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Desorption Behavior of Trichloroethene and Tetrachloroethene in U.S. Department of Energy Savannah River Site Unconfined Aquifer Sediments

R. G. Riley
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June 2006

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Summary

U.S. Department of Energy's (DOE) Savannah River Site (SRS) is evaluating the potential applicability of monitored natural attenuation (MNA) to the restoration of the unconfined groundwater aquifer known to be contaminated with the chlorinated hydrocarbon compounds trichloroethene (TCE) and tetrachloroethene (PCE). This report discusses the results from aqueous desorption experiments on SRS aquifer sediments from two different locations at the SRS (A/M Area; P-Area) with the objective of providing technically defensible TCE/PCE distribution coefficient (K_d) data and data on TCE/PCE reversible and irreversible sorption behavior needed for further MNA evaluation.

This study produced the first values of TCE and PCE desorption K_d s based on measurements of TCE and PCE retardation factors in field-contaminated SRS aquifer sediments. TCE and PCE had been in contact with these sediments for decades. Such contact times are impractical to reproduce in laboratory experiments. K_d values of TCE and PCE were 1.95 ± 0.01 L/kg ($n=2$) and 5.24 ± 0.13 L/kg ($n=2$), respectively, for sediments from the A/M Area site. A K_d value of 0.12 ± 0.07 L/kg ($n=2$) was calculated for PCE in sediments from the P-Area site. PCE K_d values differed significantly between the two sites; the PCE K_d value at the A/M area was more than a factor of fifteen higher than that calculated at the P-Area site. The reason for these differences is unknown at this time but the data, while limited, suggest the need to consider that sorption capacity may play an important role in mass balance at this site and as a result, the A/M Area may take longer to remediate. Estimated porewater concentrations of TCE and PCE in A/M Area intact core sediments, based on the high values of K_d , compared reasonably well with concentrations of TCE and PCE measured in samples of groundwater from a well proximate to the site of sediment collection. This helps to validate the higher than expected values of K_d determined for TCE and PCE from the desorption experiments.

TCE and PCE K_d values estimated from sediment organic carbon content and organic carbon distribution coefficients (K_{oc}) or surface area and octanol-water distribution coefficients (K_{ow}) tended to underestimate those that were derived from experimental measurements of retardation factors from this study. This suggests (1) the importance of site-specific experimentally derived data (as opposed to data derived from estimation methods) as a basis for assessing site recovery potential and (2) the need for site-specific data for application in software tools and numerical models addressing natural attenuation-enhanced attenuation issues at sites of groundwater contamination.

Essentially all of the TCE and PCE mass (98 to 100%) was found to be present in the effluent from the desorption experiments, providing evidence for the absence of significant irreversible sorption of TCE or PCE in the form of migration resistant fractions (MRF) in the sediments studied. Numerical modeling of TCE and PCE desorption profiles affirmed this behavior for sediments from the A/M Area site. Break-through curve tailing was observed in desorption profiles for PCE in P-Area sediments, suggesting that a portion of the PCE mass was exhibiting slow release behavior (irreversible sorption). However, the amount of PCE that exhibited migration-resistant behavior was relatively insignificant.

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1.0 Introduction

The organic contaminants trichloroethene (TCE) and tetrachloroethene (PCE) have been detected in groundwater under two locations (A/M Area and P-Area) of the U.S. Department of Energy's (DOE) Savannah River Site (SRS) located in Aiken, Barnwell, and Allendale counties in South Carolina (WSRC 1994). Possible sources of the contamination are the 704 maintenance shop and/or 105 reactor building facility located in the P-Area (Millings et al. 2003). Groundwater and unconfined aquifer sediment underlying the A/M Area are highly contaminated with TCE and PCE. Unlike P-Area, the A/M Area was the site of fabrication of reactor fuel and target assemblies for SRS reactors that resulted in large quantities of TCE and PCE being released to a process sewer system with subsequent release at the A-014 outfall. It has been estimated that between 1952 and 1979, 1,395,000 pounds of solvent mixture were released to the outfall consisting of 72% PCE and 27% TCE along with a small quantity of 1,1,1-trichloroethane (Jackson 1995; Jackson et al. 1999). Over time, contaminants migrated through the vadose zone and contaminated the unconfined groundwater aquifer.

