

PNNL-21490 WTP-RPT-220, Rev 0

Prepared for the U.S. Department of Energy under Contract DE-AC05-76RL01830

Fire Safety Tests for Cesium-Loaded Spherical Resorcinol Formaldehyde Resin: Data Summary Report

D Kim MJ Schweiger RA Peterson

September 2012



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PACIFIC NORTHWEST NATIONAL LABORATORY operated by BATTELLE for the UNITED STATES DEPARTMENT OF ENERGY under Contract DE-AC05-76RL01830

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Test Specification: 24590 PTF-TSP-RT-09-002, Rev. 0 Test Plan: TP-WTPSP-002, Rev. 3.0 Test Exceptions: 24590-PTF-TEF-RT-11-00004, Rev. 0 R&T Focus Area: Pretreatment Test Scoping Statement: None

Prepared for the U.S. Department of Energy Under Contract DE-AC05-76RL01830

Pacific Northwest National Laboratory Richland, Washington 99352

Completeness of Testing

This report describes the results of work and testing specified by Test Specification 24590-PTF-TSP-RT-09-002, Rev 0, and Test Plan TP-WTPSP-002, Rev. 3.0. The work followed the quality assurance requirements outlined in the test specification and test plan. The descriptions provided in this report are an accurate account of both the conduct of the work and the data collected. Test plan results are reported. Also reported are any unusual or anomalous occurrences that are different from expected results. The test results and this report have been reviewed and verified.

Approved:

DE Kurath

Dean E. Kurath, Manager WTP R&T Support Project

9/18/12 Date

Testing Summary

A preliminary safety evaluation of the scenario for spherical resorcinol formaldehyde (SRF) resin fire inside the ion exchange column was performed by the Hanford Tank Waste Treatment and Immobilization Plant (WTP) Fire Safety organization. The result of this evaluation suggested a potential change of the fire safety classification for the Cesium Ion Exchange Process System (CXP) emergency elution vessels, equipment, and piping. To resolve this question, the fire properties of the SRF resin were measured by Southwest Research Institute (SwRI) through a subcontract managed by Pacific Northwest National Laboratory (PNNL).

The results of initial fire safety tests on the SRF resin were documented in a previous report (PNNL-21321). The present report summarizes the results of additional tests performed by SwRI on the cesium-loaded SRF resin. The efforts by PNNL were limited to summarizing the test results provided by SwRI into one consolidated data report. The as-received SwRI report is attached to this report in the Appendix A. Where applicable, the precision and bias of each test method, as given by each American Society for Testing and Materials (ASTM) standard procedure, are included and compared with the SwRI test results of the cesium-loaded SRF resin.

Objectives

The objectives for the fire safety tests were:

- to measure the thermal conductivity and heat capacity for sodium-form SRF resin,
- to perform thermo-gravimetric analysis (TGA) with mass spectrometry of the nonradioactive cesium spiked resin offgas up to the auto ignition temperature and determine constituents of the offgas, and
- to repeat ignition test using resin loaded with nonradioactive cesium.

Table S.1 provides the objectives that applied to the SRF resin fire safety tests.

Test Exceptions

The test exception in Test Plan TP-WTPSP-002 Rev 3 that was applicable to this testing is presented in Table S.2.

Results and Performance Against Success Criteria

The success criteria for achieving the test objectives are discussed in Table S.3.

Test Objective	Objective Met?	Discussion
• Measure the thermal conductivity and heat capacity for sodium-form SRF resin.	Yes	The thermal conductivity of the cesium-loaded SRF resin was measured in accordance with ASTM D5930-09 and ASTM D5334-08. The thermal conductivity (λ in W/[mK]) as a function of temperature (T in °C) between 22 and 90°C was expressed by $\lambda = 0.000273T + 0.0769$. The specific heat capacity of the cesium-loaded SRF resin was measured by differential scanning calorimetry (DSC) in accordance with ASTM E1269-11. The average specific heat capacity ranged from 1.434 J/(gK) at 40°C to 2.001 J/(gK) at 90°C.
• Perform TGA with mass spectrometry of the nonradioactive cesium spiked resin offgas up to the auto ignition temperature and determine constituents.	Partially Yes	From the TGA combined with DSC analysis, an endothermic transition near 102°C appeared with a total weight loss of approximately 25%. Small exothermic transitions occurred around 387°C and 619°C with a large exothermic transition near 864°C with corresponding weight losses of approximately 30%, 5%, and 20%, respectively. Because SwRI did not have a TGA setup with mass spectrometer, the sample was run with TGA setup integrated with Fourier transform infrared (FTIR) for offgas analyses. The result of FTIR analyses showed that the weight loss peak at 102°C was linked to H ₂ O while all other peaks at higher temperature were linked to CO ₂ gas.
 Repeat ignition test using resin loaded with nonradioactive cesium. 	Yes	The flash ignition temperature (FIT) and spontaneous ignition temperature (SIT) of the cesium-loaded SRF resin were determined in accordance with ASTM D1929-96. The average values of FIT and SIT from duplicate tests were 555°C and 610°C, respectively. The FIT is 40°C lower than the previous result for the SRF resin without cesium and SIT is only 3°C lower, which is well below the repeatability limits of 11 to 31°C.

Table S.1.	Summary	of Test Ob	jectives and	Results

Test Exception Number	Description of Test Exception
24590-PTF-TEF-RT-11-00004, Rev 0	This test exception was issued by Bechtel National, Inc. (BNI) on December 21, 2011. This test exception incorporates the approved Request for Technology Development (RTD), 24590-WTP-RTD-RT-11-0008_Rev_000, which requests data on thermal conductivity, heat capacity, and thermal analysis of the SRF resin and repeat of the ignition tests using the cesium-loaded SRF resin.

	List Success Criteria	Explain How the Tests Did or Did Not Meet the Success Criteria
1.	Provide test results for thermal conductivity and heat capacity measurements of sodium-form SRF resin.	This success criterion was met. The thermal conductivity (λ in W/[mK]) of the cesium-loaded SRF resin as a function of temperature (T in °C) between 22 and 90°C was expressed by $\lambda = 0.000273T + 0.0769$. The average thermal conductivity was 0.083 W/(mK) at 25°C, 0.088 W/(mK) at 45°C, and 0.102 W/(mK) at 90°C. The average specific heat capacity of the cesium-loaded SRF resin ranged from 1.434 J/(gK) at 40°C to 2.001 J/(gK) at 90°C.
2.	Provide test results for TGA with mass spectrometry of the nonradioactive cesium spiked resin offgas up to the auto ignition temperature and the determined constituents.	This success criterion was partially met. From the TGA-DSC analysis, an endothermic transition near 102°C appeared with a total weight loss of approximately 25%. Small exothermic transitions occurred around 387°C and 619°C and a large exothermic transition near 864°C with corresponding weight losses of approximately 30%, 5%, and 20%, respectively. Because SwRI did not have a TGA setup with mass spectrometer, the sample was run with TGA-FTIR setup for offgas analyses. The result of FTIR analyses showed that the weight loss peak at 102°C was linked to H ₂ O while all other peaks at higher temperature were linked to CO ₂ gas.
3.	Provide test results for the spontaneous ignition temperature using resin loaded with nonradioactive cesium.	This success criterion was met. For the cesium-loaded SRF resin, the FIT was 555°C, and SIT was 610°C.

Quality Requirements

The PNNL Quality Assurance (QA) Program is based on the requirements defined in U.S. Department of Energy Order 414.1D, *Quality Assurance*, and 10 CFR 830, *Energy/Nuclear Safety Management*, Subpart A – Quality Assurance Requirements (a.k.a. the Quality Rule). PNNL has chosen to implement the following consensus standards in a graded approach:

- ASME NQA-1-2000, *Quality Assurance Requirements for Nuclear Facility Applications*, Part I, Requirements for Quality Assurance Programs for Nuclear Facilities
- ASME NQA-1-2000, Part II, Subpart 2.7, Quality Assurance Requirements for Computer Software for Nuclear Facility Applications
- ASME NQA-1-2000, Part IV, Subpart 4.2, Graded Approach Application of Quality Assurance Requirements for Research and Development.

The Waste Treatment Plant Support Project (WTPSP) implements an NQA-1-2000 Quality Assurance Program, graded on the approach presented in NQA-1-2000, Part IV, Subpart 4.2. The WTPSP Quality Assurance Manual (QA-WTPSP-0002) describes the technology life-cycle stages under the WTPSP Quality Assurance Plan (QA-WTPSP-0001). The technology life cycle includes the progression of technology development, commercialization, and retirement in process phases of basic and applied research and development (R&D), engineering and production, and operation until process completion. The life cycle is characterized by flexible and informal quality assurance activities in basic research that become more structured and formalized through the applied R&D stages. The work described in this report has been completed under the QA technology level of Applied Research. The WTPSP addresses internal verification and validation activities by conducting an independent technical review of the final data report in accordance with WTPSP procedure QA-WTPSP-601, *Document Preparation and Change*. This review verifies that the reported results are traceable, that inferences and conclusions are soundly based, and that the reported work satisfies the test plan objectives.

The Southwest Research Institute, who performed the tests described in this report, is listed on the PNNL Approved Suppliers List.

Test Conditions

This report summarizes the fire test results performed by SwRI and submitted to PNNL.

All test conditions delineated by the test plan and test exceptions were met. A summary of test conditions is provided in Table S.4.

Simulant Use

Simulant was not developed for the tests summarized in this report. The SRF resin was manufactured by Microbeads AS, a Norwegian company (<u>www.micro-beads.com</u>), and shipped in an approximately100-L steel drum. WTP provided the entire drum to PNNL. About 20 L of the resin were shipped to SwRI in a 20-L carboy. The SRF resin was loaded with cesium by SwRI according to the procedure supplied by PNNL.

Test Conditions	Were Test Conditions Followed?
1. The ignition temperature of the resin loaded with non-radioactive cesium will be determined along with the thermal conductivity and heat capacity of the resin in sodium-form. A TGA with mass spectrometry of the resin offgas up to the auto ignition temperature will also be performed and the constituents determined.	The FIT and SIT were determined in accordance with ASTM D1929-96, the thermal conductivity was measured in accordance with ASTM D5930-09 and ASTM D5334-08, and the specific heat capacity was measured by DSC in accordance with ASTM E1269-11. A TGA-DSC was performed along with FTIR to determine the gaseous constituents.

 Table S.4.
 Test-Condition Summary

Discrepancies and Follow-On Tests

No discrepancies were observed.

Acknowledgments

The authors thank the Southwest Research Institute (Eugene Horton and Radonna Spies, San Antonio, Texas) for test support. The authors also thank Jennifer Meehan and David Sherwood of the Hanford Tank Waste Treatment and Immobilization Plant project for their technical insights and much helpful discussion and support. The authors are grateful for the assistance of Renee Russell for her technical review. In addition, the authors thank Casey D. Emery, PNNL Contracts Department, for his diligent effort in negotiating this contract.

Acronyms and Abbreviations

ASTM	American Society for Testing and Materials
BNI	Bechtel National, Inc.
СХР	Cesium Ion Exchange Process System
DOE	U.S. Department of Energy
DSC	differential scanning calorimetry
FIT	flash ignition temperature
FTIR	Fourier transform infrared
GGRF	ground gel resorcinol formaldehyde
NIST	National Institute of Standards and Technology
NSSP	non-steady-state probe
ORP	Office of River Protection
PMMA	poly methyl methacrylate
PNNL	Pacific Northwest National Laboratory
PTF	Pretreatment Facility
QA	quality assurance
R&D	research and development
R&T	research and technology
RF	resorcinol formaldehyde
RTD	request for technology development
SIT	spontaneous ignition temperature
SRF	spherical resorcinol formaldehyde
SwRI	Southwest Research Institute
TGA	thermo-gravimetric analysis
WTP	Hanford Tank Waste Treatment and Immobilization Plant
WTPSP	Waste Treatment Plant Support Project

Symbols

С	calibration constant
Ι	current flowing through heater wire
L	length of heated needle
Q	heat input per unit length of heater
r	repeatability limit
R	reproducibility limit
R_W	total resistance of heater wire
S	slop obtained from temperature versus $\ln(t)$ or $\ln[1/(t - t_1)]$
t	time from the beginning of heating
t_1	heating time
Т	temperature
ΔT	temperature rise from time zero
λ	thermal conductivity

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1.0 Introduction

Ion exchange using the spherical resorcinol formaldehyde (SRF) resin has been selected by Bechtel National, Inc. (BNI) and approved by the U.S. Department of Energy (DOE) Office of River Protection (ORP) for use in the Pretreatment Facility (PTF) of the Hanford Tank Waste Treatment and Immobilization Plant (WTP). SRF is an engineered spherical form of the older ground gel resorcinol formaldehyde (GGRF) resin, also termed resorcinol formaldehyde (RF).

A draft safety evaluation of the scenario for resin fire inside the ion exchange column was performed by the WTP Fire Safety organization. The result of this draft evaluation suggested a potential change of the fire safety classification for the Cesium Ion Exchange Process System (CXP) emergency elution vessels, equipment, and piping. To help resolve this question, the fire properties of the SRF resin were measured by Southwest Research Institute (SwRI), following the American Society for Testing and Materials (ASTM) standard procedures, through a subcontract managed by Pacific Northwest National Laboratory (PNNL). The results of initial fire safety tests on the SRF resin were documented in a previous report (PNNL-21321). The present report summarizes the results of additional tests performed by SwRI on the cesium-loaded SRF resin.

Section 2.0 details the basis of the PNNL Quality Assurance (QA) Program as applied to the WTP quality requirements. Section 3.0 describes the test methods and ASTM standard procedures used in this testing. Section 4.0 summarizes the results of the experimental tests performed by SwRI. Section 5.0 provides a summary of all test results. Section 6.0 provides a list of pertinent references. An as-received SwRI report is included in the Appendix A.

2.0 Quality Assurance

The PNNL QA Program is based on the requirements defined in U.S. Department of Energy Order 414.1D, *Quality Assurance*, and 10 CFR 830, *Energy/Nuclear Safety Management*, Subpart A – Quality Assurance Requirements (a.k.a. the Quality Rule). PNNL has chosen to implement the following consensus standards in a graded approach:

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The work described in this report has been completed under the QA technology level of Applied Research. WTPSP addresses internal verification and validation activities by conducting an independent technical review of the final data report in accordance with WTPSP's procedure QA-WTPSP-601, *Document Preparation and Change*. This review verifies that the reported results are traceable, that inferences and conclusions are soundly based, and that the reported work satisfies the test plan objectives.

The Southwest Research Institute, who performed the tests described in this report, is listed on the PNNL Approved Suppliers List.

3.0 Experimental

This section describes the test methods and procedures used by SwRI to perform various tests on the cesium-loaded SRF resin. Table 3.1, presented at the end of this section, summarizes the ASTM standard test procedures used by SwRI for this testing.

