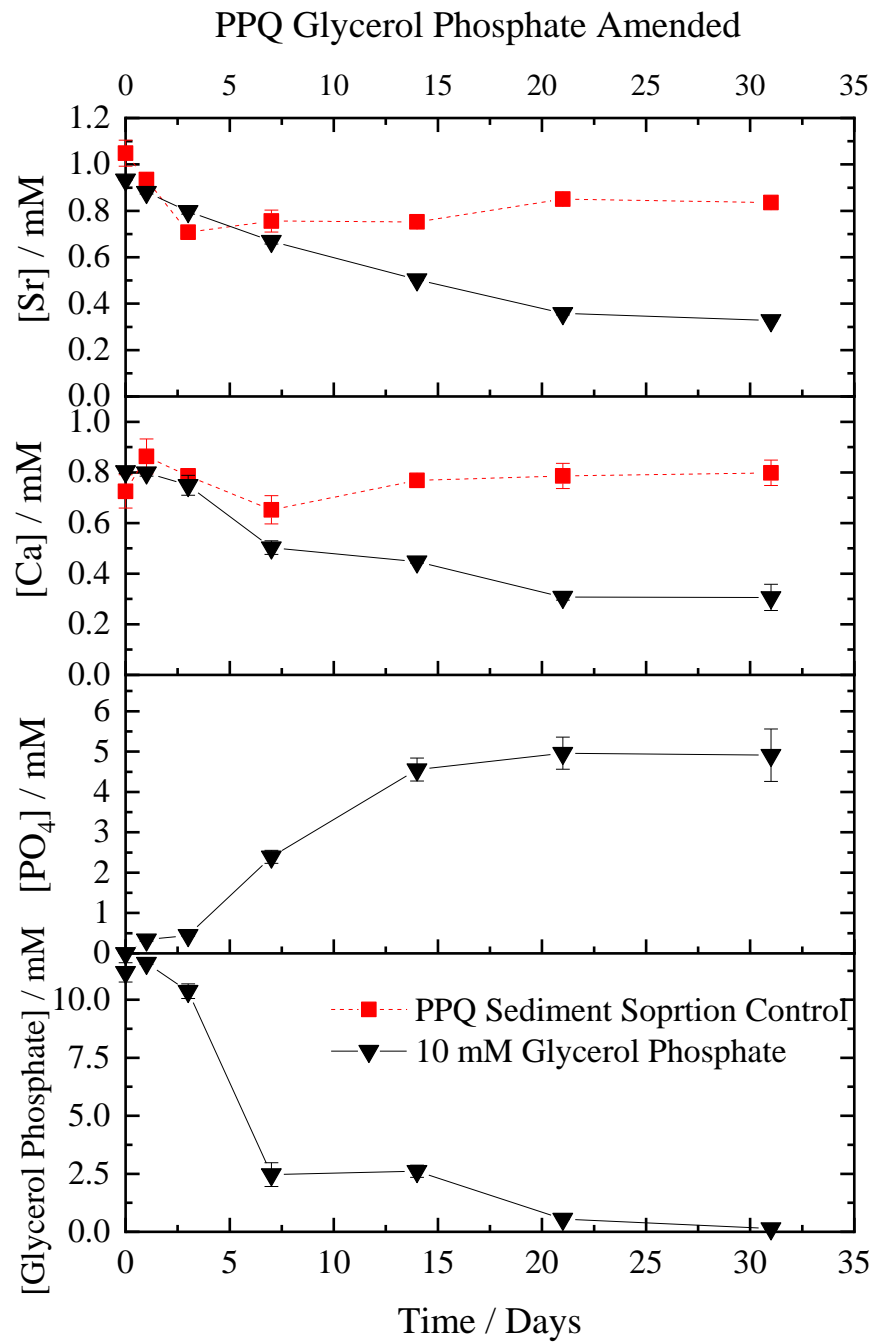
**ABSTRACT:** Historical operations at nuclear mega-facilities such as Hanford, USA, and Sellafield, UK have led to a legacy of radioactivity-contaminated land. Calcium phosphate phases (e.g., hydroxyapatite) can adsorb and/or incorporate radionuclides, including  $^{90}\text{Sr}$ . Past work has shown that aqueous injection of Ca-phosphate-generating solutions into the contaminated ground on both laboratory and field scales can reduce

Sr in the systems. Here, two microbially mediated phosphate which precipitated Ca-phosphate, (i) Ca-citrate/Na-phosphate, were tested in batch experiments alongside an abiotic phosphate), using stable Sr and site relevant groundwaters and endments led to enhanced Sr removal from the solution ent-only control. The Ca-citrate/Na-phosphate treatment phosphate 60%, and polyphosphate 55% of the initial Sr. At scanning electron microscopy showed that Sr-containing, Ca-posit on sediment grains, and XAS analyses of the sediments amended with Ca-citrate/Na-phosphate nfirm ed Sr incorporation into Ca-phosphates occurred. Overall, Ca-phosphate-generating treatments have d in a range of nuclear sites and are a key option within the toolkit for  $^{90}\text{Sr}$  groundwater remediation. *ation, in situ, mineralization, groundwater, radiostrontium*