In this report, we discuss the results of desorption experiments performed on intact cores of sediments obtained from the SRS unconfined aquifer. The nature of these sediment samples (i.e., intact cores and long contaminant residence time) afforded an opportunity to (1) obtain and compare K_d values for TCE and PCE to those obtained by means of estimation, (2) assess the effects of long contact time on TCE and PCE accumulation and behavior in these sediments, and (3) assess the implications of the results to the prediction of TCE and PCE transport and fate in SRS groundwater.

2.0 Sampling Methods

Four intact sediment cores were collected from the unconfined aquifer underlying the P- and A/M Areas (Figure 1). Two sediment cores (PGRS-1; 19.05 to 19.2 m [62.5 to 63 ft]; 19.2 to 19.35 m [63.0 to 63.5 ft]) were collected at a location under P-Area (Figure 2), while the other two cores (MRS-41; 42.21 to 42.36 m [138.5 to 139 ft]; 42.36 to 42.51 m [139 to 139.5 ft]) were collected from a location near the A-014 outfall within the A/M Area of SRS (Figure 3) using roto sonic drilling (Vangelas 2001).

Intact sediment cores were contained in stainless steel liners (0.6 cm [0.25 in.] thick by 10.2 cm [4 in.] outside diameter by 15.2 cm [6 in.] in length). Each intact core was sealed and placed in a chest and transported by overnight mail from the SRS to Pacific Northwest National Laboratory. When not subject to experimentation, sediment samples were stored at 4°C. To minimize concerns for sample stability (i.e., potential solute degradation), column desorption experiments were initiated within a week of receipt at the laboratory. This time frame was well within the findings of a previous study that showed chlorinated hydrocarbon concentrations in soils to be unaffected when stored up to 28 days at 4°C (Hewitt 1997).

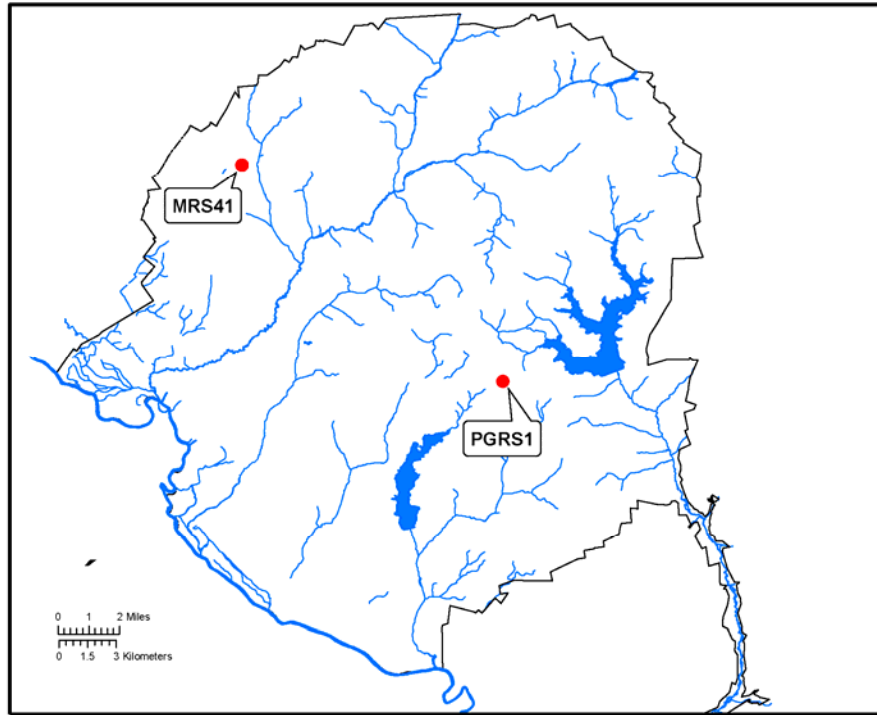


Figure 1. Location of Boreholes on the Savannah River Site where Intact Sediment Cores were Collected for Laboratory Study