3.1 Sample Preparation

PNNL provided SwRI with one 20-L carboy of SRF resin that consisted of a micro bead resin material dispersed in water. The SRF resin was loaded with cesium according to the procedure supplied by PNNL (Appendix B). For all tests described in this report, the sample was filtered and then dried for 2 to 12 hr depending on the amount of material required for each test in an oven at 70°C prior to testing. This temperature was chosen not to over-dry the resin. The resin was constantly monitored to achieve a consistency in drying condition gauged by the color.

3.2 Flash Ignition Temperature and Spontaneous Ignition Temperature Using Hot-Air Furnace

The flash ignition temperature (FIT) and spontaneous ignition temperature (SIT) of the cesium-loaded SRF resin were determined using a hot-air furnace according to ASTM D1929-96, *Standard Test Method for Determining Ignition Temperature*. The hot-air ignition furnace consists primarily of an electrical heating unit and specimen holder. The furnace tube is a vertical tube with an inside diameter of 100 ± 5 mm and a length of 230 ± 20 mm, made of ceramic that will withstand at least 750°C. The inner ceramic tube, with an inside diameter of 75 ± 5 mm, a length of 230 ± 20 mm, and a thickness of approximately 3 mm, is placed inside the furnace tube and positioned 20 ± 2 mm above the furnace floor on spacer blocks. The pilot flame is located immediately above the opening. The test apparatus is shown in Figure 3.1.

The FIT is the minimum temperature at which, under specified test conditions, sufficient flammable gases are emitted to ignite momentarily upon application of a small external pilot flame. The lowest initial air temperature at which a flash is observed during a 10-min period is recorded as the FIT.

The SIT is the minimum temperature at which the self-heating properties of the specimen lead to ignition or ignition occurs spontaneously, under specified test conditions, in the absence of any additional flame ignition source. The lowest initial air temperature at which the specimen ignites during a 10-min period is recorded as the SIT.



Figure 3.1. Schematic of Southwest Research Institute Hot-Air Furnace

3.3 Thermal Conductivity

The thermal conductivity of the cesium-loaded SRF resin was measured using the method as described in ASTM D5930-09, *Standard Test Method for Thermal Conductivity of Plastics by Means of a Transient Line-Source Technique* and ASTM D5334-08, *Standard Test Method for Thermal Conductivity of Soil and Soft Rock by Thermal Needle Probe Procedure*. The Hukseflux TP02 Non-Steady-State Probe (NSSP) was used for the analysis. A Global Specialties 1305 Dual Output DC Power Supply (0-32V, 5A) was used to provide a constant current of 0.2A to the probe. Measurement Computing's USB-TEMP with TracerDAQ® was utilized to acquire the time and temperature at 1 s intervals during the duration of the test. The test material was placed into a glass cylindrical vessel with a diameter of 55 mm and a length of 200 mm. The thermal conductivity probe was inserted into the material and temperature was monitored to ensure thermal equilibrium had been reached prior to beginning the test. The time the power remained on (180 s) was the "heating phase". Once the power was turned off, the temperature was acquired for an additional 6 min for the "cooling phase".

The NSSP principle relies on a unique property of a line source. After a short transient period the temperature rise is given as:

$$\Delta T = \frac{Q}{4\pi\lambda} \ln(t) \quad 0 < t \le t_1 \tag{1}$$

$$\Delta T = \frac{Q}{4\pi\lambda} \ln\left(\frac{t}{t-t_1}\right) \quad t > t_1 \tag{2}$$

where ΔT is temperature rise from time zero (K), Q is heat input per unit length of heater (W/m), λ is thermal conductivity [W/(mK)], t is time from the beginning of heating (s), and t_1 is heating time (s).

The thermal conductivity is calculated from the slope of the line representing the temperature versus $\ln(t)$ or $\ln[t/(t - t_1)]$. A graph was constructed for each set of data (heating and cooling). The early and late portions of the test were not used in the slope analysis. For the heating phase, the slope was calculated from the $\ln(t)$ range of 2 to 3, which corresponds to 7 to 148 s. For the cooling phase, the slope was calculated for the $\ln[t/(t - t_1)]$ range of 0.5 to 3, which corresponds to 189 to 457 s. The thermal conductivity for the heating and cooling phases was calculated from the equations:

$$\lambda = \frac{CQ}{4\pi S} \tag{3}$$

$$Q = \frac{I^2 R_W}{L} \tag{4}$$

where *C* is calibration constant and *S* is slope obtained from temperature versus $\ln(t)$ or $\ln[t/(t - t_1)]$, *I* is current flowing through heater wire (A), R_W is total resistance of heater wire, and *L* is length of heated needle (m). The final conductivity reported was calculated as the average of the heating and cooling phase thermal conductivities.

The thermal conductivity probe was calibrated before use with freshly prepared Agar water (2.5 g Agar and 500 mL deionized water). The calculated thermal conductivity differed slightly from Agar water's published thermal conductivity of 0.607 W/(mK); therefore, the calibration factor, C, was 0.91659. Freshly purchased and opened glycerol was used for the calibration verifications. The thermal conductivity for the initial and final verifications was 100% of published value for glycerol.

3.4 Specific Heat Capacity

The specific heat capacity of the cesium-loaded SRF resin was measured by differential scanning calorimetry (DSC) in accordance with ASTM E1269-11, *Standard Test Method for Determining Specific Heat Capacity by Differential Scanning Calorimetry*. The test consists of heating the sample in a controlled atmosphere through a temperature region of interest. The difference in heat flow in the sample compared to a reference material and a blank pan are used to calculate the specific heat capacity.

Specific heat capacity measurement was performed using a Seiko DSC 210. The instrument was purged with 50 mL/min of nitrogen throughout the testing. For the DSC temperature program, the temperature was held isothermally at 25°C for 4 min; the temperature was then ramped at 20°C/min to 100°C and held for 5 min. The thermal curve was collected from each analytical run for sample,

reference material, and blank pan. The thermal curve for the blank aluminum pan was collected to perform the blank subtraction. The DSC software performs the blank pan subtraction from samples and reference material automatically. The blank subtracted sample and reference material files were imported into the specific heat capacity calculation program. The program generated a specific heat capacity curve along with a table of specific heat capacities for the temperature range from 40°C to 90°C at 1°C interval for which sample heating rate was stable. Sapphire disk from the Perkin Elmer Specific Heat Kit (PN# 02190136) was used as the reference material. The instrument response was standardized against the Perkin Elmer sapphire disk and all sample results were reported using this standardization.

3.5 DSC-TGA-FTIR

Thermal and infrared analysis of the cesium-loaded SRF resin was performed using a TA Instruments Q600 SDT Simultaneous differential scanning calorimetry-thermo-gravimetric analysis (DSC-TGA) Heat Flow Analyzer and a Nicolet Magna-IR 560 Fourier Transform Infrared (FTIR) Analyzer. Two instruments were connected with a Nicolet TGA Interface. This allowed real time infrared analysis of the evolved gases from the thermal decomposition or volatilization of the sample.

DSC-TGA analysis was performed on a 14.39 mg sample placed in an alumina pan. The sample was heated from room temperature to 1000°C at a ramp rate of 20°C/min. An air purge flow rate of 60 mL/min was used to sweep the evolved gases through a heated transfer line (180°C) and then through the heated flow cell (225°C) of the TGA interface. The Nicolet Magna-IR 560 monitored the composition of the gas.

Main Analysis/Property	Test Equipment(s)	Standard Procedures used by SwRI
Flash ignition temperature	Hot-air ignition	ASTM D1929-96, Standard Test Method for
and spontaneous ignition temperature	furnace	Determining Ignition Temperature of Plastics
Thermal conductivity [W/(mK)]	Non-steady-state probe with transient line-source technique	ASTM D5930-09, Standard Test Method for Thermal Conductivity of Plastics by Means of a Transient Line-Source Technique ASTM D5334-08, Standard Test Method for Thermal Conductivity of Soil and Soft Rock by Thermal Needle Probe Procedure
Specific heat capacity[J/(gK)]	Differential scanning calorimeter	ASTM E1269-11, Standard Test Method for Determining Specific Heat Capacity by Differential Scanning Calorimetry
Thermal and infrared analyses	Differential scanning calorimeter- thermo-gravimetric analyzer with Fourier transform Infrared analyzer	Not applicable

Table 3.1. Standard Procedures Used by Southwest Research Institute

4.0 Results

The results of the cesium-loaded SRF resin fire tests reported by SwRI are summarized in this section. Where applicable, the precision and bias of each test method as given in each ASTM procedure are included. The precision and bias measures used in ASTM standards are defined as follows, per ASTM E177-10, *Standard Practice for Use of the Terms Precision and Bias in ASTM Test Methods*:

- *precision*: the closeness of agreement between independent test results obtained under stipulated conditions
- *bias*: the difference between the expectation of the test results and an accepted reference value
- repeatability: precision under repeatability conditions
- *repeatability conditions*: conditions where independent test results are obtained with the same method on identical test items in the same laboratory by the same operator using the same equipment within short intervals of time
- *repeatability limit (r)*: the value below which the absolute difference between two individual test results obtained under repeatability conditions may be expected to occur with a probability of approximately 0.95 (95%)
- reproducibility: precision under reproducibility conditions
- *reproducibility conditions*: conditions where test results are obtained with the same method on identical test items in different laboratories with different operators using different equipment
- *reproducibility limit (R)*: the value below which the absolute difference between two test results obtained under reproducibility conditions may be expected to occur with a probability of approximately 0.95 (95%).

4.1 Flash Ignition Temperature and Spontaneous Ignition Temperature

The results of FIT and SIT measurements of the cesium-loaded SRF resin and negative and positive standards are summarized in Table 4.1. The negative test was performed on marinite, which is a non-combustible material and positive test was performed on poly methyl methacrylate (PMMA). Table 4.1 also includes the repeatability and reproducibility ranges determined from interlaboratory tests involving seven laboratories, on six polymeric materials, with triplicate tests of each material (ASTM D1929-96). The difference between two tests in this study is 0°C for SIT and 10°C for FIT, which is within the range of the repeatability limits for FIT (8 to 13°C). It should be noted that the ASTM precision estimate is for the difference between averages determined from triplicate tests, whereas the difference reported in this study is based on one test per each run. Information on bias of this test method is not available. It is noted that the FIT is 40°C lower than the previous results for the SRF resin without cesium although SIT is only 3°C lower, which is well below the repeatability limits of 11 to 31°C.

Property	FIT (°C)	SIT (°C)			
Initial Run	550	610			
Duplicate Run	560	610			
Average	555	610			
Difference	10	0			
Precision estimates ^(a)					
r	8-13	11-31			
R	27-117	47–103			
Negative and positive standards					
Marinite (negative standard)	No ignition	No ignition			
PMMA (positive standard)	390	480			
(a) Determined from interlaboratory tests involving seven laboratories on six polymeric materials with triplicate tests (ASTM D1929-96).					

 Table 4.1.
 Flash Ignition Temperature and Spontaneous Ignition Temperature of Cesium-Loaded SRF Resin

4.2 Thermal Conductivity Results

Table 4.2 summarizes thermal conductivities of the cesium-loaded SRF resin measured at three temperatures with three measurements at a minimum at each temperature.

Temperature (°C)	Thermal Conductivity [W/(mK)]	Average Thermal Conductivity [W/(mK)]
22	0.0855	0.0834
22	0.0816	
22	0.0860	
22	0.0806	
45	0.0872	0.0882
45	0.0899	
45	0.0874	
90	0.1010	0.1018
90	0.1021	
90	0.1023	

Table 4.2. Thermal Conductivity with Temperature

The thermal conductivity was plotted against temperature and a linear least-square fit was established as shown in Figure 4.1. The thermal conductivity at various temperatures between 22 and 90°C can be estimated from the following linear regression equation:

$$\lambda = 0.000273T + 0.0769 \quad (22^{\circ}C \le T \le 90^{\circ}C) \tag{5}$$

where λ is thermal conductivity [W/(mK)] and *T* is temperature (°C).

The results of an interlaboratory study involving this test method used for Ottawa sand and paraffin wax indicated a measurement precision of $\pm 10\%$ and $\pm 15\%$, respectively, with a tendency to a positive bias (higher value) over the known values for the materials studied (ASTM D5334-08).



Figure 4.1. Thermal Conductivity as a Function of Temperature

4.3 Specific Heat Capacity Results

Table 4.3 summarizes the results of specific heat capacity measurements following the standard procedure ASTM E1269-11 for temperatures between 40°C and 90°C. Table 4.3 summarizes the results of duplicate tests and their average and Figure 4.2 shows the specific heat capacity as a function of temperature curves for duplicates tests and average.

The precision of the test method was determined from interlaboratory trials involving seven laboratories on three different solid materials (biphenyl ether, National Institute of Standards and Technology (NIST) linear polyethylene, and indium metal) over the temperature range from 40°C to 80°C. The relative repeatability limit was 6.2% and relative reproducibility limit was 8.4%. The relative difference from duplicate tests in this study ranged from 1.9% (at 40°C) to 3.7% (at 90°C), which is smaller than the relative repeatability limit of 6.2%.

The bias of the test method was obtained from the above mentioned interlaboratory study by comparing the mean specific heat capacity at 67°C to the literature value: the relative difference from the corresponding literature value was +0.95% for biphenyl ether, -1.1% for NIST linear polyethylene, and +0.8% or +1.8% for indium metal from two different references.

Specific Heat Capacity [J/(gK)]		• •	Specific Heat Capacity [J/(gK)]					
Т, С	1st run	Duplicate	Average		Т, С	1st run	Duplicate	Average
40	1.420	1.447	1.434		66	1.643	1.688	1.666
41	1.432	1.459	1.446		67	1.653	1.699	1.676
42	1.442	1.469	1.456		68	1.664	1.712	1.688
43	1.450	1.478	1.464		69	1.675	1.724	1.700
44	1.459	1.486	1.473		70	1.687	1.736	1.712
45	1.468	1.495	1.482		71	1.699	1.748	1.724
46	1.477	1.505	1.491		72	1.713	1.763	1.738
47	1.487	1.516	1.502		73	1.727	1.780	1.754
48	1.494	1.525	1.510		74	1.741	1.794	1.768
49	1.501	1.532	1.517		75	1.755	1.809	1.782
50	1.509	1.541	1.525		76	1.770	1.826	1.798
51	1.515	1.549	1.532		77	1.787	1.845	1.816
52	1.523	1.560	1.542		78	1.804	1.864	1.834
53	1.531	1.569	1.550		79	1.818	1.879	1.849
54	1.538	1.576	1.557		80	1.830	1.893	1.862
55	1.545	1.585	1.565		81	1.844	1.908	1.876
56	1.554	1.594	1.574		82	1.858	1.923	1.891
57	1.562	1.603	1.583		83	1.871	1.938	1.905
58	1.569	1.609	1.589		84	1.885	1.953	1.919
59	1.577	1.616	1.597		85	1.898	1.968	1.933
60	1.585	1.624	1.605		86	1.912	1.984	1.948
61	1.595	1.634	1.615		87	1.926	1.998	1.962
62	1.604	1.644	1.624		88	1.941	2.013	1.977
63	1.615	1.655	1.635		89	1.954	2.026	1.990
64	1.624	1.666	1.645		90	1.965	2.037	2.001
65	1.633	1.676	1.655					

 Table 4.3.
 Specific Heat Capacity Test Results of SRF Resin



Figure 4.2. Specific Heat Capacity as a Function of Temperature

4.4 DSC-TGA-FTIR Results

Figure 4.3 shows the thermal behavior of the sample as displayed with weight change and heat flow plotted as a function of temperature. From the DSC analysis, the heat flow curve shows an endothermic transition near 102°C. This endotherm corresponded to gradual weight loss between room temperature and roughly 300°C with a total weight loss of approximately 25%. Small exothermic transitions occur around 387°C and 619°C and a large exothermic transition takes place near 864°C. Each of these exotherms corresponded to weight losses of approximately 30%, 5%, and 20%, respectively.