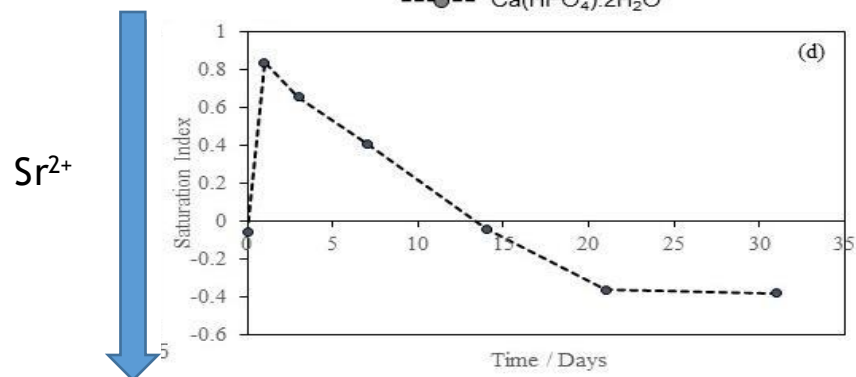




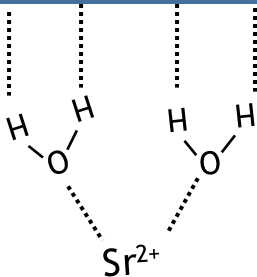


# Sr Speciation during Mineralisation

Released into solution



Amorphous  $\text{Ca-PO}_4$  Phases

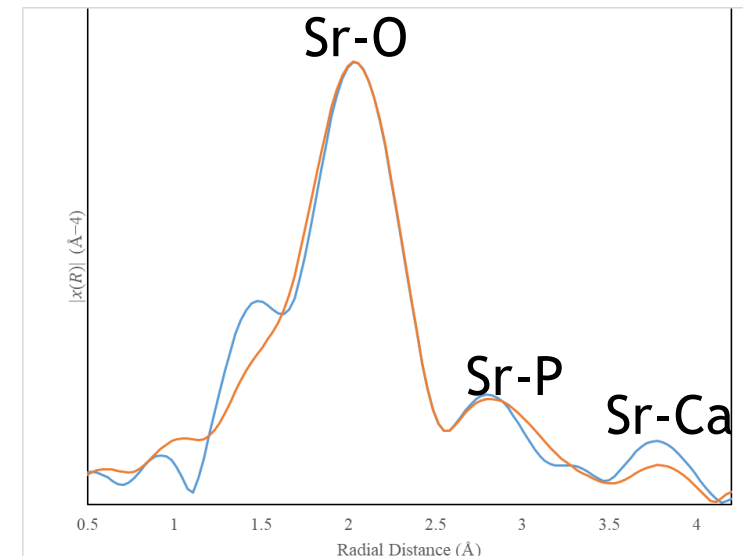


Rapid sorption to  
 $\text{Ca-PO}_4$  phases

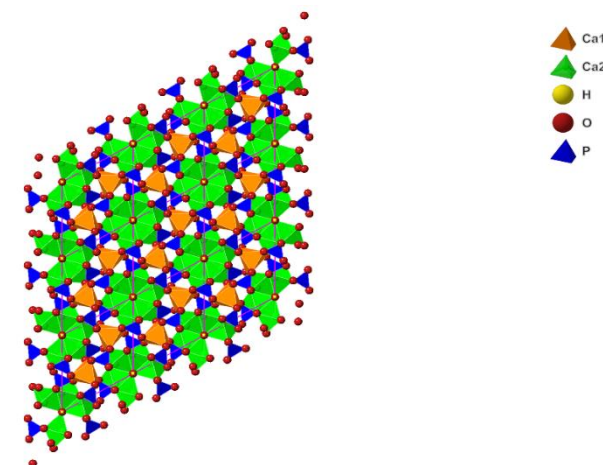
Recrystallisation

Incorporation into  
poorly ordered  
hydroxyapatite-like  
phase

Incorporation of  $\text{Sr}^{2+}$

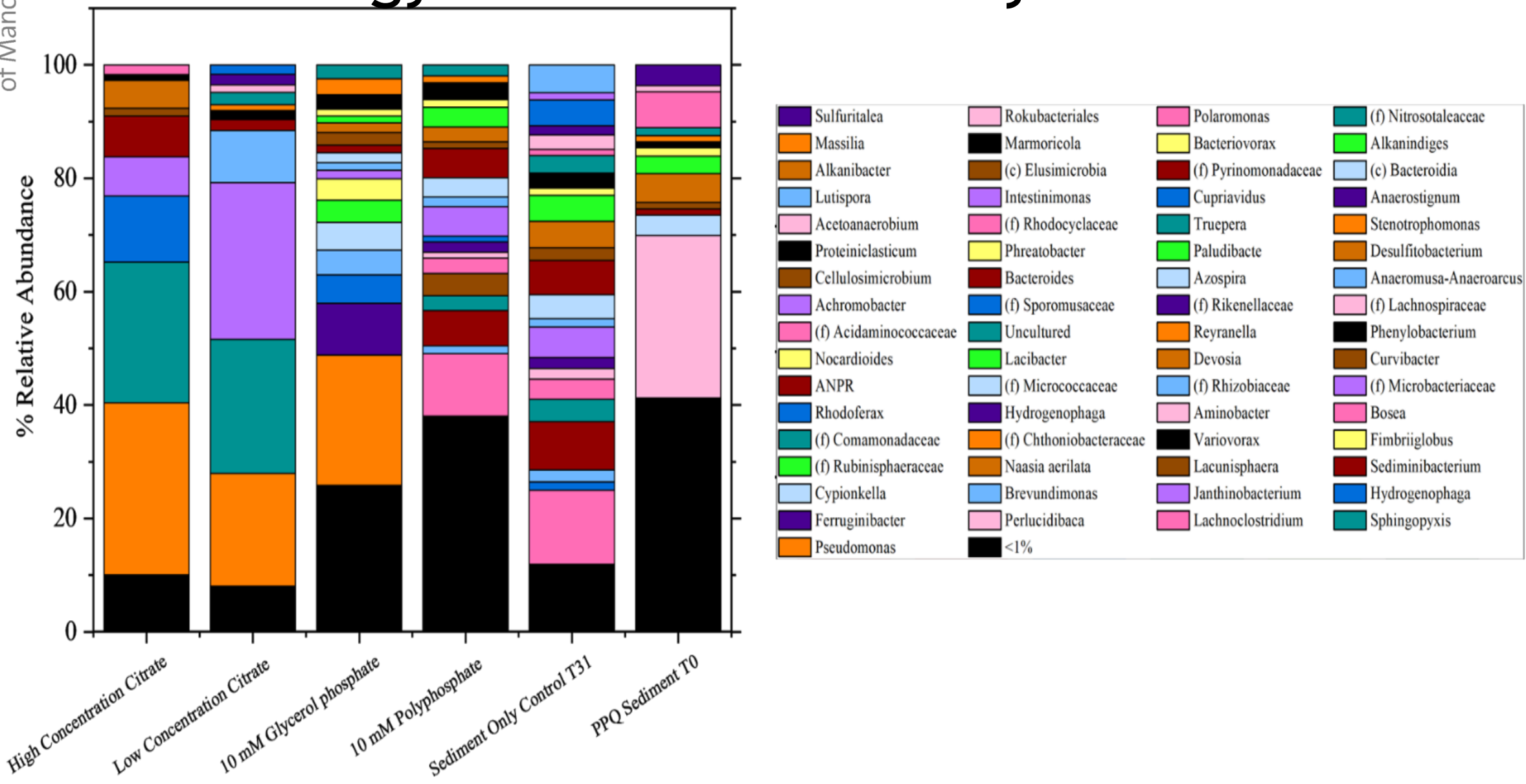


Crystalline  $\text{Ca-PO}_4$  Phases e.g. Hydroxyapatite