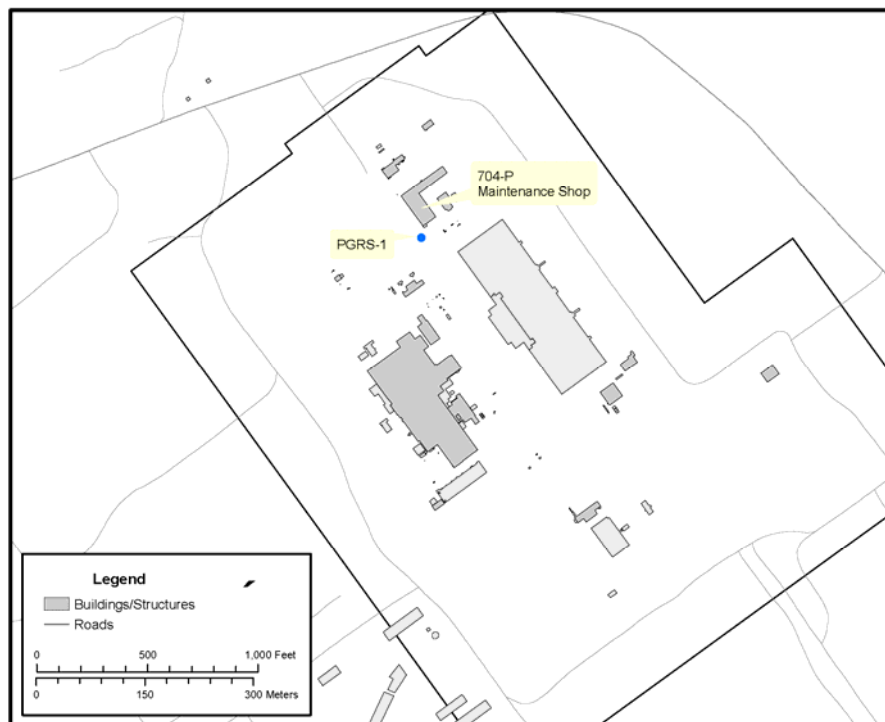


Figure 2. Location of Borehole in the P-Area Where intact Sediment Cores were Collected for Laboratory Study

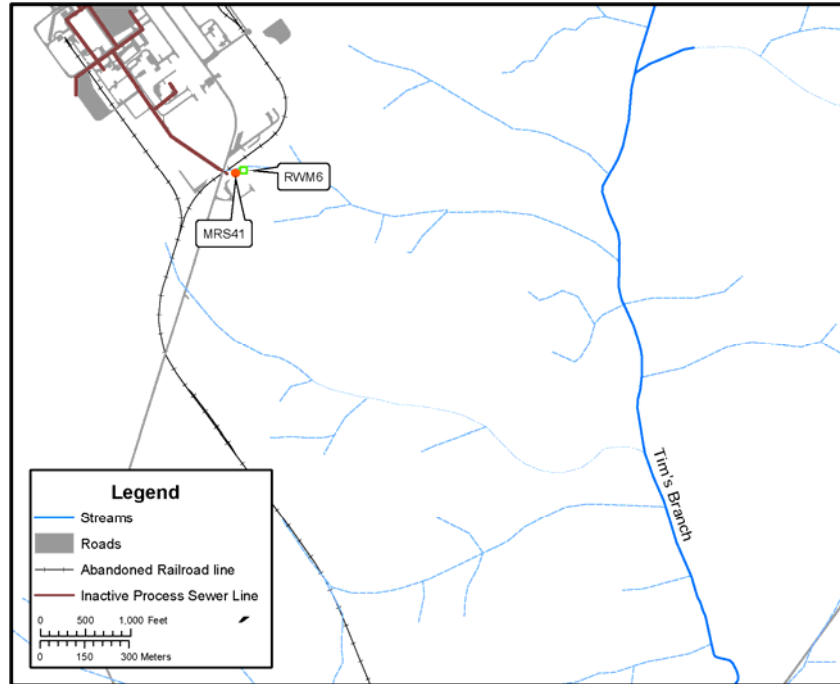


Figure 3. Location of Borehole in the A/M Area where Intact Sediment Cores were Collected for Laboratory Study

3.0 Sample Preparation and Analysis Methodology

3.1 Sediment Bulk Fraction Distribution Analysis

The wet sieving/hydrometer method (ASA 1986), Part 1, was applied in duplicate to determine the particle size distribution of all four sediment samples. Sediment sub-samples were never air- or oven-dried to minimize the effects of particle aggregation that can affect the separation of clay grains from the coarser material. The quartz density default value of 2.65 g/cm^3 was used to calculate the particle size distribution of Savannah River sediments based on their classification as sand-dominated sediments. The error in using this simplifying assumption has been shown to be insignificant when applied to sand-dominated sediments such as those found at DOE's Hanford Site (Serne et al. 2004).