Figure 4.4 displays the weight change curve with the derivative overlaid. In Figure 4.5 the weight change, its derivative, and the heat flow curves are overlaid. The derivative of weight change aids in distinguishing some subtle transitions as shown in Figure 4.5. The first weight loss peak at 91°C may be due to moisture. Major weight loss peaks occur at 356°C and 866°C. Other smaller peaks appear at 340, 365, 375, 614 and 624°C.



Figure 4.3. Weight Change from TGA and Heat Flow from DSC of the Cesium-Loaded SRF Resin



Figure 4.4. Weight Change and Its Derivative from TGA of the Cesium-Loaded SRF Resin



Figure 4.5. Overlay of Weight Change, Its Derivative, and Heat Flow

FTIR analysis utilized a Gram-Schmidt reconstruction to plot the change in infrared intensity as a function of time is shown in Figure 4.6. This was used to correlate infrared spectra with thermal events occurring during the temperature ramp on the TGA. The labeled peaks correspond closely with the derivative curve peaks shown in the TGA plots. Table 4.4 summarizes the FTIR analysis results produced by adding the data in a selected time range for the major peaks in Figure 4.6. The FTIR spectra collected from 2.95-3.89 min (84-103°C) show noisy H_2O peaks at the wavenumber range of 4000-3400 cm⁻¹ and 2000-1300 cm⁻¹. All other spectra collected from the peaks approximately centered at 391, 575, 811, 863, and 931°C show CO₂ peaks at 3900-3500 cm⁻¹, 2400-2200 cm⁻¹, and 750-600 cm⁻¹. No peaks indicative of hydrocarbons or CO gas were observed.

For information purposes, the DSC-TGA was performed using a N_2 purge of 100 mL/min instead of air. Figure 4.7 shows the DSC-TGA results performed under N_2 purge and Figure 4.8 compares the TGA curves from tests under air and N_2 . Under the N_2 purge the sample does not show the major weight loss peaks observed at 356 and 866°C but rather shows continuous weight loss from 150°C to the end of the test. A slight weight increase observed at 750°C from the test under N_2 is likely an experimental noise.



Figure 4.6. Gram-Schmidt Reconstruction for FTIR of the Cesium-Loaded SRF Resin

Gram-Schmidt Reconstruction			
Peak time (min)	Time range (min)	Temperature range (°C)	FTIR Results
3.1	2.95 - 3.89	84 - 103	Peaks at 4000-3400 cm ⁻¹ and 2000-1300 cm ⁻¹ , representing H_2O
18.3	16.50 - 19.18	355 - 409	
27.5	27.09 - 28.03	567 - 586	
39.3	37.02 - 40.10	765 - 827	Peaks at 3900-3500 cm ⁻¹ , 2400-2200 cm ⁻¹ , and $750-600 \text{ cm}^{-1}$ representing CO ₂
41.9	40.77 - 43.86	840 - 902	750 000 cm , representing CO ₂
45.3	44.13 - 46.14	908 - 948	



Figure 4.7. DSC-TGA in N₂ for the Cesium-Loaded SRF Resin



Figure 4.8. Comparison of DSC-TGA Curves in Air and N_2 for the Cesium-Loaded SRF Resin

5.0 Summary

The average FIT and SIT of the cesium-loaded SRF resin were 555°C and 610°C, respectively, which were 40°C (for FIT) and 3°C (for SIT) lower than the previous results for the SRF resin without cesium. The thermal conductivity (λ in W/[mK]) of the cesium-loaded SRF resin as a function of temperature (T in °C) between 22 and 90°C was expressed by a linear regression equation, $\lambda = 0.000273T + 0.0769$. The average specific heat capacity of the cesium-loaded SRF resin ranged from 1.434 J/(gK) at 40°C to 2.001 J/(gK) at 90°C. From the TGA-DSC analysis, an endothermic transition near 102°C appeared with a total weight loss of approximately 25%. Small exothermic transitions occurred around 387°C and 619°C and a large exothermic transition near 864°C with corresponding weight losses of approximately 30%, 5%, and 20%, respectively. The result of FTIR analyses showed that the weight loss peak at 102°C is linked to H₂O while all other peaks at higher temperature were linked to CO₂ gas.

6.0 References

ASTM D1929-96. 1996. *Standard Test Method for Determining Ignition Temperature of Plastics*, ASTM International, West Conshohocken, PA.

ASTM D5334-08. 2008. *Standard Test Method for Thermal Conductivity of Soil and Soft Rock by Thermal Needle Probe Procedure*, ASTM International, West Conshohocken, PA.

ASTM D5930-09. 2009. Standard Test Method for Thermal Conductivity of Plastics by Means of a Transient Line-Source Technique, ASTM International, West Conshohocken, PA.

ASTM E1269-11. 2011. *Standard Test Method for Determining Specific Heat Capacity by Differential Scanning Calorimetry*, ASTM International, West Conshohocken, PA.

Kim, D, MJ Schweiger, and RA Peterson. 2012. Fire Safety Tests for Spherical Resorcinol Formaldehyde Resin: Data Summary Report, PNNL-21321, WTP-RPT-218, Rev 0, Pacific Northwest National Laboratory, Richland, WA.

Not Publically Available

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TP-WTPSP-002, Rev 3.0. 2011. Cesium Ion Exchange Simulant Testing in Support of M-6, Pacific Northwest National Laboratory, Richland, WA.
Appendix A

SwRI Report

6220 CULEBRA RD. 78238-5166 • P.O. DRAWER 28510 78228-0510 • SAN ANTONIO, TEXAS, USA • (210) 684-5111 • WWW.SWRI.ORG

Chemistry and Chemical Engineering Division Department of Analytical and Environmental Chemistry

May 22, 2012

Battelle Memorial Institute - PNNL MSIN K6-24 790 6th Street Richland, Washington 99352

Attn: Ms. Renee Russell

Subject:Purchase Order No.:167930SDG Number:486163SwRI Project No.:13295.12.00XSwRI Task Order Number:120319-5SwRI Sample Receipt Number:46112Samples Received:11.11.11, 02.15.12Required Analysis:Various/ See RFP 190709

Dear Ms Russell:

Please find the enclosed revised results for the sample and additional volume received on the above referenced date. Please note that this revision reflects updates to the Narrative and the FTIR Section. Should you have any questions, please do not hesitate to call me at 210-522-3242.

Sincerelv Radonna Spies

Group Leader

APPROVED:

Michael J. Dammann Director

RS: aa

Encl



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SAN ANTONIO, TEXAS, USA

 (210) 684-5111
 WWW.SWRLORG
 Chemistry and Chemical Engineering
 Department of Analytical and Environmental Chemistry

April 23, 2012

Battelle Memorial Institute - PNNL MSIN K6-24 790 6th Street Richland, Washington 99354

Attention: Renee Russell

Subject:

Purchase Order No.: SDG Number: SwRI Project No.: SwRI Task Order No.: SwRI Sample Receipt No.: Samples Received: Required Analysis: 167930 486163 13295.12.00X 120319-5 46112 11/11/11, 02/15/12 Various / See RFP 190709

Dear Ms. Russell,

Please find the enclosed results for the sample and additional volume received on the above referenced dates. Should you have any questions, please feel free to contact me at (210) 522-3242, radonna.spies@swri.org.

Best Regards, Radonna Spi**e**s

Group Leader

Aichael L. Dammann Director RS:dr



NARRATIVE



NARRATIVE

The spherical resorcinol formaldehyde resin was loaded with cesium according to the procedure described in Appendix B of the SOW. In addition to a thermal gravimetric off-gas analysis, the thermal conductivity and specific heat capacity were measured. The resin was dried at 95 °C prior to testing.

Thermal Conductivity

The thermal conductivity of the resin was measured using the transient heat method as described in ASTM D5334. The Hukseflux TP02 Non-Steady-State Probe (NSSP) for thermal conductivity measurement was used for the analysis. A Global Specialties 1305 Dual Output DC Power Supply (0-32V, 5A) was used to provide a constant current of 0.200 A to the probe when initiated. Measurement Computing's USB-TEMP with TracerDAQ® was utilized to acquire the time and temperature at 1 second intervals during the duration of the test. The material was place in to glass cylindrical vessel having a diameter of 55 mm and a length of 200 mm. The thermal conductivity probe was inserted into the material and temperature monitored to ensure thermal equilibrium had been reached prior to beginning the test. The program was setup to monitor the stabilized temperature for 1 minute prior to beginning the test. The power was turned on 1 minute into the test and turned off 4 minutes into the test. The time the power remained on (180 s) was the "heating phase". Once the power was turned off, the temperature was acquired for an additional 6 minutes; this was the 'cooling phase".

The NSSP principle relies on a unique property of a line source: after a short transient period the temperature rise, ΔT , only depends on heater power, Q, and medium thermal conductivity, λ :

$$\Delta T = (Q / 4 \pi \lambda) (\ln t) \qquad 0 < t < t_1$$

$$\Delta T = (Q / 4 \pi \lambda) (\ln t / (t-t_1)) \qquad t > t_1$$

Where:

 ΔT = temperature rise from time zero (K) Q = heater input per unit length of heater (W/m) t = time from the beginning of heating (s) t₁ = heating time

Therefore, by measuring the heater power (Q) and recording temperature (T) versus time (t), the thermal conductivity (λ) can be calculated since the temperature rise varies linearly with the logarithm of time. The thermal conductivity is equivalent to the slope of the line representing the temperature vs ln t for the heating phase, and the ln(t/(t-t1)) for the cooling phase. A graph was constructed for each set of data (heating and cooling). The data graphed was evenly space with the logarithm of time (x-axis). The early and late portions of the test were not used in the slope analysis. For the heating phase, the slope was calculated from the ln(t) range of 2 to 5, which corresponds to 7 - 148 seconds. For the cooling phase, the slope was calculated for the ln(t/(t-t1)) range of 0.5 to 3, which corresponds to 189 - 457 seconds. The thermal conductivity for the heating and cooling phases was calculated using the equation below, and the final thermal conductivity reported was

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Client: Battelle Memorial PNNL SDG: 486163 SwRI Project Number: 13295.12.008 SwRI Task Order Number: 120319-5

calculated using the average of the heating and cooling phase thermal conductivities. $\lambda = (CQ/4 \pi S)$

Where: Q = I2 (R/L) = (EI/L) Q = heat input (W/m) C = calibration constant $\lambda = thermal conductivity [W/(m·K)]$ S = slope used to compute thermal conductivity t = time (s) I = current flowing through heater wire (A) R = total resistance of heater wire (ohm) L = length of heated needle (m), andE = measured voltage (V)

The thermal conductivity of the sample was measured in triplicate, at a minimum, and at three temperatures: 22 °C (ambient), 45 °C and 90 °C. The average thermal conductivity at each temperature was plotted and a linear best-fit trendline was established. Therefore, the thermal conductivity can be closely estimated using the following linear regression equation:

Thermal Conductivity, $\lambda = (0.0003 \text{ x T}) + 0.0767$

Where: T = temperature in $^{\circ}$ K or $^{\circ}$ C

The thermal conductivity probe was calibrated before use with freshly prepared Agar water (2.5 g Agar / 500mL deionized water). The calculated thermal conductivity differed slightly from water's published thermal conductivity of 0.607 W/mK; therefore, the calibration factor, C, was adjusted to 0.91659.

Freshly purchased and opened glycerol was used for the calibration verifications. The thermal conductivity for the initial and final verifications calculated to 100% of published value for glycerol.

Specific Heat Capacity

Specific heat capacity, Cp, was measured by differential scanning calorimetry (DSC) in accordance with ASTM E1269 and the instrument manufacturer's instructions. The test consists of heating the sample at a controlled atmosphere through a temperature region of interest. The difference in heat flow into the sample compared to a reference material and a blank pan are used to calculate the specific heat capacity. Samples were analyzed using a Seiko DSC 210. The instrument was purged with 50 mL/min of nitrogen throughout the testing. For the DSC temperature program, the temperature was held isothermally at 25 °C for 4 minutes; the temperature was then ramped to 100 °C and held for 5 minutes. The thermal curve was collected for each analytical run. The thermal curve for the blank aluminum pan was collected to perform the blank subtraction. The DSC software

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contains a specific heat capacity program that performs the blank pan subtraction from samples and reference standards automatically. The blank subtracted sample and standard files were imported into the Cp calculation program. The program generated a specific heat capacity curve (Cp vs Temp) along with a table of specific heat capacities from 40-90 °C at 1 °C intervals.

Sapphire disk (PN# 0219 1483) from the Perkin Elmer Specific Heat Kit (PN #0219 0136) was used as the standard material. The specific heat capacity of the sapphire disk was verified against NIST SRM 720 with the following results:

Perkii Enner Sappline Disk (021) 1485)								
Temp. °C	Published Cp J/g ^o K	Calculated Cp J/g ^o K	% Rec					
46.85	0.8194	0.76	92.8%					
56.85	0.8380	0.782	93.3%					
66.85	0.8556	0.798	93.3%					
76.85	0.8721	0.809	92.8%					
86.85	0.8878	0.809	91.1%					

Perkin Elmer Sapphire Disk (0219 1483)

The specific heat capacity of the sample was determined in duplicate. The RPDs of the heat capacities for the sample and duplicate at 40 °C and 90 °C were 1.88% and 3.55%, respectively.

DSC-TGA-FTIR

Thermal and infrared analysis on the Cs loaded resorcinol resin was accomplished using a TA Instruments Q600 SDT Simultaneous DSC-TGA Heat Flow Analyzer and a Nicolet Magna-IR 560 Fourier Transform Infrared Analyzer (FT-IR). The two instruments were connected with a Nicolet TGA Interface. This allowed the real time infrared analysis of the evolved gases from the thermal decomposition or volatilization of the sample material.

TGA analysis was performed on a 14.39 mg sample quantity placed in a ceramic alumina pan. The sample was heated from room temperature to 1000°C at a ramp rate of 20°C/min. An air purge gas flow rate of 60 ml/min was used to sweep the evolved gases through a heated transfer line (180°C) and then through the heated flow cell (225°C) of the TGA Interface. The Nicolet Magna-IR 560 monitored the composition of the gas in real time.