# Microbiology of Sr biomineral systems



# U Bioremediation at Sellafield

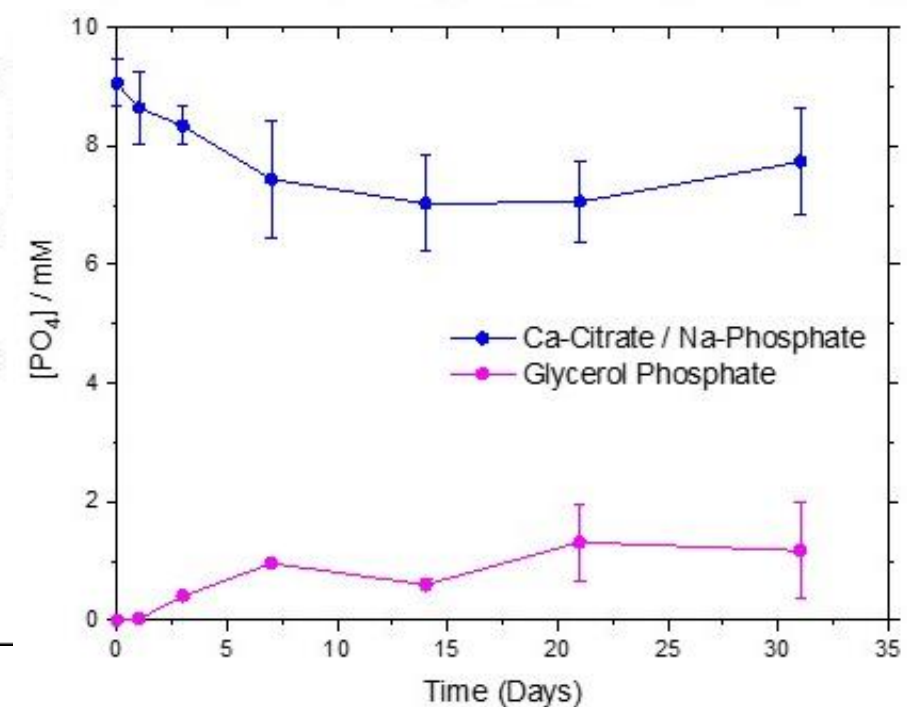
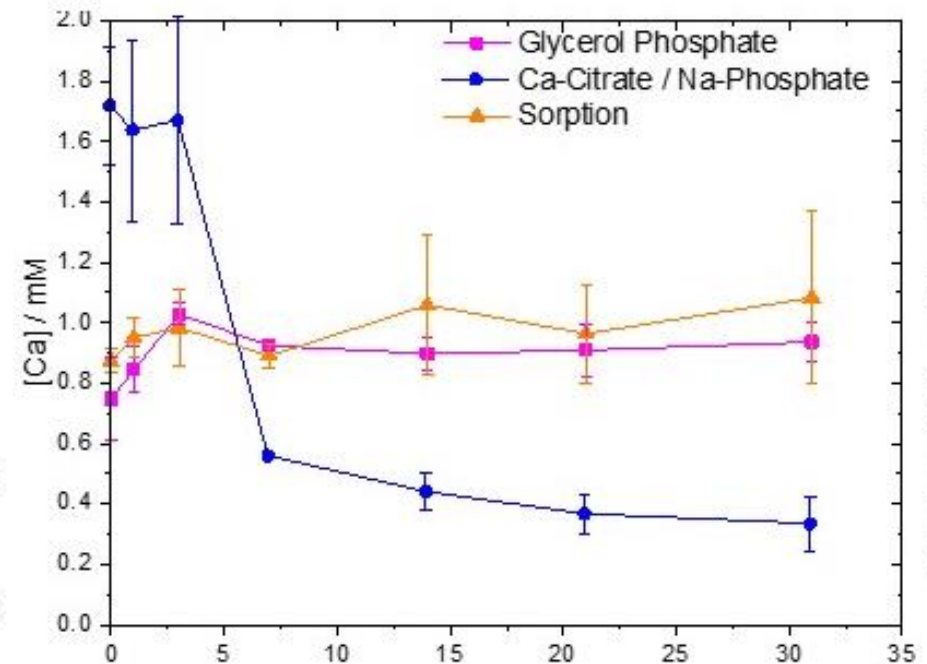
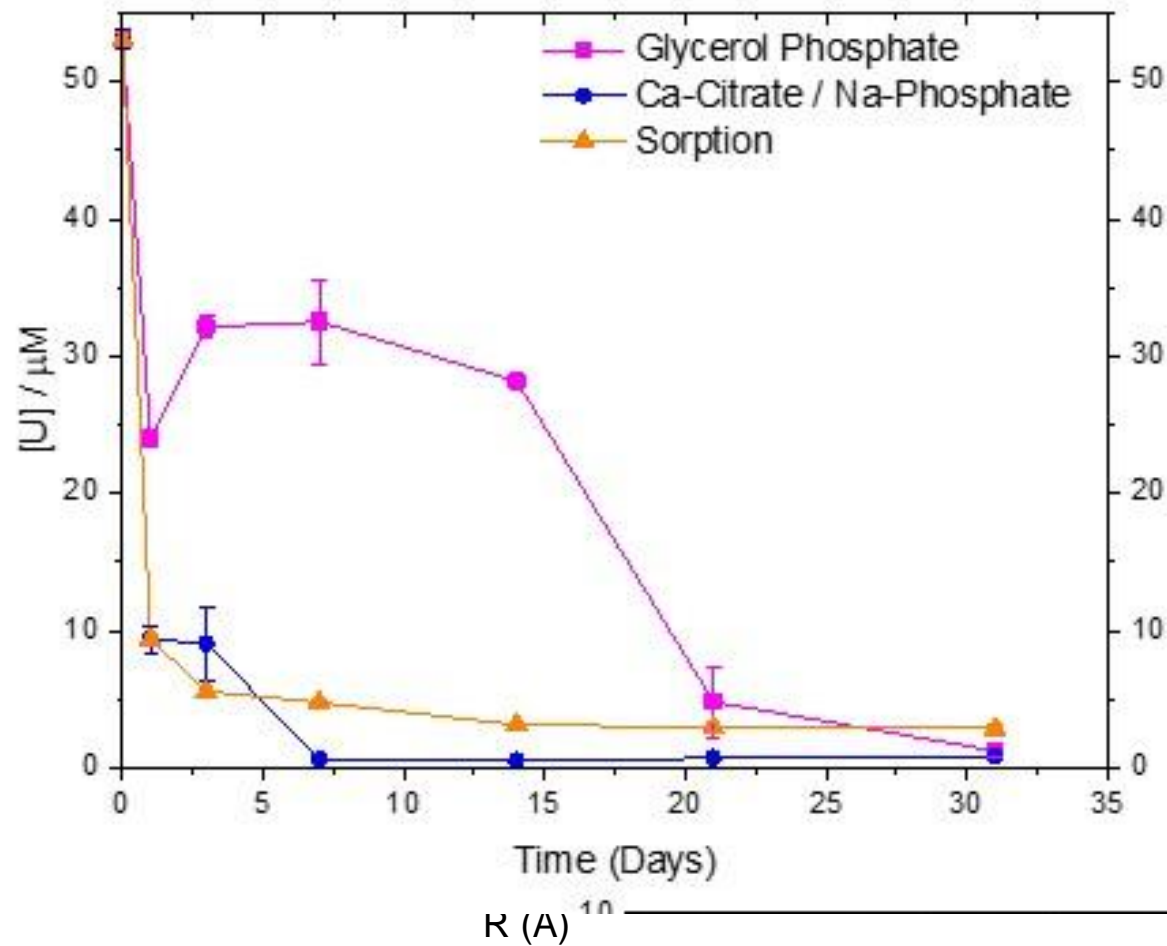
- U mobility in the environment is controlled by pH and redox
- Past work on  $U^{(VI)}$  bioreduction using Fe(III)-reducing bacteria [1],[2]
- Sellafield sub-surface has oxic regions and anoxic regions [2]
- Oxic conditions present challenges to redox-driven  $U^{(VI)}$  remediation
- Application of in situ phosphate technologies to  $U^{(VI)}$  under Sellafield relevant conditions