3.2 Carbon Analysis

The total carbon content of the sediment samples (<2 mm fraction) was performed in duplicate and determined using an internal PNNL procedure based on standard method (ASTM 2001). Total carbon in all samples was determined using a Shimadzu TOC-V Total Organic Carbon Analyzer with combustion at approximately 980°C (1796°F).

Inorganic carbon contents for the sediment samples (<2 mm fraction) were also determined in duplicate using the Shimadzu TOC-V Total Organic Carbon Analyzer. Inorganic carbon was released through

acid-assisted evolution (50% hydrochloric acid) with heating to 200°C. Organic carbon was calculated as the difference between the measured total and organic carbon.

3.3 Surface Area Analysis

The surface areas of bulk sediments were determined using a combination of nitrogen adsorption BET (Brunauer et al. 1938) and calculation using geometric formulas for the surface area and volume of a sphere. Due to sample size restrictions on the instrumentation (Quantachrome Autosorb-6B; Quantachrome Instruments, Boynton Beach, Florida), only particles less than 2 mm in diameter were measured via BET. The remaining material was sieved into five fractions representing discrete particle diameters: 2.00 to 4.75 mm, 4.75 to 9.5 mm, 9.50 to 19.0 mm, 19.0 to 25.0 mm, and >25.0 mm.

3.4 Experimental Desorption System and Solute Elution

Four column desorption experiments were performed, designated T30 (PGRS-1, 62.5 to 63 ft), T-31 (PGRS-1, 63 to 63.5 ft), T38 (MRS-41, 138.5 to 139 ft), and T-39 (MRS-41, 139 to 139.5 ft). Each column desorption experiment consisted of injecting oxygen-saturated deionized water into the solute-laden sediment core at a constant flow rate for 20 to 100 pore volumes and collecting effluent samples for solute (PCE, TCE) analysis (Figure 4). The column system was designed specifically to minimize mass loss of the volatile solutes during desorption experiments. A 20-L (5.28-gal) bottle containing the water was connected to a Hitachi L-6200 HPLC pump, which supplied a constant flow rate (~0.54 ml per minute) through stainless steel tubing to the column inlet (total volume 1,010 cm³) located at the bottom of the column. The column effluent was plumbed to a flow-through electrical conductivity (EC) electrode (20 µl volume) and then to a Valco 10-way sampling valve and a Kloehn syringe pump with an eight-channel multiplexing head. The EC electrode was connected to an EC meter and a data logging system. EC data was collected at a rate of 2 points per second and averaged for 1 minute. The resulting data was used as a conservative breakthrough tracer based on the small change in electrical conductivity between the groundwater in the sediment column and influent water. On the effluent end of the column, water flowed through a 1.000 ml stainless steel loop connected to the Valco valve and then to an effluent bottle. At specified times (controlled by a computer program), this loop was isolated from the effluent stream and connected to the Kloehn syringe pump. The syringe pump pulled a 0.1-ml waste sample (ejected to waste), followed by a 1.0 ml sample that was injected into a 40-ml VOA vial with zero headspace. Zero headspace was maintained in each vial by releasing water displaced by the injected 1-ml sample into a syringe attached to the vial (i.e., the syringe needle penetrated through the vial septum into the bottom of each water filled vial). This system was used as a fraction collector, and sampling was more frequent initially (every 75 minutes 1 to 1.75 pore volumes), then less frequent. From 1.75 to 5 pore volumes the sample collection frequency was one per 300 minutes. From 5 to 20 pore volumes, the sample collection frequency was one per 40 hours, giving a total experiment time of 200 to 220 hours. For experiments T38 and T39, sampling continued to 95 pore volumes (1000 hours) at a frequency of one sample every 90 hours.