The thermal behavior of the sample was displayed with weight % and heat flow plotted as a function of temperature. From the TGA/DSC analysis (Figure 1) the heat flow curve shows an endothermic transition near 100 °C. Small exothermic transitions occur around 387 °C and 620° C. A large exotherm takes place near 864 °C. Each of these exotherms corresponded to weight losses of approximately 30%, 5% and 20%, respectively.

The weight loss curve with the derivative curve overlay is presented in Figure 2. The derivative curve aids in distinguishing some subtle transitions.

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Figure 3 is the weight loss curve with the derivative curve and heat flow curve overlaid. The first weight loss at around 91 °C may be due to moisture or volatiles. Major weight losses occur at approximately 350 °C and 850 °C. Other smaller disassociations appear around 340, 365, 375, 614, and 624 °C.

FTIR analysis utilized a Gram-Schmidt reconstruction to plot (Figure 4) the change in infrared intensity as a function of time. This was used to correlate infrared spectra with thermal events occurring during the temperature ramp on the TGA. The labeled peaks correspond closely with the derivative curve peaks shown in the TGA plots.

The linked spectrum (Figure 5) is the first spectrum collected near the start of the experiment. Subsequent "Coadded Spectrums" are single spectrums produced by coadding data in a selected time region of the Gram-Schmidt reconstruction profile. Coadded spectrums were chosen to relate to the TGA thermal events and time frames.

As the sample approached 100 °C, the FTIR spectra collected from 2.95-3.89 min (Figure 6) shows noisy moisture peaks appearing to develop in the 3500-4000 cm-1 and the 1300-2000 cm-1 ranges. In Figure 7, the FTIR acquisition during the first major weight loss at 350 °C (16.50-19.18 min) shows peaks at 2350 cm-1 and 670 cm which are attributed to carbon dioxide (CO₂). The following co-added spectrum at 27-09-28.03 min (Figure 8) contained the same CO₂ peaks but at lower intensity. Figures 9 and 10 are spectra collected during the second major weight loss event at 850 °C (37.02-40.10 min and 40.77-43.86 min, respectively). In addition to the large peaks CO₂ peak at 2350 cm-1, numerous peaks in the 3500-4000 cm-1 and 620-720 cm-1 ranges are present and are indicative of CO₂. During the analysis, no peaks indicitive of hydrocarbons (2800-3200 cm-1) were observed.

TGA analysis in Nitrogen

For informational purposes, the thermogravimetric analysis was repeated using a nitrogen purge of 100 mL/min instead of air and covering the sample containers.

"I certify that this data package is in compliance with the terms and conditions of the contract, both technically and for completeness, for other than the conditions detailed above. Release of the data contained in this hardcopy data package and in the computer-readable data submitted on diskette has been authorized by the laboratory manager or his/her designee, as verified by the following signature. This report shall not be reproduced except in full without the written approval of SwRI."

05/22/12

Group Leader

Date

000006

SOUTHWEST RESEARCH INSTITUTE CLIENT: Battelle Memorial Ins. PNNL TASK ORDER#: 120319-5 SRR#: 46112 SDG#: 486163 VTSR: 120319-5 PROJECT #: 13295.12.008

SAMPLE RECEIPT, TASK ORDER & CHAIN OF CUSTODY

Revision: 3

This Receipt was Revised Apr 23 2012 4:05PM

VTSR: 11/11/11

Time: 08:30:00

Manager: HORTON, GENE Logged in by: KBill Creation Date: 11/17/11

Project: 13295.12.008 Proposal #: 01-63774 Client: Battelle Memorial Institute

Southwest Research Institute

Notes

Container 1: Test Sample Container 2: Extra

Project is Nuclear Safety Related, 10 CFR 50, Part 21, and Appendix B. For Div 01 contact Joann Boyd @ X2169 or M. Valenti @ X3682. Have SPQP (Latest Revision) available at the time of prep or analysis of samples. All personnel must be QA Nuclear Certified.

POC: Eugene Horton X3457.

Disposal Instructions: Contact Eugene Horton for sample disposal authorization.

*** KBill Nov 17 2011 1:58PM ***

Revised to add additional sample SWRI delivered at 10:50AM on 02/15/2012. *** KBill Feb 15 2012 1:51PM ***

System ID	Customer ID	CED	Matrix	Containers	Special Reqs.	
486163	(2/15) CS Loaded Resorcinol Resin			2		
479505	Resorcinol-Formaldehyde Polymer			2		
Contain	ers: 4		Samples: 2			

These documents are associated with this receipt: 109148[RFP #190709], 109444[Tech Prop], 109445[Cost Prop], 109773[Contract 167930], 109898[COC for SRR 46069], 109900[MSDS for SRR 46069], 110146[Sample Pic], 113577[(2/15) Sample Pic], 115168[Form 170]

Thermometer: Temperature: Management of the second s

RM-DD5

Laboratory Task Order

TO #: 120319-5 Revision: 2

SDG: 486163 VTSR: 11/11/11, 02/15/12 CASE: 167930 SRR #'s: 46112 Client(s): Battelle Memorial Institute Project(s): 13295.12.008 Manager(s): HORTON, GENE To PM: 03/08/12 To QA: 03/09/12 To Client: 03/09/12

Instructions · MEASURE THE SODIUM-FORM RF RESIN BED CONDUCTIVITY AND CAPACITY. HEAT CAPACITY VIA DSC ASTM E1269 THERMAL CONDUCTIVITY VIA ASTM D5334/D5930 · PERFORM A TGA (THERMAL GRAVIMETRIC ANALYSIS) WITH MASS SPECTROMETRY OF THE RESIN OFFGAS UP TO THE AUTO IGNITION TEMPERATURE AND DETERMINE CONSTITUENTS. THIS WILL BE ACHIEVED IN SEVERAL STEPS SINCE SWRI DOES NOT HAVE A TGA-MS SETUP. WE DO HAVE A TGA-FTIR SETUP. THE SAMPLE WILL FIRST BE RUN VIA TG/DTA-FTIR (THERMAL GRAVIMETRIC / DIFFERENTIAL THERMAL ANALYSIS- FOURIER TRANSFORM INFRARED SPECTROMETRY) THIS TECHNIQUE WILL ALLOW US TO DETERMINE THE MASS LOSS VS. TEMPERATURE DATA WHILE MEASURING ANY THERMAL TRANSITIONS, AND THE FTIR WILL ALLOW IDENTIFICATION OF THE GASSES RELEASED DURING COMBUSTION. THESE TESTS WILL BE DONE USING AIR AS THE ATMOSPHERIC GAS. AFTER REVIEWING THE DATA, IF SUFFICIENT EVIDENCE IS OBTAINED THAT A MASS SPECTROMETRY ANALYSIS OF THE OFF GAS WOULD BE BENEFICIAL, A SECOND RUN WILL BE PERFORMED AND GRAB GAS SAMPLES, TRAPPED VIA SUMMA CANISTER AT THE APPROPRIATE INFLECTION POINT(S) OF THE TGA CURVE WILL BE TAKEN. THESE SAMPLES WILL BE RUN VIA GC AND/OR GC-MS FOR EXPECTED SPECIES. REVISION 1, DRMZ 04/20/12: TASK ORDER REVISED TO ADD ASTM D1929, IGNITION TEST, ALONG WITH A NOTE INDICATING ORIGINAL SAMPLE RECEIVED ON 11/11/11 AND ADDITIONAL VOLUME ON 02/15/12. REVISION 2, DRMZ 04/23/12: TASK ORDER REVISED TO INDICATE THE INITIAL RESORCINOL-FORMALDEHYDE POLYMER SAMPLE (LAB SYSTEM ID 479235) WAS RECEIVED AND LISTED IN SAMPLE RECEIPT REPORT # 46069. WHEN THE SAMPLE WAS TRANSFERRED TO FIRE TECH, THAT SECTION CREATED SRR 46112 TO NOTATE IT/TRACK IT UNDER THEIR SYSTEM AND ALSO TO LIST THE ADDITIONAL SAMPLE RECEIVED ON 02/15/12.

Documents Related to this task order: 109148[RFP #190709], 109444[Tech Prop], 109445[Cost Prop], 109773[Contract 167930], 109898[COC for SRR 46069], 109900[MSDS for SRR 46069], 110146[Sample Pic], 113577[(2/15) Sample Pic], 115168[Form 170]

Deliverables --> Hard Copy: no EDD: no PDF: -YES-

Test: ASTM D Section: FT AI	1929 DMIN	Hold	ling: 30 days f Indard Test Me	from VTSR ethod For Determining Ignition Temperature Of Plastics				
System ID	Туре	Cont	Matrix	Customer ID	VTSR	Method Date		
486163		1		(2/15) CS Loaded Resorcinol Resin	11 Nov 11	11 Dec 11		

Test: DSC_Sw						
Section: WETCHEM Differentia		Matrix	Customer ID	CED	Method Date	
486163		1		(2/15) CS Loaded Resorcinol Resin		

Test: WET-MI	SC	Hold	ding: 60 days f	rom CED		
Section: WETCHEM Any miscellaneous wetchem test.				is wetchem test.		Cnt: 1
System ID	Туре	Cont	Matrix	Customer ID	CED	Method Date
486163		1		(2/15) CS Loaded Resorcinol Resin		

Test: WET-SC Section: WET	LID CHEM	Ho	olding: 60 days fr .ny miscellaneou	s from CED ous wetchem solid test.				
System ID	System ID Type C		Matrix	Customer ID	CED	Method	Date	
486163		1		(2/15) CS Loaded Resorcinol Resin				

General Sample Chain of Custody

Ver (8/25/2011)

Southwest Research Institute

Chemistry and Chemical Engineering Division

6220 Culebra Road

San Antonio, Texas 78238-5166

SRR #:	46069	Thermometer:	027
Project	13295.12.00X	Temperature:	22.0
CASE:	167930	Airbill #:	404742686287
Customer:	BATTELLE PNNL	Logged in by:	DROMAN
Samples Received:	Nov 11 2011 8:30AM	Logged in:	Nov 11 2011 9:43AM
Manager:	HORTON, GENE		

Disposal Contact Eugen Horton for sample disposal authorization. *** DROMAN Nov 11 2011 9:43AM ***

System ID	Customer ID	CED	Matrix	# Cont	Special Requirements	Sample Condition	
479235	RESORCINOL-FORMALDEHYDE POLYMER		Intact				
Reliquished b	y (Print/Signature):				Date	Time	
	FED FX				which	0830	
Received by (Print/Signature): DINO NOMAN	5	N	_	Date	Time	
Reliquished b	y (Print/Signature):				Date	Time	
Received by (Print/Signature):				Date	Time	
Reliquished b	y (Print/Signature):		<u> </u>		Dəte	Time	

ASTM D 1929 – 96 (Re-approved 2011), Standard Test Method for Determining Ignition Temperature of Plastics



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ASTM D 1929 - 96 (Rc-approved 2011), STANDARD TEST **METHOD** FOR DETERMINING **IGNITION TEMPERATURE OF PLASTICS**

MATERIAL ID: RESORCINOL FORMALDEHYDE RESIN

FINAL REPORT Consisting of 10 Pages

SwRI® Project No. 01.13295.12.008g Test Date: February 17, 20-24, 2012 Report Date: April 23, 2012

Prepared for:

Battelle Memorial Institute 902 Battelle Blvd, K6-79 P.O. Box 999 Richland, WA 99352

Submitted by:

Eugene Horton

Senior Engineering Technologist Material Flammability Section

Approved by:

Matthew S. Blais, Ph.D. Director Fire Technology Department

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1.0 INTRODUCTION

ASTM D 1929, Standard Test Method for Determining Ignition Temperature of Plastics, covers a laboratory determination of the spontaneous ignition temperature (SIT) and flash ignition temperature (FIT) of plastics using a hot-air furnace. The hot-air ignition furnace consists primarily of an electrical heating unit and specimen holder. The furnace tube is a vertical tube with an inside diameter of 100 ± 5 mm and a length of 230 ± 20 mm, made of ceramic that will withstand at least 750 °C. The inner ceramic tube, with an inside diameter of 75 ± 5 mm, a length of 230 ± 20 mm, and a thickness of approximately 3 mm, is placed inside the furnace tube and positioned 20 ± 2 mm above the furnace floor on spacer blocks. The pilot flame is located immediately above the opening. The test apparatus is shown in Figure 1.



Figure 1. Schematic of Southwest Research Institute (SwRI) Hot-Air Furnace.

SIT is the minimum temperature at which the self-heating properties of the specimen lead to ignition or ignition occurs of itself, under specified test conditions, in the absence of any additional flame ignition source. The lowest air temperature at which the specimen ignites during a 10-min period is recorded as the spontaneous ignition temperature.

FIT is the minimum temperature at which, under specified test conditions, sufficient flammable gases are emitted to ignite momentarily upon application of a small external pilot flame. The lowest air temperature at which a flash is observed during a 10-min period is recorded as the flash ignition temperature.

2.0 SAMPLE IDENTIFICATION AND PREPARATION

Battelle Memorial Institute, located in Richland, Washington, provided a material identified as *Resorcinol Formaldehyde Resin* for testing in accordance with ASTM D 1929. On November 11, 2011, SwRI received approximately 20 liters of the specimen. further described in Table 1. On February 15, 2012, the Material Flammability Section received 1 bottle of Cesium loaded *Resorcinol Formaldehyde Resin* processed by our chemistry department, per instructions from the client.

Material ID	Description*	Color*	Received Mass*	Nominal Tested Mass*	
Resorcinol Formaldehyde Resin	Micro beads loaded with Cesium	Red	563.84 g	3 g	

TABLE 1. SAMPLE DESCRIPTION PROVIDED BY BATTELLE MEMORIAL INSTITUTE.

* Measured by SwRI personnel.

Upon receipt, the samples were placed in a controlled environment maintained at 23 °C \pm 2 °C (73 °F \pm 5 °F) and 50 % \pm 5% relative humidity for not less than 40 h prior to testing. SwRI personnel weighed the material to a specimen weight of ~3 g on the day of testing

3.0 RESULTS

Testing was conducted February 17, 20-24, 2012. Table 2 contains the results for the material provided by Battelle Memorial Institute. Test results are accurate to ± 5 °C, and are presented in detail at the end of this report. Two known material standards were tested to confirm equipment operation against known parameters. These results are also reported in Table 2. These test results relate only to the behavior of test specimens under the particular conditions of the test. They are not intended to be used, and shall not be used, to assess the potential fire hazards of a material in use.