[1] L. Newsome, K. Morris, D. Trivedi, A. Bewsher, and J. R. Lloyd, "Biostimulation by Glycerol Phosphate to Precipitate Recalcitrant Uranium(IV) Phosphate," *Environ. Sci. Technol.*, 49(18), 11070–11078 (2015).

[2] L. Newsome, K. Morris, D. Trivedi, N. Atherton, and J. R. Lloyd, "Microbial reduction of uranium(VI) in sediments of different lithologies collected from Sellafield," *Appl. Geochemistry*, 51, 55–64 (2014).

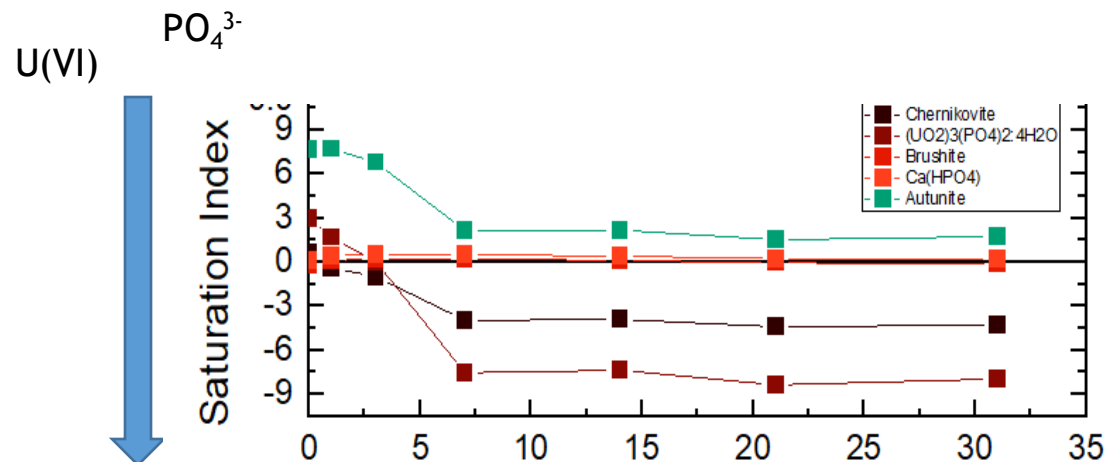
# U(VI) Biomineralisation





# U(VI) Phosphate Mineralisation

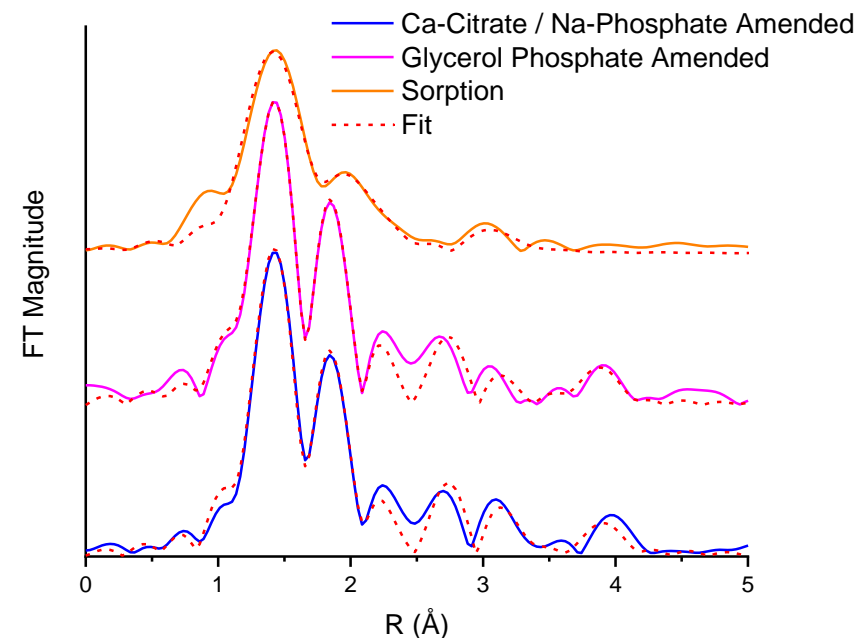
Released into solution



Ca-autunite, Chernikovite and Uranyl orthophosphate are predicted to be oversaturated