To fully characterize the slow release of organic contaminants from the sediments, a stop-flow time period was used in experiments T38 and T39. Contaminants can be released slowly enough that even during the residence time of the experiments (about 10 hours in this case), the concentration of solutes is not detectable in the column effluent. Therefore, during the desorption tail portion of the column

experiment (at about 60 pore volumes, when the solute concentration decreased to detection limits), the flow was stopped for 237 hours. Flow was then resumed for an additional 30 pore volumes. The increase in solute concentration by the stop flow event (and stop flow time) can be used to calculate the contaminant release rate on the kinetically-controlled portion of the desorption profile.

After completion of a desorption experiment, the column was dismantled, and sub-samples of the sediment core were analyzed for solutes according to the procedures described in Sections 3.5 and 3.6.

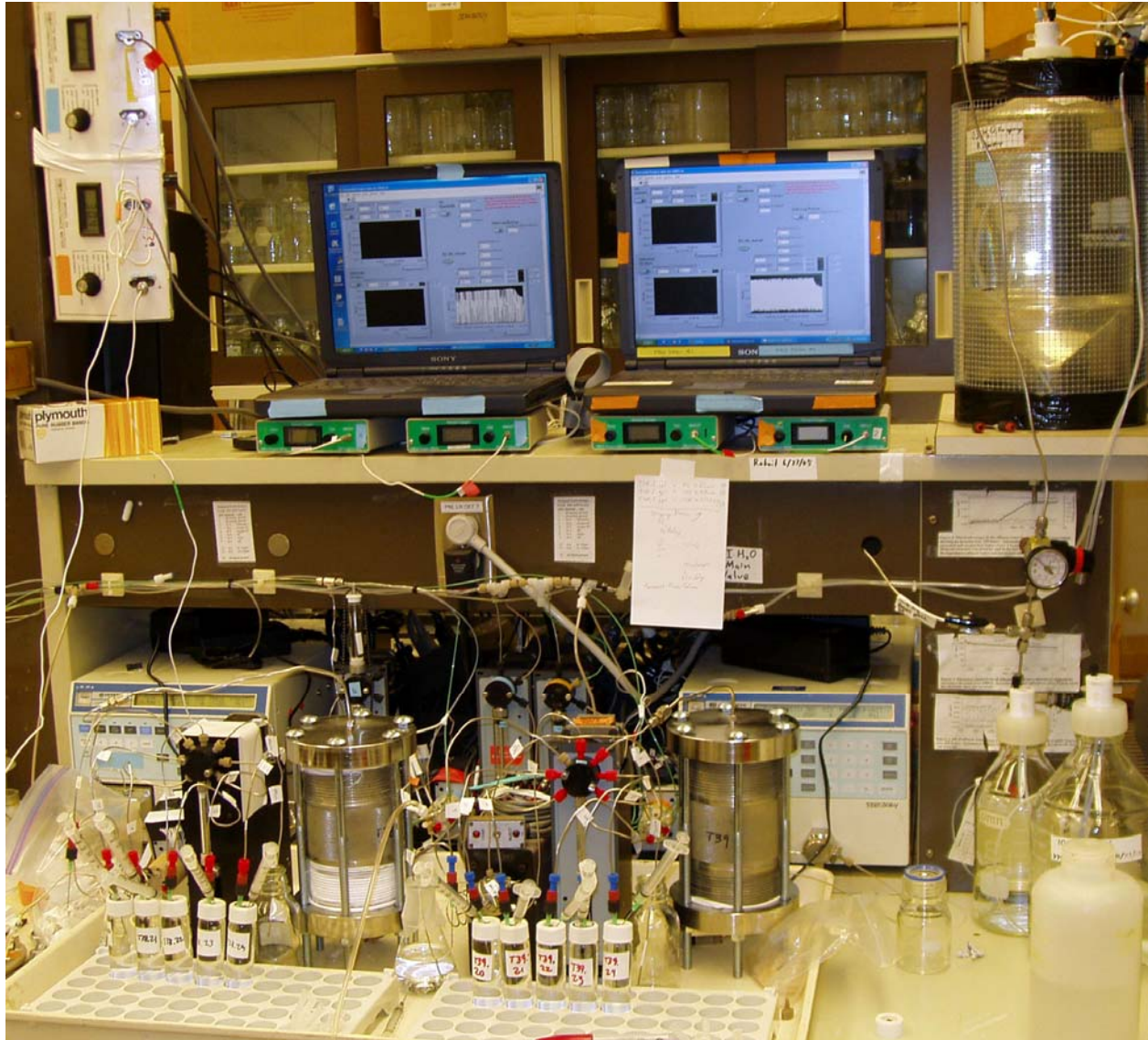


Figure 4. Desorption System Setup (Column shown connected to multiple switching valve to which is attached multiple column effluent vials.)