Material ID	SIT	FIT
Resorcinol	610 °C	550 °C
Formaldehyde Resin with Cesium	1130 °F	1022 °F
	480 °C	390 °C
PMMA	896 ^o F	734 ⁰F
Marinite	Did Not Burn	Did Not Burn

ASTM D 1929 TEST DATA SHEET - SPONTANEOUS IGNITION

Client: Operator: Test Date(s): Project Number: Material ID*: Description*

Battelle Memorial Institute A. Lowry February 24, 2012 01.13295.12.008g Resorcinol-Formaldehyde Polymer (Cesium loaded) Small beads

Spontaneous February 15, 2012 Date Prepared by SwRI: Prior to Testing

Red Average Sample Mass: 3.00 g

Ignition Type:

Receipt Date:

Color:

SPONTANEOUS IGNITION TEMPERATURE (*C): 610

RESULTS

TestID	Initial	Final	Mass Loss	Initial T	Initial Temperature (°C)		Final T			
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(g)	(9)							
12-055BAT008N1S	3.00	0.87	2.13	598	600	650	598	598	648	No
12-055BAT008N2S	3.00	0.99	2.01	604	610	661	613	614	659	Yes
						740				

SPONTANEOUS IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Time	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test Time
	(min:s)	(min:s)	(min:s)					(min:s)
12-055BAT008N1S	:38	2:14	NA	No	Yes	No	No	10:38
12-055BAT008N2S	0:41	1:49	2:01	No	Yes	No	No	5:20

Resorcinol-Formaldehyde Polymer - Cesium Loaded



Note: Containment screen placed over the specimen holder

ASTM D 1929 TEST DATA SHEET - FLASH IGNITION

Client: Battelle Memorial Institute Ignition Type: Flash Operator: A. Lowry Receipt Date: February 15, 2012 Test Date(s): February 24, 2012 Date Prepared by SwRI: Prior to Testing 01.13295.12.008g Project Number: Material ID*; Resorcinol-Formaldehyde Polymer (Cesium loaded) Color: Red Description*: Small beads Average Sample Mass: 3.00 g

FLASH IGNITION TEMPERATURE (°C): 550

RESULTS

Test ID	Initial	Final	Mass	Initial	Initial Temperature (°C)			Final Temperature (°C)		
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(9)	(g)							-
12-055BAT008K8F	3.00	1.43	1.57	552	550	601	579	564	600	Yes
12-054BAT008K5F	3.00	1.41	1.59	540	540	591	552	548	590	No
12-054BAT008K6F	3.00	1.24	1.76	590	590	641	599	599	639	No
12-055BAT008K7F	3.00	1.40	1.60	574	570	620	594	583	616	No

FLASH IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Type	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test Time
	(min:s)	(min:s)	(min:s)	(min:s)				(min:s)
12-055BAT008K8F	0:43	NO	Flaming at 2:35	NO	Yes	NO	NO	5:00
12-054BAT008K5F	0:57	NO	NO	NO	Yes	NO	NO	10:58
12-054BAT008K6F	0:45	1:47	Flaming Flash	NO	Yes	NO	NO	10:46
12-055BAT008K7F	0:46	NO	Flaming Flash	NO	Yes	NO	NO	5:00

Resorcinol-Formaldehyde Polymer - Cesium Loaded



Note: Containment screen placed over the specimen holder

ASTM D 1929 TEST DATA SHEET - SPONTANEOUS IGNITION

Client Battelle Memorial Institute Ignition Type: Spontaneous Operator: A. Lowry Receipt Date: February 15, 2012 February 24, 2012 Test Date(s): Date Prepared by SwRI: Prior to Testing Project Number: 01.13295.12.008g Material ID* Resorcinol- Formaldehyde Polymer (Cesium loaded) Color: Red Description* Small beads Average Sample Mass: 3.00 g

SPONTANEOUS IGNITION TEMPERATURE (*C): 610

RESULTS

Test ID	Initial	Final	Mass Loss	Initial T	emperatu	re (°C)	Final T	ture (°C)		
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(g)	(g)							
12-055BAT008L2S	3.00	1.41	1.59	605	610	661	628	623	659	Yes
12-055BAT008L3S	3.00	0.96	2.04	597	600	650	599	602	651	NO
12-055BAT008L1S	3.00	1.32	1.68	619	620	672	637	632	671	Yes

SPONTANEOUS IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Time	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test Time
	(min:s)	(min:s)	(min:s)					(min:s)
12-055BAT008L2S	:44	1:28	1:48	NO	Yes	NO	NO	5:00
12-055BAT008L3S	0:47	1:58	NA	NO	Yes	NO	NO	10:48
12-055BAT008L1S	:43	1:35	1:43	NO	Yes	NO	NO	5:00
		-						

Resorcinol-Formaldehyde Polymer



NOTE: Duplicate Test Run

Screen placed over sample cup

ASTM D 1929 TEST DATA SHEET - FLASH IGNITION

Client:	Battelle Memorial Institute	Ignition Type:	Flash
Operator:	A. Lowry	Receipt Date:	February 15, 2012
Test Date(s):	February 24, 2012	Date Prepared by SwRI:	Prior to Testing
Project Number:	01.13295.12.008g		
Material ID*	Resorcinol- Formaldehyde Polymer (Cesium loaded)	Color:	Red
Description*:	Small beads	Average Sample Mass:	3.00 g

FLASH IGNITION TEMPERATURE (°C): 560

RESULTS

Test ID	Initial	Final	Mass	Initial Temperature (°C)			Final T			
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(g)	(g)	-						-
12-055BAT008M3F	3.00	0.95	2.05	569	560	611	575	570	607	Yes
12-055BAT008M2F	3.00	0.92	2.08	554	550	600	549	550	600	NO
12-055BAT008M1F	3.00	1.00	2.00	540	540	590	540	540	588	NO
							1			

FLASH IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Type	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test
	(min:s)	(min:s)	(min:s)	(min:s)				(min:s)
12-055BAT008M3F	0:40	NO	Flaming at 2:20	NO	Yes	NÖ	NO	5:06
12-055BAT008M2F	0:38	NO	NA	NO	Yes	NO	NO	10:38
12-055BAT008M1F	:44	NO	NA	NO	Yes	NO	NO	10:44



Resorcinal-Formaldehyde Polymer

Battelle Memorial Institute

Screen placed over sample cup

ASTM D 1929 TEST DATA SHEET - SPONTANEOUS IGNITION

Client: Operator: Test Date(s): Project Number: Material ID*:

Description*:

Battelle Memorial Institute A. Lowry February 17, 2012 01.13295.12.008g PMMA Solid Block
 Ignition Type:
 Spontaneous

 Receipt Date:
 February 15, 2012

 Date Prepared by SwR1:
 Prior to Testing

Color:BlackAverage Sample Mass:3.07 gAverage Sample Mass:3.07 g

SPONTANEOUS IGNITION TEMPERATURE (*C): 480

RESULTS

TestID	Initial	Final	Mass Loss	Initial Te	emperatu	re (°C)	Final T			
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(g)	(9)							
12-048BAT008G1S	3.15	0.64	2.51	414	410	466	415	408	465	No
12-048BAT008G2S	3.19	0.49	2.70	419	420	476	427	421	480	No
12-048BAT008G3S	2.94	0.07	2.87	430	430	486	447	436	486	No
12-048BAT008G4S	2.95	0.01	2.94	442	440	496	448	449	499	No
12-048BAT008G5S	3.00	0.72	2.28	489	480	536	540	784	551	Yes
12-048BAT008G6S	3.19	0.00	3.19	474	470	525	471	477	529	No

SPONTANEOUS IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Time	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test Time
	(min:s)	(min:s)	(min:s)					(min:s)
12-048BAT008G1S	0:35	NO	NA	NO	NO	Yes	Yes	10:36
12-048BAT008G2S	0:42	NO	NA	NO	NO	Yes	Yes	10:42
12-048BAT008G3S	0:34	NO	NA	NO	NO	Yes	Yes	10:34
12-048BAT008G4S	0:34	NO	NA	NO	NO	Yes	Yes	10:34
12-048BAT008G5S	0:33	NO	Flaming at 2:49	NO	NO	Yes	Yes	4:20
12-048BAT008G6S	0:33	NO	NA	NO	NO	Yes	Yes	10:34



PMMA Positive Standard

ASTM D 1929 TEST DATA SHEET - FLASH IGNITION

Client: Operator: Test Date(s): Project Number: Material ID*: Description*:

Battelle Memorial Institute A Lowry February 20-22, 2012 01.13295.12.008g PMMA Solid Block

Ignition Type: Flash Receipt Date: February 15, 2012 Date Prepared by SwRI: Prior to Testing

Black Average Sample Mass 3.08 g

Color:

FLASH IGNITION TEMPERATURE (°C): 390

RESULTS

Test ID	Initial	Finat	Mass	Initial T	emperatu	re (°C)	Final T	empera	ture (°C)	
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(g)	(g)							-
12-051BAT008H1F	3.19	1.66	1.53	436	440	497	463	502	504	Yes
12-052BAT008H2F	3.19	2.22	0.97	435	430	490	534	427	489	Yes
12-052BAT008H3F	3.07	2.13	0.94	420	420	475	352	421	476	Yes
12-052BAT008H4F	3.19	2.25	0.94	414	410	463	386	411	463	Yes
12-052BAT008H5F	3.10	1.94	1.16	402	400	454	490	405	455	Yes
12-052BAT008H6F	2.98	1.72	1.26	395	390	450	463	395	447	Yes
12-052BAT008H7F	2.84	1.29	1.55	384	380	435	369	373	434	NO

FLASH IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Type	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test Time
	(min:s)	(min:s)	(min:s)	(min:s)	Ũ			(min:s)
12-051BAT008H1F	0:42	NO	Flaming at 2:10	NO	NO	Yes	Yes	4:20
12-052BAT008H2F	0:45	NO	Flaming at 3:01	NO	NO	Yes	Yes	4:24
12-052BAT008H3F	0:40	NO	Flaming at 3:22	NO	NO	Yes	Yes	4:40
12-052BAT008H4F	0:35	NO	Flaming at 3:59	NO	NO	Yes	Yes	5:00
12-052BAT008H5F	0:37	NO	Flaming at 5:15	NO	NO	Yes	Yes	6:20
12-052BAT008H6F	0:40	NO	Flaming at 6:22	NO	NO	Yes	Yes	7:32
12-052BAT008H7F	0:35	NO	NA	NO	NO	Yes	Yes	10:36



9

PMMA Positive Standard

Battelle Memorial Institute

ASTM D 1929 TEST DATA SHEET - SPONTANEOUS IGNITION

Client	Bat
Operator:	A. L
Test Date(s):	Fet
Project Number:	01
Material ID*:	Ma
Description*:	Sol

Battelle Memorial Institute A. Lowry February 22, 2012 01.13295,12,008g Marinite Solid block

Ignition Type:	Spontaneous
Receipt Date:	NA
Date Prepared by SwRI;	Prior to Testing
Color:	White

Original Thickness: 20 mm Average Sample Mass: 3.17 g

SPONTANEOUS IGNITION TEMPERATURE (°C) : NA

RESULTS

Test ID	Initial	Final	Mass Loss	Initial Te	emperatu	re (°C)	Final T	empera	ture (°C)	
	Mass	Mass	Loss	Sample	Air	Furnace	Sample	Air	Furnace	Ignition
	(g)	(g)	(g)							
12-053BAT008I1S	3.17	2.65	0.52	760	750	813	759	755	811	No

SPONTANEOUS IGNITION OBSERVATIONS

	Insertion Time	Observed Smoke	Combustion Time	Observed Soot	Observed Charring	Observed Melt	Observed Bubbling	Total Test Time
	(min:s)	(m in :s)	(min:s)					(min:s)
12-053BAT008I1S	0:46	NA	NA	None	None	None	None	10:46

Marinite Negative Standard



Battelle Memorial Institute

WETCHEM ANALYSIS

Fraction: WETCHEM Pages: 02000 51 Af 4.23, 12

RAW DATA

Thermal Conductivity Data

Thermal Conductivity - ASTM D5334/D5930

Sample ID: Water Calibration 031612 (22-01-WCS6)

	Heating	Cooling	Average	Current:	<u>0.2</u> A	Heating:	On (s)	Off (s)	Delta
Slope:	<u>0.4309</u>	<u>0.4315</u>	0.4312	R:	<u>89.73</u> ohms/m	5	60	240	180
Therm Cond	0.663	0.662	0.662	Q:	3.5892 W/m		<u></u>		
TV	0.607	0.607	0.607	C:	1 Cal Factor				
% Rec	109%	109%	109%		-				

Heating Curve



	Beg	End	Delta	
Range	<u>2</u>	<u>5</u>	0.15	
	Goal In(t)	Time	ln(t)	Temp
	2	7	1.9459101	3.7853
	2.15	9	2.1972246	4.0095
	2.3	10	2.3025851	4.0518
	2.45	12	2.4849066	4.1648
	2.6	13	2.5649494	4.2025
	2.75	16	2.7725887	4.3091
	2.9	18	2.8903718	4.3642
	3.05	21	3.0445224	4.4772
	3.2	25	3.2188758	4.5277
	3.35	29	3.3672958	4.5644
	3.5	33	3.4965076	4.5791
	3.65	38	3.6375862	4.6204
	3.8	45	3.8066625	4.6994
	3.95	52	3.9512437	4.7123
	4.1	60	4.0943446	4.8408
	4.25	70	4.2484952	4.9308
	4.4	81	4.3944492	4.9969
	4.55	95	4.5538769	5.0125
	4.7	110	4.7004804	5.052
	4.85	128	4.8520303	5.153
	5	148	4.9972123	5.242

Thermal Conductivity - ASTM D5334/D5930

Sample ID: Water Calibration 031612 (22-01-WCS6)



1.251

1.3486

191 2.8543782

189 3.0445224

2.875

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	Glycerine ICV 031612					Actual Test Temperature:		22 °C		
	Heating	Cooling	Average	Current:	<u>0.2</u> A	Heating:	On (s)	Off (s)	Delta	
Slope:	<u>0.911</u>	<u>0.879</u>	0.895	R:	89.73 ohms/m	0	60	240	180	
Therm Cond	0.287	0.298	0.293	Q:	3.5892 W/m					
TV	0.292	0.292	0.292	C:	0.91659 Cal Factor					
%Rec	98.4%	102%	100%							

Heating Curve



	Beg	End	Delta	
Range	<u>2</u>	<u>5</u>	0.15	
	Goal In(t)	Time,s	ln(t)	т, °С
	2	7	1.9459101	29.1958
	2.15	9	2.1972246	29.4337
	2.3	10	2.3025851	29.5761
	2.45	12	2.4849066	29.7488
	2.6	13	2.5649494	29.8535
	2.75	16	2.7725887	30.039
	2.9	18	2.8903718	30.163
	3.05	21	3.0445224	30.2943
	3.2	25	3.2188758	30.4228
	3.35	29	3.3672958	30.5403
	3.5	33	3.4965076	30.644
	3.65	38	3.6375862	30.7357
	3.8	45	3.8066625	30.8605
	3.95	52	3.9512437	31.0027
	4.1	60	4.0943446	31.1651
	4.25	70	4.2484952	31.3393
	4.4	81	4.3944492	31.4934
	4.55	95	4.5538769	31.6108
	4.7	110	4.7004804	31.7235
	4.85	128	4.8520303	31.9307
	5	148	4.9972123	32.0865

Thermal Conductivity - ASTM D5334/D5930

Sample ID: <u>Glycerine ICV 031612</u>

Actual Test Temperature:

22 °C



Beg	End	Delta	
0.5	<u>3</u>	0.125	
Goal In(t/(t-t1))	Time,s	ln(t/(t-t1))	т, °С
0.5	457	0.5006659	26.2925
0.625	387	0.6257059	26.4234
0.75	341	0.7504781	26.5229
0.875	309	0.8735289	26.5938
1	285	0.9985288	26.7292
1.125	267	1.1213405	26.8342
1.25	252	1.252763	26.9696
1.375	241	1.3739231	27.0294
1.5	232	1.4954937	27.1307
1.625	224	1.6274564	27.2651
1.75	218	1.7469089	27.3691
1.875	213	1.8647846	27.5063
2	208	2.0053336	27.6167
2.125	204	2.1400662	27.7318
2.25	201	2.2587825	27.8523
2.375	198	2.3978953	27.9683
2.5	196	2.5055259	28.0639
2.625	194	2.6288008	28.166
2.75	192	2.7725887	28.2543
2.875	191	2.8543782	28.3638
3	189	3.0445224	28.5275

Thermal Conductivity - ASTM D5334/D5930

SwRIID: <u>486163 25</u>			Client ID: (2/15) Cs Loaded Resorcinol Resin			Actual Test Temp	22 °C	-	
Slope: Therm Cond	Heating <u>3.0559</u> 0.0857	Cooling <u>3.0711</u> 0.0852	Average 3.0635	Current: R:	0.2 A 89.73 ohms/m	Heating	On (s) <u>60</u>	Off (s) <u>240</u>	Delta 180
Thermal Con	ductivity, W/	/(m [·] K) =	0.0855	C:	0.91659 Cal Factor				



Beg	End	Delta	
2	5	0.15	
Goal In(t)	Time,s	ln(t)	т, °С
2	7	1.9459101	29.8388
2.15	9	2.1972246	30.4641
2.3	10	2.3025851	30.8385
2.45	12	2.4849066	31.3283
2.6	13	2.5649494	31.718
2.75	16	2.7725887	32.2194
2.9	18	2.8903718	32.5446
3.05	21	3.0445224	33.0309
3.2	25	3.2188758	33.625
3.35	29	3.3672958	34.0184
3.5	33	3.4965076	34.4529
3.65	38	3.6375862	34.8826
3.8	45	3.8066625	35.3897
3.95	52	3.9512437	35.8958
4.1	60	4.0943446	36.3559
4.25	70	4.2484952	36.7903
4.4	81	4.3944492	37.23
4.55	95	4.5538769	37.6695
4.7	110	4.7004804	38.1873
4.85	128	4.8520303	38.6419
5	148	4.9972123	39.0217

020008

Thermal Conductivity - ASTM D5334/D5930

SwRI ID:	<u>486163 25</u>	Client ID:	(2/15) Cs	Loaded Resorcinol Resin		Actual Test Tempe	erature:	22 °C	
Cooling Cu	J rve y = 3.0711x + 25.612 R ² = 0.9994				Range	Beg <u>0.5</u> Goal In(t/(t-t1)) 0.5	End <u>3</u> Time,s 457	Delta 0.125 In(t/(t-t1)) 0.5006659	т, °с 27.0976
5.5						0.625	387	0.6257059	27.5192
37 -						0.75	341	0.7504781	27.9057
35						0.875	309	0.8735289	28.3491
L 33				Cooling Raw		1	285	0.9985288	28.6599
- 55						1.125	267	1.1213405	29.0055
31 -		*****		Cooling Selected		1.25	252	1.252763	29.4898
29				—— Linear (Cooling Selected)		1.375	241	1.3739231	29.8425
27						1.5	232	1.4954937	30.1694
21 -						1.625	224	1.6274564	30.61
25 -	······		۱			1.75	218	1.7469089	30.9908
	0 2	4	6			1.875	213	1.8647846	31.3082
	ln(t/((t-t1))				2	208	2.0053336	31.7703
						2.125	204	2.1400662	32.2781
						2.25	201	2.2587825	32.6079
						2.375	198	2.3978953	33.0374
						2.5	196	2.5055259	33.4457
						2.625	194	2.6288008	33.6864
						2.75	192	2.7725887	34.0541
						2.875	. 191	2.8543782	34.3404
						3	189	3.0445224	34.8287

020009

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163_25</u>	<u>4</u>	Client ID:	(2/15) Cs Loaded Resor	cinol Resin	Actual Test Ter	nperature:	22 °C	
Slope	Heating <u>3.093</u>	Cooling <u>3.327</u>	Average 3.210	Current: R:	<u>0.2</u> A <u>89.73</u> ohms/m	Heating	On (s) 60	Off (s) 240	Delta 180
Therm Cond	0.0846	0.0787	0.0816	Q: C:	3.5892 W/m <u>0.91659</u> Cal Factor			0.0	
Thermal Con	ductivity, W/	′(m [·] K) =	0.0816						



Beg	End	Delta	
<u>2</u>	<u>5</u>	0.15	
Goal In(t)	Time,s	ln(t)	т, °С
2	7	1.9459101	31.5902
2.15	9	2.1972246	32.2832
2.3	10	2.3025851	32.874
2.45	12	2.4849066	33.1771
2.6	13	2.5649494	33.4883
2.75	16	2.7725887	34.3016
2.9	18	2.8903718	34.6171
3.05	21	3.0445224	35.0915
3.2	25	3.2188758	35.5794
3.35	29	3.3672958	36.026
3.5	33	3.4965076	36.4587
3.65	38	3.6375862	36.9058
3.8	45	3.8066625	37.3683
3.95	52	3.9512437	37.8634
4.1	60	4.0943446	38.2607
4.25	70	4.2484952	38.7864
4.4	81	4.3944492	39.2317
4.55	95	4.5538769	39.6923
4.7	110	4.7004804	40.1327
4.85	128	4.8520303	40.5975
5	148	4.9972123	41.0166

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 25d</u>	Client ID:	<u>(2/15) Cs</u>	Loaded Resorcinol Resin	<u></u>	Actual Test Tempe	erature:	22 °C	
Cooling Cur 43 41 39 37 5 33 31 29 27 25 0	ve y = 3.327x + 26.591 R ² = 0.9994	4 t10)		Cooling Raw Cooling Selected Linear (Cooling Selected)	Range	Beg <u>0.5</u> Goal In(t/(t-t1)) 0.5 0.625 0.75 0.875 1.125 1.125 1.25 1.25 1.375 1.5 1.625 1.75 1.875 2 2 2 125	End <u>3</u> Time,s 457 387 341 309 285 267 252 241 232 224 218 213 208 204	Delta 0.125 In(t/(t-t1)) 0.5006659 0.6257059 0.7504781 0.8735289 0.9985288 1.1213405 1.252763 1.3739231 1.4954937 1.6274564 1.7469089 1.8647846 2.0053336 2.1400662	T, °C 28.372 28.748 29.1027 29.5071 29.8799 30.2904 30.7172 31.1098 31.4702 31.9139 32.3382 32.8009 33.2202 33.7612
				anna ann an an ann an ann ann ann ann a		2.25 2.375	201 198	2.2587825 2.3978953	34.1006 34.599
						2.5	196	2.5055259	35.0505

020011

2.625

2.75

2.875

3

194

192

191

2.6288008

2.7725887

2.8543782

189 3.0445224

35.4534

35.7768

36.0991

36.6705

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	486163 25	<u>t</u>	Client ID:	(2/15) Cs Loaded Reso	rcinol Resin	_	Actual Test Terr	perature:	22 °C	-
Slope Therm Cond	Heating <u>3.0203</u> 0.0867	Cooling <u>3.0655</u> 0.0854	Average 3.0429 0.0860	Current: R: Q: C:	<u>0.2</u> A <u>89.73</u> ohms/m 3.5892 W/m 0.91659 Cal Factor		Heating:	On (s) <u>60</u>	Off (s) <u>240</u>	Delta 180
Thermal Cond	luctivity, W/	/(m [·] K) =	0.0860]						
Heating Curv	e					Range	Beg <u>2</u>	End <u>5</u>	Delta 0.15	



Beg	End	Delta		
2	<u>5</u>	0.15		
Goal In(t)	Time,s	ln(t)	т, °С	
2	7	1.9459101	32.4917	
2.15	9	2.1972246	33.379	
2.3	10	2.3025851	33.5593	
2.45	12	2.4849066	34.0736	
2.6	13	2.5649494	34.3498	
2.75	16	2.7725887	34.9769	
2.9	18	2.8903718	35.3397	
3.05	21	3.0445224	35.8577	
3.2	25	3.2188758	36.3644	
3.35	29	3.3672958	36.7714	
3.5	33	3.4965076	37.2129	
3.65	38	3.6375862	37.6671	
3.8	45	3.8066625	38.1739	
3.95	52	3.9512437	38.5902	
4.1	60	4.0943446	39.0164	
4.25	70	4.2484952	39.5026	
4.4	81	4.3944492	39.9322	
4.55	95	4.5538769	40.4071	
4.7	110	4.7004804	40.8062	
4.85	128	4.8520303	41.2544	
5	148	4.9972123	41.6742	
Thermal Conductivity - ASTM D5334/D5930

ample ID: <u>48</u>	<u>6163_25t</u>	Client ID:	(2/15) Cs	S Loaded Resorcinol Resin		Actual Test Tempe	rature:	22 °C	
ample ID: $\underline{48}$ Cooling Curve 43 $\Psi =$ 41 39 37 35 33 31 29 27 25	6163_25t 3.0655x + 27.45 R ² = 0.9992	Client ID:	(2/15) C:	 Loaded Resorcinol Resin Cooling Raw Cooling Selected Linear (Cooling Selected) 	Range	Actual Test Tempe Beg <u>0.5</u> Goal In(t/(t-t1)) 0.5 0.625 0.75 0.875 1 1 1.125 1.25 1.375 1.5 1.5 1.625	End <u>3</u> Time,s 457 387 341 309 285 267 252 241 232 224 224	22 °C Delta 0.125 In(t/(t-t1)) 0.5006659 0.6257059 0.7504781 0.8735289 0.9985288 1.1213405 1.252763 1.3739231 1.4954937 1.6274564 1.7469089	T, °C 28.9609 29.3533 29.7345 30.0761 30.5048 31.2388 31.6578 31.997 32.4882 32.8554
0	2	4	6			1.875	213	1.8647846	33.23
	in(t/(t-t1))				2	208	2.0053336	33.6492
						2.125	204	2.1400662	34.0928
						2.25	201	2.258/825	34.4267
						2.375	199	2.33/8933	54.6701

020013

2.5

2.625

2.75

2.875

3

196

194

192

191

2.5055259

2.6288008

2.7725887

2.8543782 189 3.0445224 35.2731 35.485

35.8558

36.1389

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163_25q</u>		Client ID:	(2/15) Cs Loaded Reso	Actual Test Te	22 °C			
Slope Therm Cond	Heating <u>3.1238</u> 0.0838	Cooling <u>3.372</u> 0.0776	Average 3.2479 0.0806	Current: R: Q: C:	<u>0.2</u> A <u>89.73</u> ohms/m 3.5892 W/m 0.91659 Cal Factor	Heating:	On (s) <u>60</u>	Off (s) <u>240</u>	Delta 180
Thermal Cond	luctivity, W	/(m [·] K) =	0.0806]					

Range



Beg	End	Delta	
<u>2</u>	<u>5</u>	0.15	
Goal In(t)	Time,s	ln(t)	T, °C
2	7	1.9459101	31.6879
2.15	9	2.1972246	32.6217
2.3	10	2.3025851	32.8287
2.45	12	2.4849066	33.4028
2.6	13	2.5649494	33.7277
2.75	16	2.7725887	34.3835
2.9	18	2.8903718	34.7428
3.05	21	3.0445224	35.2592
3.2	25	3.2188758	35.7707
3.35	29	3.3672958	36.24
3.5	33	3.4965076	36.7137
3.65	38	3.6375862	37.1634
3.8	45	3.8066625	37.6567
3.95	52	3.9512437	38.1325
4.1	60	4.0943446	38.5379
4.25	70	4.2484952	39.0452
4.4	81	4.3944492	39.454
4.55	95	4.5538769	39.9499
4.7	110	4.7004804	40.3192
4.85	128	4.8520303	40.8076
5	148	4.9972123	41.1784

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 25q</u>	Client ID:	<u>(2/15) C</u> :	s Loaded Resorcinol Resin		Actual Test Tempe	erature:	22 °C	
Cooling Cur	νe y = 3.372x + 25.958		1820-y		Range	Beg <u>0.5</u> Goal In(t/(t-t1))	End <u>3</u>	Delta 0.125	T °C
41	R ² = 0.9984					0.5	457	0.5006659	27.3702
39						0.625	387	0.6257059	27.9851
27						0.75	341	0.7504781	28.457
5/	13					0.875	309	0.8735289	28.914
⊢ ³⁵ +						1	285	0.9985288	29.358
33			Hannon			1.125	267	1.1213405	29.7769
31				Cooling Selected		1.25	252	1.252763	30.2342
29				Linear (Cooling Selected)		1.375	241	1.3739231	30.6739
23						1.5	232	1.4954937	31.0427
27						1.625	224	1.6274564	31.5252
25 -		·······				1.75	218	1.7469089	31.8827
0	2	4	6			1.875	213	1.8647846	32.3831
	ln(t/(t-t:	1))				2	208	2.0053336	32.8273
						2.125	204	2.1400662	33.2174
						2.25	201	2.2587825	33.6036
						2.375	198	2.3978953	34.0886
						2.5	196	2.5055259	34.4077
						2.625	194	2.6288008	34.8996

2.75

2.875

3

192

191

2.7725887

2.8543782 189 3.0445224

35.1765 35.5036

Thermal Conductivity - ASTM D5334/D5930



0
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0
0
-

4.2484952

4.3944492

4.5538769

4.7004804

4.8520303

4.9972123

70

81

95l

110

128

148

4.25

4.4

4.7

4.85

5

4.55

42.8344

43.2785

43.6682

44.0134

44.4924

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 45</u>	Client ID:	(2/15) Cs	Loaded Resorcinol Resin		Actual Test Tempe	erature:	45 °C	
Cooling Curv 50 45	νe <u>y = 3.1927x + 31.127</u> R ² = 0.9975				Range	Beg <u>0.5</u> Goal In(t/(t-t1)) 0.5 0.625 0.75	End <u>3</u> Time,s 457 387 341	Delta 0.125 In(t/(t-t1)) 0.5006659 0.6257059 0.7504781	T, °C 32.997 33.2917 33.5878
40			Watana and a second			0.875	309 	0.8735289	33.8904 34.2848
⊢	and the second sec			Cooling Raw		1.125	267	1.1213405	34.7076
35	······································			Cooling Selected		1.25	252	1.252763	35.0408
				—— Linear (Cooling Selected)		1.375	241	1.3739231	35.3531
30	*****	*****	******			1.5	232	1.4954937	35.7624
						1.625	224	1.6274564	36.2172
25						1.75	218	1.7469089	36.6053
0	2	4	6			1.875	213	1.8647846	36.9968
	ln(t/(t-t:	1))				2	208	2.0053336	37.4601
L						2.125	204	2.1400662	37.9716
						2.25	201	2.2587825	38.3079
						2.375	198	2.3978953	38.7431
						2.5	196	2.5055259	39.0827
						2.625	194	2.6288008	39.4513
						2./5	192	2.7725887	40.0984
						2.8/5	191	2.8543782	40.2794
						3	189	3.0445224	41.146