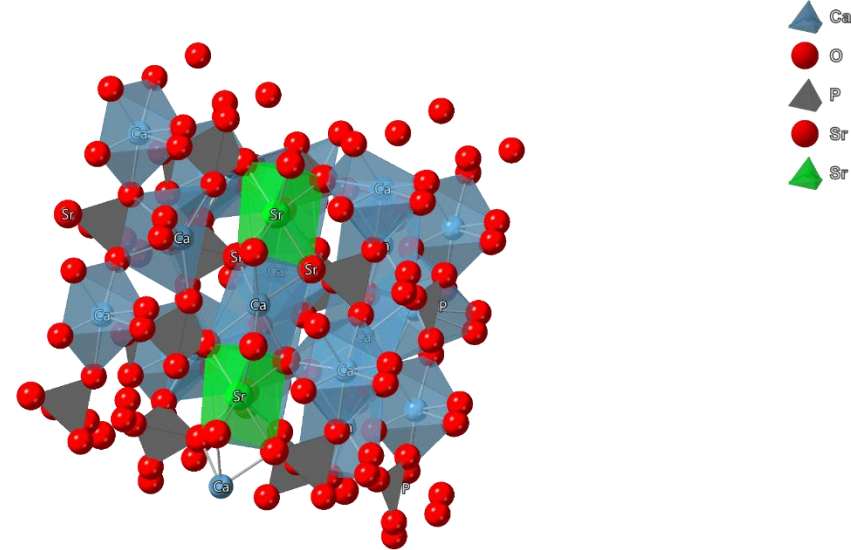


EXAFS analysis suggested uranyl orthophosphate is the predominate phase

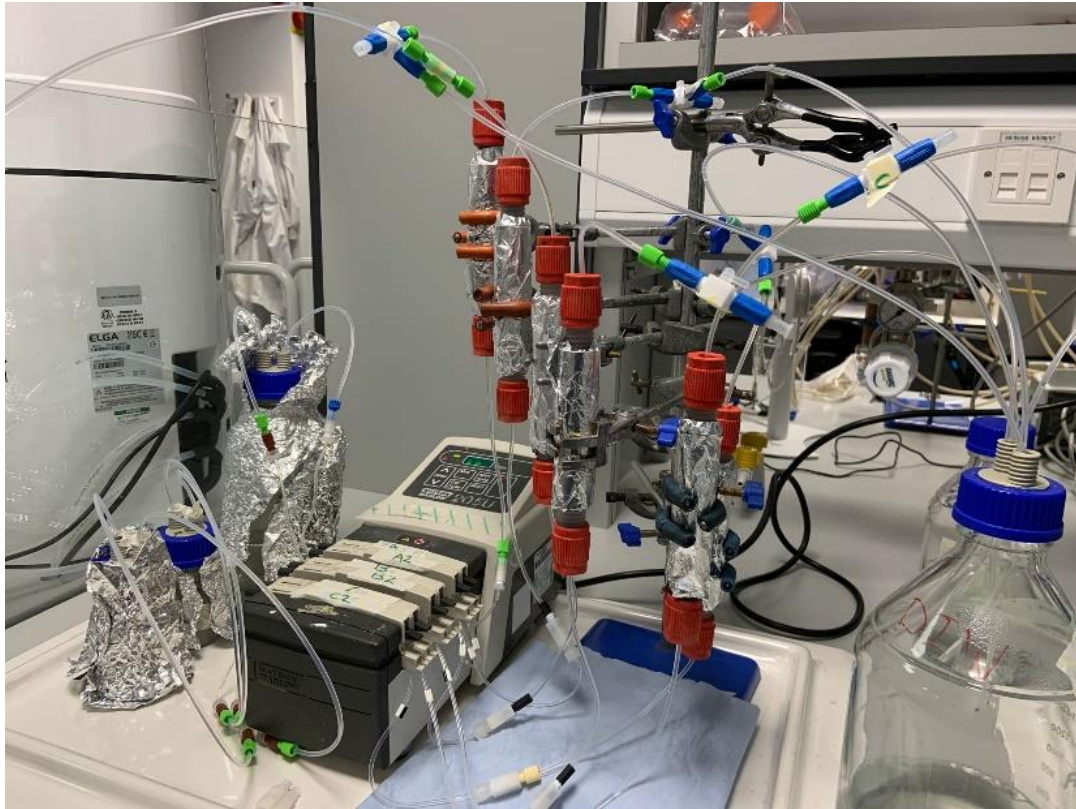


# Summary of Batch Oxidic Microcosms

- Ca-Citrate/Na-phosphate highly successful for both  $U^{VI}$  and  $Sr^{2+}$
- Glycerol phosphate successful for both  $U^{VI}$  and  $Sr^{2+}$
- Sr/U bearing phosphate minerals formed in both treatments



# Flowing Column Experiments

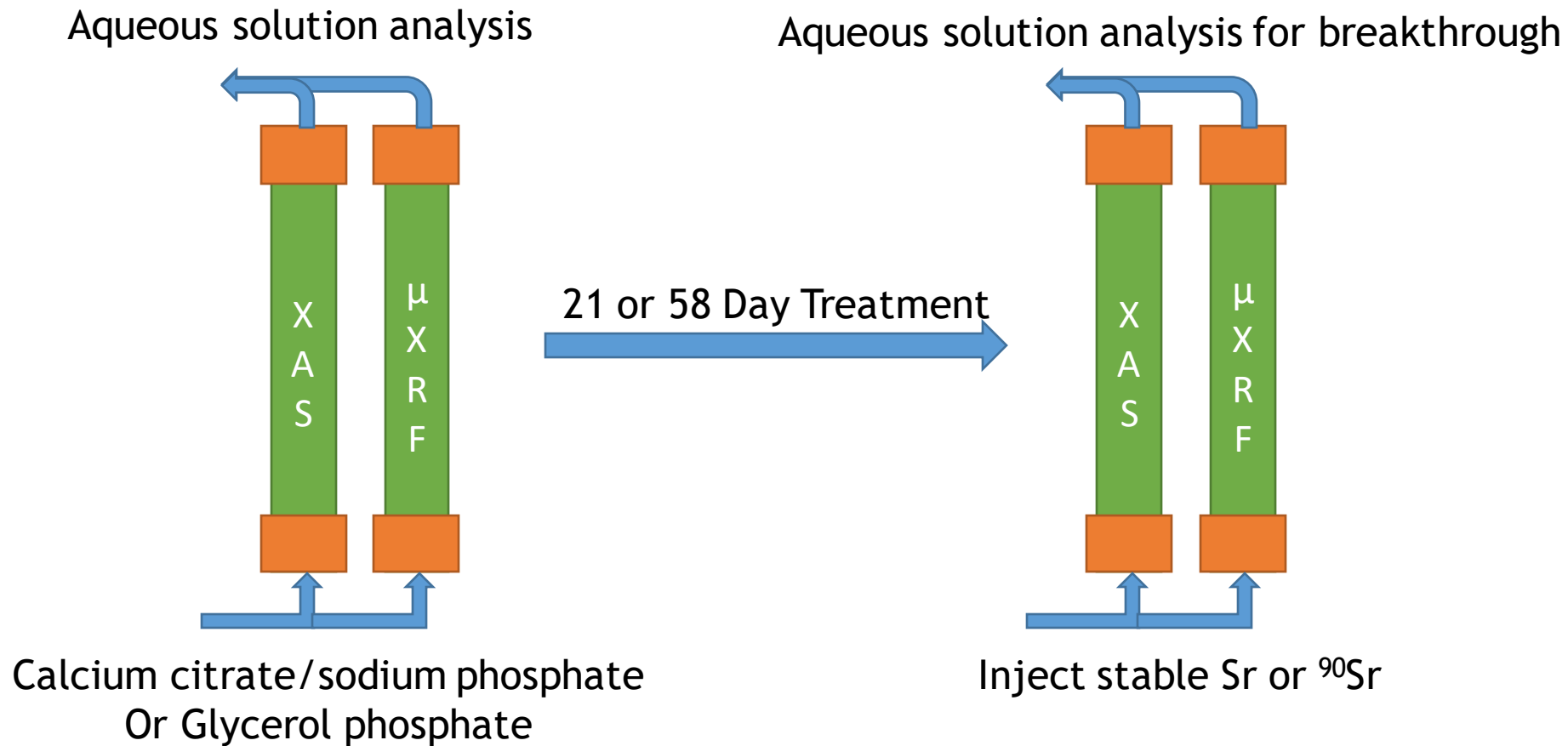


- Columns enabling  $^{90}\text{Sr}$  experiments under flowing conditions
- Flow regime representative of Sellafield subsurface
- Treated columns for 21 days with Ca-phosphate amendments:
  - Calcium citrate/sodium phosphate
  - Glycerol phosphate

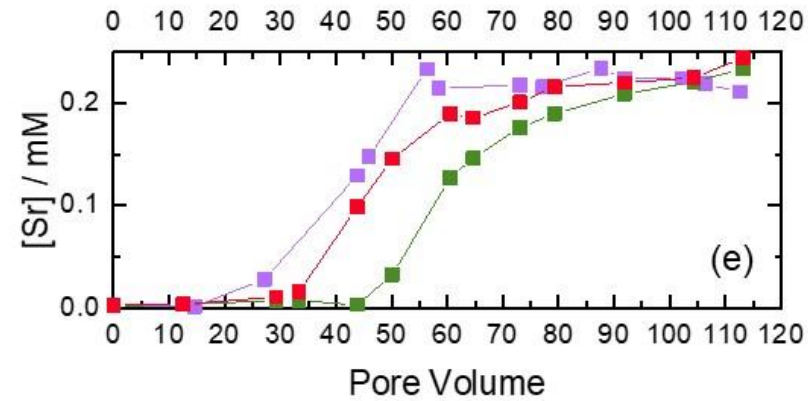
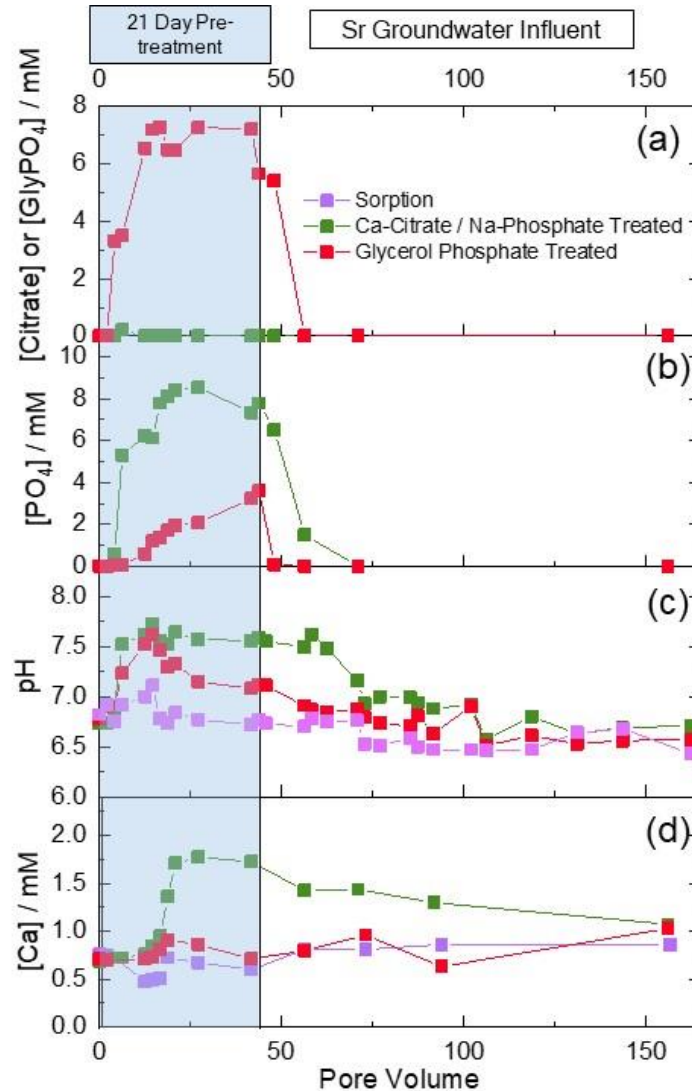
(With a lot of help from Gareth Law and Mallory Ho (Uni of Helsinki) M. S. Ho, G. F. Vettese, K. Morris, J. R. Lloyd, C. Boothman, W. R. Bower, S. Shaw, and G. T. W. Law, "Retention of immobile Se(0) in flow-through aquifer column systems during bioreduction and oxic-remobilization," *Sci. Total Environ.*, 834(February), 155332 (2022). [M. S. Ho, G. F. Vettese, P. H. Keto, S. P. Lamminmäki, M. Vikman, E. Myllykylä, K. Dardenne, and G. T. W. Law, "Mechanisms Governing  $^{90}\text{Sr}$  Removal and Remobilisation in a VLLW Surface Disposal Concept," *Minerals*, 13(3), 436 (2023).