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 450</u>	ł	Client ID:	<u>(2/15) C</u>	s Loaded Reso	rcinol Resin		Actual Test Temp	erature:	45 °C	
Slope: Therm Cond	Heating <u>2.839</u> 0.0922 ductivity, W/	Cooling <u>2.9849</u> 0.0877 ((m [·] K) =	Average 2.91195 0.0899 0.0899	1	Current R Q C	<u>0.2</u> A <u>89.73</u> ohms, 3.5892 W/m <u>0.91659</u> Cal Fa	/m ctor	Heating:	On (s) <u>50</u>	Off (s) <u>240</u>	Delta 190
Heating Curv	e						Range	Beg	End <u>5</u>	Delta 0.15	
50								Goal In(t)	Time	ln(t)	т, °С
		y = 2.83	9x + 31.9					2	7	1.9459101	37.5256
45		R ² = (0.9997					2.15	9	2.1972246	38.1181
			1 states					2.3	10	2.3025851	38.4662
40								2.45	12	2.4849066	38.9353
	100	and the second s				Raw		2.6	13	2.5649494	39.1966
25								2.75	16	2.7725887	39.7901
35		2			A Heating	Selected		2.9	18	2.8903718	40.0039
					—— Linear (H	Heating Selected)		3.05	21	3.0445224	40.4962
30								3.2	25	3.2188758	41.038
								3.35	29	3.3672958	41.4315
25			·····	······				3.5	33	3.4965076	41.8421
0	2		4	6				3.65	38	3.6375862	42.2599
		ln(t)						3.8	45	3.8066625	42.6783
								3.95	52	3.9512437	43.0674
								4.1	60	4.0943446	43.4748
								4.25	70	4.2484952	43.954

4.4

4.55

4.7

4.85

5

81

95

4.3944492

4.5538769

110 4.7004804

128 4.8520303

148 4.9972123

44.4621

44.8861

45.2409

45.6622

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 45d</u>	Client ID:	(2/15) Cs	Loaded Resorcinol Resin		Actual Test Tempe	rature:	45 °C	
Cooling Cur	rve y = 2.9849x + 32.323 $P^2 = 0.0004$				Range	Beg <u>0.6</u> Goal In(t/(t-t1))	End <u>3.1</u> Time,s	Delta 0.125 In(t/(t-t1))	T, °C
45 —	R ⁻ = 0.9994	1				0.6 0.725	421 368	0.6002151 0.7262994	34.1379 34.4853
	and the second se					0.85 0.975	332 305	0.8493079 0.9753796	<u>34.9108</u> 35.1504
40 −	and the second s	in States		Cooling Raw		1.1	285	1.0986123	35.5756
35 —	and the second s			Cooling Selected		1.35	205	1.3555227	36.3701
				—— Linear (Cooling Selected)		1.475	246	1.4799798	36.6699
30 —				, , , , , , , , , , , , , , , , , , , ,		1.6	238	1.6010697	37.0559
						1.725	231	1.7288456	37.4363
25 -			······································			1.85	225	1.8607523	37.9061
0	2	4	6			1.975	221	1.9641755	38.1932
	ln(t/(t·	-t1))				2.1	217	2.0840605	38.595
<u> </u>						2.225	213	2.2257979	39.0287
						2.35	210	2.3513/53	39.4248
						2.4/5	207	2.4995054	39.808
						2.6	205	2.0149598	40.179
						2./25	203	2.7482566	40.5191
						2.85	202	2.823361	40.7928
						2.9/5	200	2.995/323	41.2404
						5.1	199	3.0900602	41.4104

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 451</u>	<u>t</u>	Client ID:	(2/15) Cs Loaded Resor	cinol Resin	_	Actual Test Tem	perature:	45 °C	-
Slope: Therm Cond Thermal Cond	Heating <u>2.894</u> 0.0905 ductivity, W/	Cooling <u>3.0993</u> 0.0845	Average 2.99665 0.0874 0.0874	Current: R: Q: C:	<u>0.2</u> A <u>89.73</u> ohms/m 3.5892 W/m <u>0.91659</u> Cal Factor		Heating:	On (s) <u>60</u>	Off (s) <u>240</u>	Delta 180
Heating Curv	'e			•		Range	Beg <u>2</u>	End 5	Delta 0.15	



Beg	End	Delta	
<u> </u>	<u>5</u>	0.15	
Goal In(t)	Time,s	ln(t)	T, °C
2	7	1.9459101	37.6623
2.15	9	2.1972246	38.2949
2.3	10	2.3025851	38.6411
2.45	12	2.4849066	39.1166
2.6	13	2.5649494	39.2595
2.75	16	2.7725887	39.9049
2.9	18	2.8903718	40.3745
3.05	21	3.0445224	40.8264
3.2	25	3.2188758	41.3009
3.35	29	3.3672958	41.7117
3.5	33	3.4965076	42.0649
3.65	38	3.6375862	42.5115
3.8	45	3.8066625	42.968
3.95	52	3.9512437	43.389
4.1	60	4.0943446	43.8244
4.25	70	4.2484952	44.2616
4.4	81	4.3944492	44.7205
4.55	95	4.5538769	45.1454
4.7	110	4.7004804	45.5218
4.85	128	4.8520303	45.9348
5	148	4.9972123	46.4469

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 45t</u>	Client ID:	(2/15) Cs	Loaded Resorcinol Resin		Actual Test Tempe	45 °C		
Cooling Cur	ve				Range	Beg <u>0.5</u>	End <u>3</u>	Delta 0.125	
50 -	y = 3.0993x + 32.563 $R^2 = 0.9992$					Goal In(t/(t-t1))	Time,s	ln(t/(t-t1))	т, °С
	N = 0.5552					0.5	457	0.5006659	34.2172
45						0.625	387	0.6257059	34.5369
	and the second					0.75	341	0.7504781	34.9266
40	and the second se					0.875	309	0.8735289	35.1855
⊢	and the second s			Cooling Raw		1	285	0.9985288	35.6335
35				Cooling Selected		1.125	267	1.1213405	36.0145
	2 contraction of the second se					1.25	252	1.252763	36.3719
20				Linear (Cooling Selected)		1.3/5	241	1.3739231	36.7084
30						1.5	232	1.4954937	37.1484
						1.625	224	1.6274564	37.5763
25 +			1			1./5	218	1.7469089	38.0222
0	2	4	6			1.8/5	213	1.864/846	38.3603
	In(t/(t-	·t1))				2 125	208	2.0053336	38.7921
L						2.125	204	2.1400662	39.2711
						2.25	201	2.258/825	39.6606
						2.3/5	100	2.39/8953	40.0944
						2.5	104	2.5055259	40.2981
						2.025	194	2.0288008	40.661
						2./5		2.7725887	41.2282
						2.8/5	191	2.8543/82	41.3//3
						3	183	5.0445224	41.9069

Thermal Conductivity - ASTM D5334/D5930



45.3194

45.6608

45.9283

46.3014

46.743

4.4

4.55

4.7

5

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81

95

110

128

148

4.3944492 4.5538769

4.7004804

4.8520303

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163_90</u>	Client ID:	(2/15) Cs	Loaded Resorcinol Resin		Actual Test Tempe	rature:	90 °C	
Cooling Cu	rve				Range	Beg <u>0.5</u>	End <u>3</u>	Delta 0.125	- 9-
50 -	$R^2 = 0.9981$	and the strick under setting	() () () () () () () () () () () () () (Goal In(t/(t-t1))	Time,s	ln(t/(t-t1))	T, °C
		-				0.5	457	0.5006559	35./10/
45	10010010002-000100000000000000000000000					0.025	307	0.0257055	26 25 29
	and the second					0.75	309	0.7304781	36 4318
40	1 Alexandre					1	285	0.0735205	36 813
⊢	Alerent			Cooling Raw		1.125	267	1.1213405	37.1123
35				Cooling Selected		1.25	252	1.252763	37.3673
				—— Linear (Cooling Selected)		1.375	241	1.3739231	37.6596
30 -						1.5	232	1.4954937	37.9936
						1.625	224	1.6274564	38.4086
25						1.75	218	1.7469089	38.6783
0	2	4	6			1.875	213	1.8647846	39.0718
	ln(t/(t-t1))				2	208	2.0053336	39.4678
L						2.125	204	2.1400662	39.7834
						2.25	201	2.2587825	40.0225
						2.375	198	2.3978953	40.4573
						2.5	196	2.5055259	40.821
						2.625	194	2.6288008	41.1265
						2.75	192	2.7725887	41.5045
						2.875	191	2.8543782	41.7824
						3	189	3.0445224	42.2427

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 90</u>	<u>d</u>	Client ID:	(2/15) Cs Loaded Reso	rcinol Resin	Actual Test Te	mperature:	90 °C	-
Slope Therm Cond	Heating <u>2.4622</u> 0.1063	Cooling <u>2.6639</u> 0.0983	Average 2.56305 0.1021	Current: R: Q:	<u>0.2</u> A <u>89.73</u> ohms/m 3.5892 W/m	Heating:	On (s) <u>60</u>	Off (s) <u>300</u>	Delta 240
Thermal Con	ductivity, W	/(m [·] K) =	0.1021]	0.91659 Cal Factor				



Beg	End	Delta	
<u>2</u>	<u>5</u>	0.15	
Goal In(t)	Time,s	ln(t)	Temp
2	7	1.9459101	39.61
2.15	9	2.1972246	39.9867
2.3	10	2.3025851	40.3269
2.45	12	2.4849066	40.7643
2.6	13	2.5649494	41.018
2.75	16	2.7725887	41.5841
2.9	18	2.8903718	41.8475
3.05	21	3.0445224	42.246
3.2	25	3.2188758	42.6417
3.35	29	3.3672958	43.0554
3.5	33	3.4965076	43.3265
3.65	38	3.6375862	43.6903
3.8	45	3.8066625	44.1504
3.95	52	3.9512437	44.5076
4.1	60	4.0943446	44.8402
4.25	70	4.2484952	45.2601
4.4	81	4.3944492	45.5314
4.55	95	4.5538769	45.908
4.7	110	4.7004804	46.2485
4.85	128	4.8520303	46.6634
5	148	4.9972123	46.9009

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 90d</u>	Client ID:	(2/15) Cs	Loaded Resorcinol Resin		Actual Test Tempe	erature:	90 °C	,
Cooling Curr	ve				Range	Beg <u>0.6</u>	End <u>3.1</u>	Delta 0.125	
50 -	y = 2.6639x + 34.417					Goal In(t/(t-t1))	Time,s	in(t/(t-t1))	Temp
	R ² = 0.9971	-				0.6	532	0.5998897	36.206
45 -						0.725	465	0.725937	36.4817
	. 44	Jack Contraction of the second				0.85	419	0.8504851	36.758
40		*				0.975	385	0.9765096	37.1336
	and the second sec			Cooling Pour		1.1	360	1.0986123	37.3172
	and the second second					1.225	340	1.2237754	37.62
35				Cooling Selected		1.35	324	1.3499267	37.9733
				Linear (Cooling Selected)		1.475	311	1.477113	38.3041
30 +						1.6	301	1.5962364	38.5295
						1.725	292	1.7255101	38.8666
25 -)			1.85	285	1.8458267	39.1907
0	2	4	6			1.975	279	1.9676501	39.493
	ln(t/((t-t1))				2.1	273	2.1129642	39.9454
						2.225	269	2.2274155	40.3195
						2.35	265	2.360854	40.6907
						2.475	262	2.4773021	40.9634
						2.6	259	2.6123891	41.3915
						2.725	257	2.7158627	41.655
						2.85	255	2.8332133	42.0848
						2.975	253	2.9684401	42.4835

3.1

251

3.1275577

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163_90</u>	t	Client ID:	(2/15) Cs Loaded Reso	rcinol Resin	Actual Test Te	mperature:	<u>90</u> °C	-
Slope Therm Cond	Heating <u>2.4973</u> 0.1048	Cooling <u>2.6191</u> 0.1000	Average 2.5582 0.1023	Current: R: Q: C:	<u>0.2</u> A <u>89.73</u> ohms/m 3.5892 W/m 0.91659 Cal Factor	Heating:	On (s) <u>60</u>	Off (s) <u>240</u>	Delta 180
Thermal Cond	ductivity, W	/(m [·] K) =	0.1023]					
						_			

Range



Beg	End	Delta	
<u> </u>	<u>5</u>	0.15	
Goal In(t)	Time,s	ln(t)	T, °C
2	7	1.9459101	39.3838
2.15	9	2.1972246	40.1309
2.3	10	2.3025851	40.2836
2.45	12	2.4849066	40.7101
2.6	13	2.5649494	40.9291
2.75	16	2.7725887	41.3744
2.9	18	2.8903718	41.6915
3.05	21	3.0445224	42.0883
3.2	25	3.2188758	42.5941
3.35	29	3.3672958	42.936
3.5	33	3.4965076	43.2455
3.65	38	3.6375862	43.653
3.8	45	3.8066625	44.1458
3.95	52	3.9512437	44.4758
4.1	60	4.0943446	44.8457
4.25	70	4.2484952	45.2434
4.4	81	4.3944492	45.5347
4.55	95	4.5538769	45.8832
4.7	110	4.7004804	46.2591
4.85	128	4.8520303	46.6305
5	148	4.9972123	46.8933

Thermal Conductivity - ASTM D5334/D5930

Sample ID:	<u>486163 90t</u>	Client ID: (2/15) Cs Loaded Resorcinol Resin				Actual Test Temperature:		90 °C	
Cooling Cu	rve				Range	Beg <u>0.5</u>	End <u>3</u>	Delta 0.125	
50 -	y = 2.6191x + 34.527					Goal In(t/(t-t1))	Time,s	ln(t/(t-t1))	т, °С
	R ² = 0.9977	-				0.5	457	0.5006659	35.9959
45						0.625	387	0.6257059	36.3156
	-					0.75	341	0.7504781	36.5121
40						0.875	309	0.8735289	36.9053
	and the second second			Cooling Bow		1	285	0.9985288	37.0861
	1 and the second					1.125	267	1.1213405	37.4441
35				Cooling Selected		1.25	252	1.252763	37.821
				Linear (Cooling Selected)		1.375	241	1.3739231	38.0886
30						1.5	232	1.4954937	38.3452
						1.625	224	1.6274564	38.6849
25 -						1.75	218	1.7469089	38.9545
0	2	4	6			1.875	213	1.8647846	39.2823
	ln(t/((t-t1))		_		2	208	2.0053336	39.6948
						2.125	204	2.1400662	40.0461
						2.25	201	2.2587825	40.3873
						2.375	198	2.3978953	40.7803
						2.5	196	2.5055259	41.0603
						2.625	194	2.6288008	41.4838