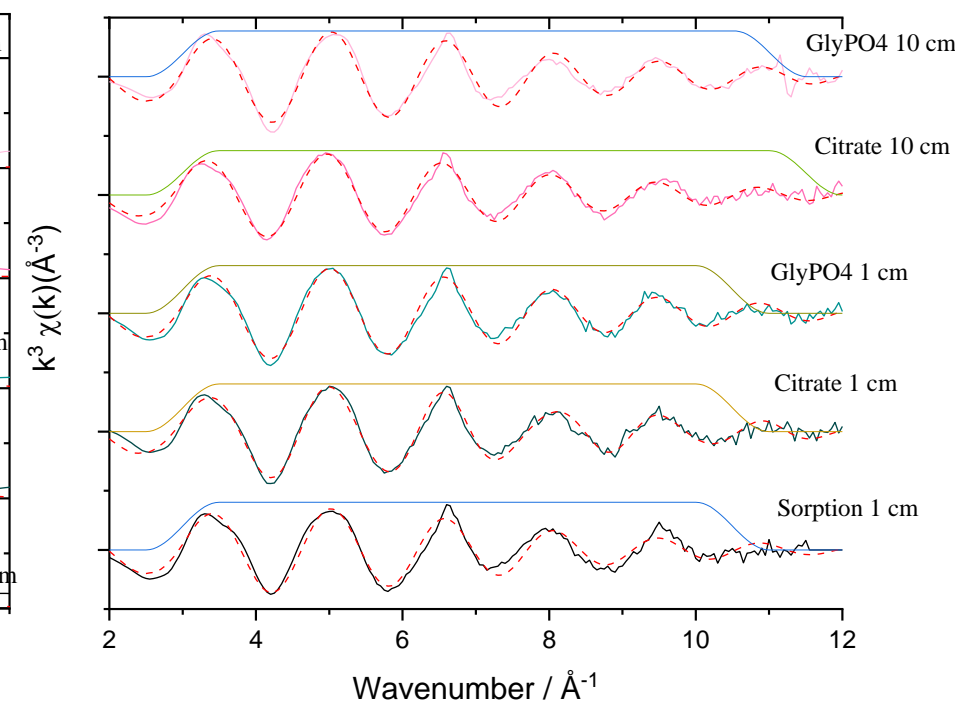
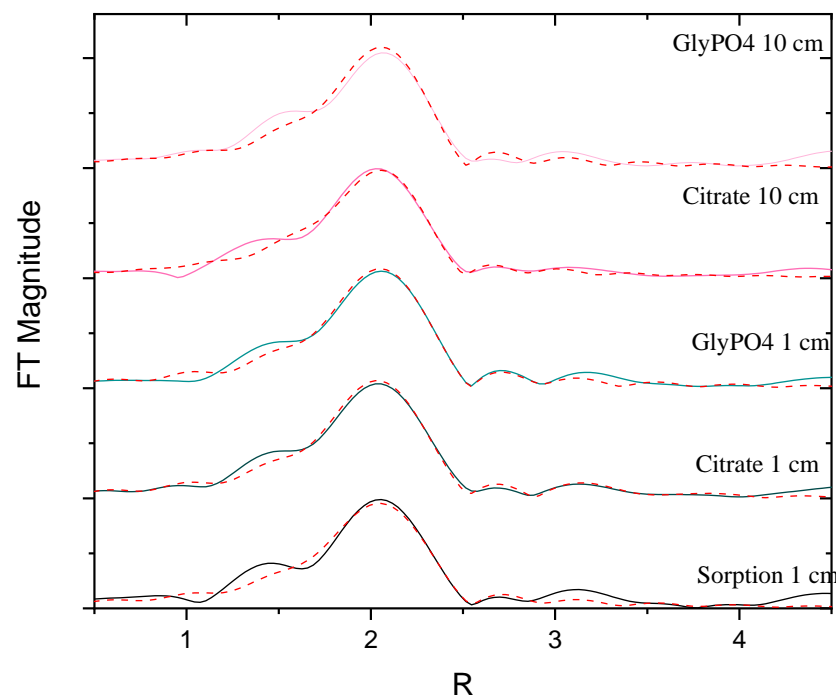
# Flowing Column Experiments



# Column Results Stable Sr at 20 ppm

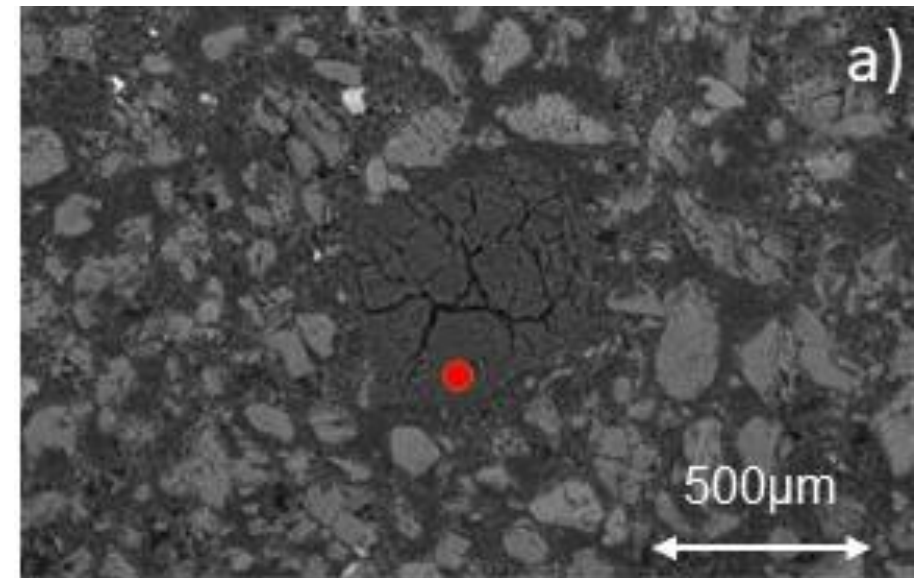
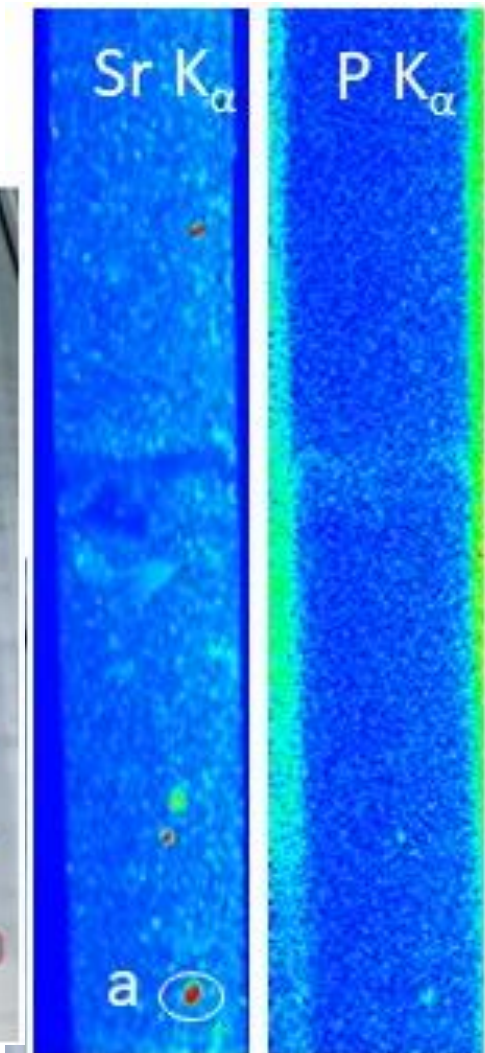
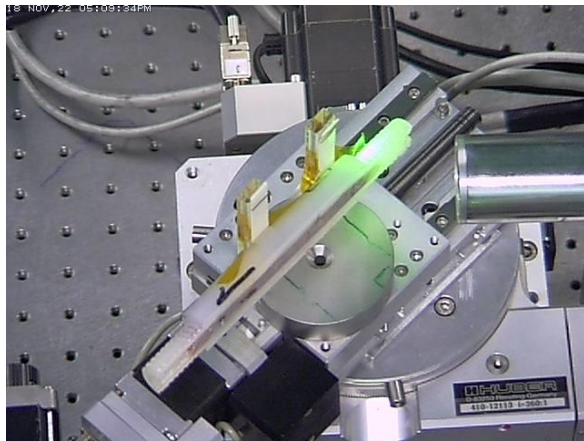


# Solid Phase Analysis of Columns



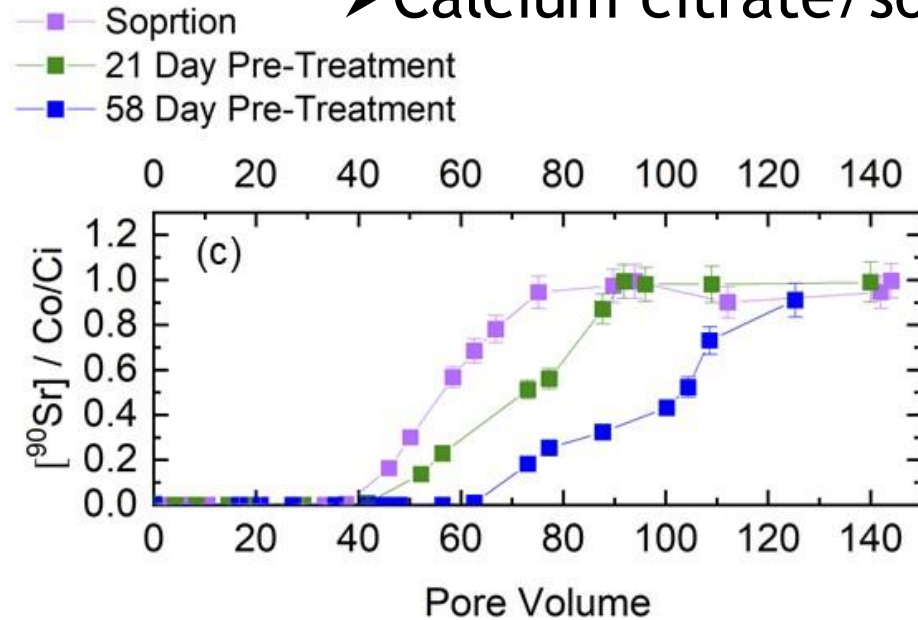
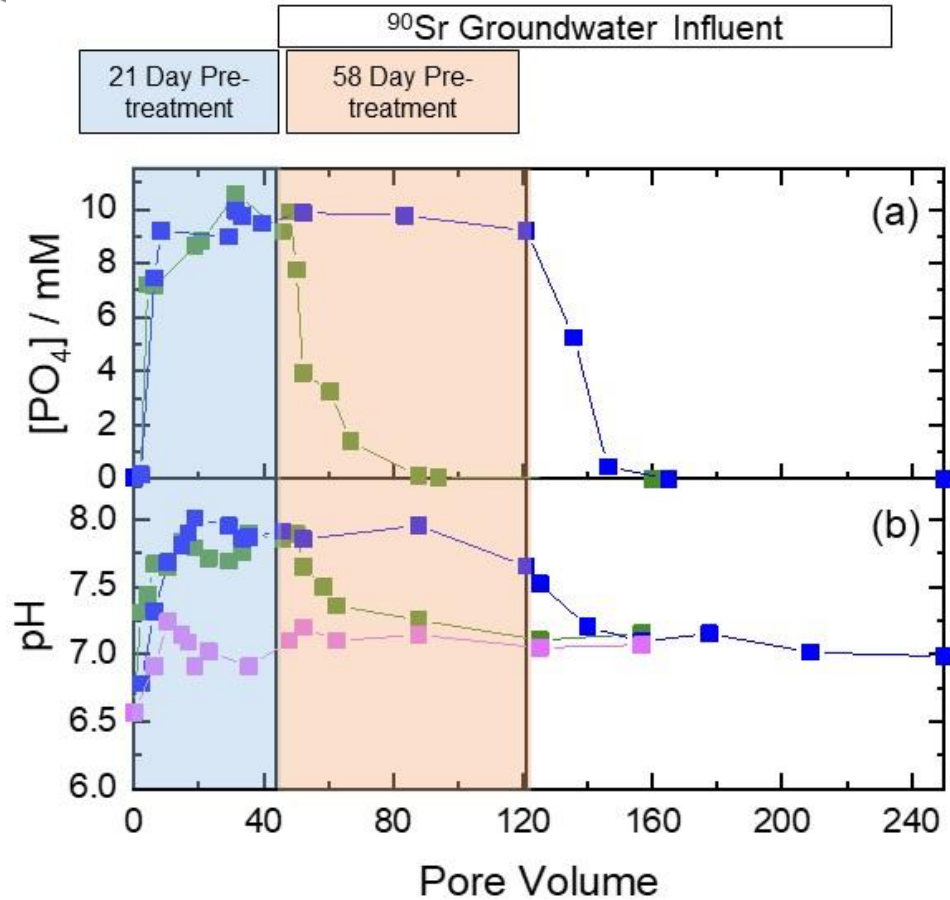


# Micro Focus XRF



# $^{90}\text{Sr}$ Flowing Column Experiments

- Pre-treated columns for 21 and 58 days with Ca-phosphate amendment:
- Calcium citrate/sodium phosphate



# Summary of Phosphate Mineralisation Study

- Batch experiments - removal of  $\text{Sr}^{2+}$  using phosphate biomineralisation (oxic conditions)
- $\text{U}^{\text{VI}}$  phosphate biomineralisation under Sellafield representative conditions
- Dynamic flowing column study -  $^{90}\text{Sr}$  sequestration via in situ phosphate biomineralisation





# Thank You for Listening

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Manchester Geomicrobiology Group



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