2.7725887

2.8543782

189 3.0445224

192

191

2.75

2.875

3

41.8673

42.1581

Thermal Conductivity - ASTM D5334/D5930

Sample ID: Glycerine CCV 031712

	Heating	Cooling	Average
Slope	<u>0.9391</u>	<u>0.8567</u>	0.8979
Therm Cond	0.279	0.306	0.292
TV	<u>0.292</u>	0.292	0.292
Recovery	95.5%	105%	100%

Current: R: Q: C:

<u>0.2</u> A <u>89.73</u> ohms/m 3.5892 W/m <u>0.91659</u> Cal Factor

Range

 Heating:
 On (s)
 Off (s)
 Delta

 60
 240

180

45				
43	y = 0.1	3391x + 37.782	-	
41	H مفیشیش	-= 0.99 <i>/1</i>		
39	the second secon	44-2	10 1 10 10 10	
37				
35		2013-1100)		Heating Raw
33				👗 Heating Selected
31			-	—— Linear (Heating Selected)
29				,
27				
25		Y		
	2	1	6	

Beg	End	Delta	
<u>2</u>	<u>5</u>	0.15	
Goal In(t)	Time,s	ln(t)	т, °С
2	7	1.9459101	39.5172
2.15	9	2.1972246	39.7857
2.3	10	2.3025851	39.9414
2.45	12	2.4849066	40.1418
2.6	13	2.5649494	40.1828
2.75	16	2.7725887	40.4147
2.9	18	2.8903718	40.5348
3.05	21	3.0445224	40.6904
3.2	25	3.2188758	40.8789
3.35	29	3.3672958	40.9924
3.5	33	3.4965076	41.1098
3.65	38	3.6375862	41.2103
3.8	45	3.8066625	41.3484
3.95	52	3.9512437	41.4489
4.1	60	4.0943446	41.5478
4.25	70	4.2484952	41.7409
4.4	81	4.3944492	41.9201
4.55	95	4.5538769	42.0884
4.7	110	4.7004804	42.1953
4.85	128	4.8520303	42.2632
5	148	4.9972123	42.5018

Thermal Conductivity - ASTM D5334/D5930

Sample ID: Glycerine CCV 031712



Beg	End	Delta	
<u>0.5</u>	<u>3</u>	0.125	
Goal In(t/(t-t1))	Time,s	ln(t/(t-t1))	T, °C
0.5	457	0.5006659	36.25
0.625	387	0.6257059	36.3099
0.75	341	0.7504781	36.4167
0.875	309	0.8735289	36.4653
1	285	0.9985288	36.6429
1.125	267	1.1213405	36.713
1.25	252	1.252763	36.7726
1.375	241	1.3739231	36.8684
1.5	232	1.4954937	37.0073
1.625	224	1.6274564	37.1305
1.75	218	1.7469089	37.2739
1.875	213	1.8647846	37.4128
2	208	2.0053336	37.4699
2.125	204	2.1400662	37.5883
2.25	201	2.2587825	37.7077
2.375	198	2.3978953	37.8298
2.5	196	2.5055259	37.8924
2.625	194	2.6288008	37.9806
2.75	192	2.7725887	38.1092
2.875	191	2.8543782	38.2185
3	189	3.0445224	38.4053

SOUTHWEST RESEARCH INSTITUTE CLIENT: Battelle Memorial Ins. PNNL TASK ORDER#: 120319-5 SRR#: 46112 SDG#: 486163 VTSR: 120319-5 PROJECT #: 13295.12.008

Specific Heat Capacity Data

<< DSC	>>		Tei	Temperature Program:				
Data Nam	e:120316-486163	-ср			[C]	[C/min]	[min]	[sec]
Date:	12/ 3/16 17:55		1*	25 -	25	10	4	0.2
Sample:	486163		2*	25 -	100	20	5	0.2
	9.68	mg						

Comments: Operator rss oval sensor, al open

Reference:

0 mg



020031

SwRI



<< DSC	>>	Temperature Program:					
Data Name:120316-sap483-Cp v				[C]	[C/min]	[min]	[sec]
Date:	12/3/16 16:33	1*	25 -	25	10	4	0.2
Sample:	PE sapphire 483	2*	25 -	100	20	5	0.2
	9.04 mg						

Comments: Operator rss oval sensor, al open

Reference:

0 mg



Perkin Elmer #0219 1483 Sapphire Disk Verification against NIST SRM 720

SwRI



















SOUTHWEST RESEARCH INSTITUTE CLIENT: Battelle Memorial Ins. PNNL TASK ORDER#: 120319-5 SRR#: 46112 SDG#: 486163 VTSR: 120319-5 PROJECT #: 13295.12.008

TGA/DSC-FTIR Data

020044rev1 145.22.12



Figure 1. Weight loss curve with heat flow for Cs-loaded resorcinol resin.





Figure 2. Weight loss curve and the derivative curve for Cs-loaded resorcinol resin.

020045Arev1 195.2212



Figure 3. Weight loss curve with the derivative curve and heat flow curve included.



Figure 4. Gram-Schmidt Reconstruction for TGA/DSC analyis of the Cs-loaded resorcinol resin (see Figure 1)



Figure 5. Initial FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin.

020045Crev1 145.22.12




Figure 6. FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin at 2.95-3.89 min.



Figure 7. FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin at 16.50-19.80 min



Figure 8. FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin at 27.09-28.03 min.



Figure 9. FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin at 37.02-40.10 min.

020045GRV | A.5.22.12



Figure 10. FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin at 40.77-43.86 min.



Figure 11. FTIR spectrum collected from TGA/DSC analyis of the Cs-loaded resorcinol resin at 44.13-46.14 min.

SOUTHWEST RESEARCH INSTITUTE CLIENT: Battelle Memorial Ins. PNNL TASK ORDER#: 120319-5 SRR#: 46112 SDG#: 486163 VTSR: 120319-5 PROJECT #: 13295.12.008

TGA/DSC-FTIR in N₂



File: Nuclear Sample 486163 (in N2 with cer...

020047

Overlay of Sample 486163 run with air purge (TGA/FTIR) and run with nitrogen purge



SOUTHWEST RESEARCH INSTITUTE CLIENT: Battelle Memorial Ins. PNNL TASK ORDER#: 120319-5 SRR#: 46112 SDG#: 486163 VTSR: 120319-5 PROJECT #: 13295.12.008

Logbook Copies

TITLE BATTELLE RESORCINAL SWAL PROJECT NO. 13795, 17.008 91 RESIN (CESSIUM LOADSNG) BOOK NO. 04-0406-036 Work continued from Page REAGENTS: IM NOOH - 40. Og OF NOOH WHS ADDE (INDEG 3949) WAS ADDED TO LOL OF D. I. H20 ADD DESSOLUED. TOLUTION WAS ALIONED TO COOL PRIOR TO USE F.U. OF 1.01 OBTALLED. 02/13/12 1 MNa OH - 40. 02 OF NaOH (INORG 9603) WAS ADDED To 1.06 OF D.I. H20 AND DESSAVED Saures WAS ALLOWED TO COOL PREDER TO CISE A F.V. OF X.OL OBTAS ED. 111 22/13/12 O. 5M HAD - 32. ONL OF HAD (INORG 9651) WAS ADDED To A F.V. OF 1000 al WITH D.I. 4.0. 12/13/12 O. OIM GARO, /IN NOH - 1.95% CONO3 (INORG 1003) WAS ADDED To 900 mL OF /M Na OH (40,04 NaOH (INDRG 965) NER I.OL D. I. H.D) AND DESERVED. BROUGHT TO A F.V. OF 1000 AL USTALL SAME IM NOOH 102/13/12 PROCEDURE: RESTA 445 ALTQUOTED INTO Y EQUAL PARTS OF 250 ML BED VOLUMES TO VEELD A TOTAL BED VOLUME 1000 ml. EACH ALIQUOT WAS TREATED THE SAME AND USED THE SAME REAGENTS, AS DESCRIBED IN THE FOROWIANG METHOD (SEE ATTACHED) : TABLE 1. Balance # 19 ess 3/13/12 www.scientificbindery88yrs.com Work continued to Page SIGNATURE 07/13/12 DATE 3/13/12 DISCLOSED TO AND UNDERSTOOD BY DATE WITNESS



Appendix B

Cesium Loading Procedure

Spherical Resorcinol Formaldehyde (SRF) Resin Preparations for Loading with Cesium

Caution: It is important that the SRF resin not be allowed to dry. There is no potential for a safety hazard should this occur; however, the ion exchange performance of the SRF resin will be severely reduced and inaccurate data generated.

1. Determine the amount of SRF resin needed for experimental testing. Fill container, no more than $1/5^{th}$ full, with SRF resin. Use volume measurement for performing each step.

Note: Given the relatively large volume of resin needed for testing, use a Büchner funnel to facilitate solution-resin contact.

- 2. The dry bed density of the SRF resin is approximately 0.456 g/mL. If the resin will be stored for a long period, bubble inert gas (e.g., Ar or N_2) over the resin for >30 min and seal the container.
- 3. Complete the "Bulk Pretreatment" portion of Table 1 (e.g., water rinses, resin expansion, and conversion) as follows:
 - 3.1 After adding the required solution volumes into the container, gently swirl by hand every 10 min during the specified duration. Alternatively, the analyst may choose to allow solution contact with the resin in a Büchner funnel.
 - 3.2 Decant the solutions to a waste container and retain the resin in the beaker. Alternatively, the analyst may choose to allow solution contact with the resin in a Büchner funnel, pulling the solution slowly through the resin with vacuum. Use care not to allow the resin to fully dry between solution contacts.
 - 3.3 Record each of the start/stop times, volumes, and solutions used in the space provided in Table 2.
 - 3.4 After completing the bulk pretreatment steps, transfer the resin to the testing vessel.

Process/Pretreatment Step	Solution	Volume	Time	Mixing	Flowrate
Bulk Pretreatment				<u> </u>	
Water Rinse	DI Water	$3 \text{ RV}^{(a)}$	30 min	Swirl ^(b)	NA ^(c)
Resin Expansion – 1 st	1 <u>M</u> NaOH	3 RV	30 min	Swirl	NA
Resin Expansion -2^{nd}	1 <u>M</u> NaOH	3 RV	>4 h	Soak	NA
Water Rinse -1^{st}	DI Water	3RV	30 min	Swirl	NA
Water Rinse -2^{nd}	DI Water	3RV	30 min	Swirl	NA
Water Rinse -3^{rd}	DI Water	3RV	30 min	Swirl	NA
Resin Conversion -1^{st}	0.5 <u>M</u> HNO ₃	3 RV	30 min	Swirl	NA
Resin Conversion -2^{nd}	0.5 <u>M</u> HNO ₃	3 RV	30 min	Swirl	NA
Resin Conversion -3^{rd}	0.5 <u>M</u> HNO ₃	3 RV	30 min	Swirl	NA
Water Rinse – 4 th	DI Water	3 RV	1 min	Swirl	NA
Resin Expansion	1 <u>M</u> NaOH	3 RV	1 h	Swirl	NA
Cesium Loading ^(d)	+0.01 <u>M</u> CsNO ₃	3 RV	1 h	Swirl	NA
Water Rinse – 5 th	DI Water	10 RV	1 min	Swirl	NA

 Table 1. Ion Exchanger Pretreatment and Process Steps

(a) Resin volume (RV).

(b) Gently swirling by hand every 10 min.

(c) Not applicable (NA). Alternatively, the analyst may choose to use a Büchner funnel and flow the solutions slowly through the resin bed at less than 6 RV/hr.

(d) Cesium loading should occur using a solution of 1 <u>M</u> NaOH plus 0.01 <u>M</u> CsNO₃.

Note: Table 2 provides the analyst with the option of preconditioning up to five resin samples all at the same time or up to two different times. If additional preconditioning is needed, the analyst may photocopy the table for additional cycles or record the data directly into the LRB.

Bulk Pretreatment	Solution Used	Samples Processed? $1^{st} 2^{nd} 3^{rd} 4^{th} 5^{th}$	Volume (mL)	Start Time	Stop Time	Step Duration
Water Rinse	DI Water					
Resin Expansion -1^{st}	1 <u>M</u> NaOH					
Resin Expansion -2^{nd}	1 <u>M</u> NaOH					
Water Rinse – 1 st	DI Water					
Water Rinse -2^{nd}	DI Water					
Water Rinse – 3 rd	DI Water					
Resin Conversion -1^{st}	0.5 <u>M</u> HNO ₃					
Resin Conversion -2^{nd}	0.5 <u>M</u> HNO ₃					
Resin Conversion -3^{rd}	0.5 <u>M</u> HNO ₃					
Water Rinse – 4 th	DI Water					
Resin Expansion	1 <u>M</u> NaOH					
Cesium Loading	+0.01 <u>M</u> CsNO ₃					
Water Rinse – 5 th	DI Water					
Bulk Pretreatment	Solution Used	Samples Processed? $1^{st} 2^{nd} 3^{rd} 4^{th} 5^{th}$	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse	Solution Used DI Water	Samples Processed? $1^{st} 2^{nd} 3^{rd} 4^{th} 5^{th}$	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion – 1 st	Solution Used DI Water 1 <u>M</u> NaOH	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion – 1 st Resin Expansion – 2 nd	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk PretreatmentWater RinseResin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -2^{nd}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -2^{nd} Water Rinse -3^{rd}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water DI Water DI Water	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -2^{nd} Water Rinse -3^{rd} Resin Conversion -1^{st}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water DI Water 0.5 <u>M</u> HNO ₃	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -3^{rd} Water Rinse -3^{rd} Resin Conversion -1^{st} Resin Conversion -2^{nd}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water DI Water 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -2^{nd} Water Rinse -3^{rd} Resin Conversion -1^{st} Resin Conversion -2^{nd} Resin Conversion -3^{rd}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water DI Water 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -2^{nd} Water Rinse -3^{rd} Resin Conversion -1^{st} Resin Conversion -2^{nd} Resin Conversion -3^{rd} Water Rinse -4^{th}	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water DI Water 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃ DI Water	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -2^{nd} Water Rinse -3^{rd} Resin Conversion -1^{st} Resin Conversion -2^{nd} Resin Conversion -3^{rd} Water Rinse -4^{th} Resin Expansion	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃ DI Water 1 <u>M</u> NaOH	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration
Bulk Pretreatment Water Rinse Resin Expansion -1^{st} Resin Expansion -2^{nd} Water Rinse -1^{st} Water Rinse -3^{rd} Water Rinse -3^{rd} Resin Conversion -1^{st} Resin Conversion -2^{nd} Resin Conversion -3^{rd} Water Rinse -4^{th} Resin Expansion Cesium Loading	Solution Used DI Water 1 <u>M</u> NaOH 1 <u>M</u> NaOH DI Water DI Water 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃ 0.5 <u>M</u> HNO ₃ DI Water 1 <u>M</u> NaOH +0.01 <u>M</u> CsNO ₃	Samples Processed? 1 st 2 nd 3 rd 4 th 5 th	Volume (mL)	Start Time	Stop Time	Step Duration

 Table 2. Example Data Sheet for Bulk Ion Exchanger Pretreatment

Observations:

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