3.5 Analysis of Water Samples

Effluent samples from desorption experiments were collected in 40-ml VOA vials. An aliquot of the headspace from the VOA vials was analyzed for TCE and PCE using a Hewlett Packard 5890 gas chromatograph equipped with a HP 5989A mass spectrometer and an OI Analytical 4560 purge and trap system. A capillary column J&W Scientific, DB-624, 75 m, inner diameter (ID) 0.45 mm was used to separate solute compounds. In order to provide optimum sensitivity and selectivity, the selected ion monitoring mode of the mass spectrometer was used. The detection limit for TCE and PCE in water samples was 0.01 µg/L.

3.6 Extraction of Sediments

A DIONEX accelerated solvent extraction system (ASE-200) was used to extract residual TCE and PCE from the post-column desorption experiment sediments using methanol as the extraction solvent. Recovery of aqueous solutions of CCl₄ spiked into sediments from previous studies was between 82% and 101% at an optimal temperature of 40°C, depending on the type of soil (Riley et al. 2005). Recovery of TCE and PCE was assumed to be similar to CCl₄ (i.e., 82% to 101%) given that TCE and PCE are less volatile than CCl₄ (Mackay et al. 1993).

3.7 Analysis of Sediment Extracts by Gas Chromatography and Gas Chromatography/Mass Spectrometry

Methanol extracts of sediments were diluted 50 to 200 times in boiled Milli-Q water and analyzed using a Hewlett Packard 5890 gas chromatograph fitted with a purge and trap system and photoionization and electron capture detectors. TCE and PCE were separated on a 105 m x 0.53 mm megabore capillary column (Restek Corporation) and quantified using a commercial standard (SUPELCO EPA 8260A Calibration Mix) and four-level calibration. The detection limit for TCE and PCE was 0.02 µg/L.

Purge and trap GC/MS was used to analyze samples containing low concentrations of TCE and PCE. Methanol extracts of sediments were diluted 50 times in boiled Milli-Q water and analyzed using a Hewlett Packard 5890 gas chromatograph equipped with a HP 5989A mass spectrometer and an OI Analytical 4560 purge and trap system. A capillary column J&W Scientific, DB-624, 75 m, ID 0.45 mm was used to separate solute compounds. In order to provide optimum sensitivity and selectivity, the selected ion monitoring mode of the mass spectrometer was used.

4.0 Data Analysis and Modeling Methods

4.1 Determination of Solute Sorption Parameters from 1-D Column Experiments

Chemical parameters defining the mass of solutes (PCE, TCE) on the sediment surface and the rate of desorption were determined from the one-dimensional (1-D) column breakthrough curves (profiles) by area integration and with an inverse parameter estimation code (CXTFIT, Toride et al. 1993, 1999), as described below. This approach was used to characterize the physical and chemical processes that were occurring in the sediments and yielded solute partition coefficients (K_d values) and kinetic parameters describing slow solute release from sediments. Both physical and chemical processes control the shape of the tracer and solute desorption curve. Idealized 1-D flow of a nonsorbing (conservative) tracer through a homogeneous sediment column defines both a “pore volume” (volume of liquid in the 100% mobile pore space) and the physical breakthrough curve spreading caused by flow through porous media, as defined by the longitudinal hydrodynamic dispersion (D_L), as defined by:

$$D_L = D_o + \alpha_L v \quad (1)$$

where D_o is molecular diffusion, α_L is the longitudinal dispersivity, and v is the interstitial velocity. Idealized transport of a sorbing solute in the same homogeneous sediment column will be subject to the same longitudinal dispersion (producing breakthrough curve spreading) but will lag relative to the tracer due to the reversible sorption/desorption (i.e., the overall solute velocity is slower because the solute is sorbed on the surface a portion of the time). The “retardation factor” (R_f) is defined by the ratio of velocities of the tracer/solute so that a value of 1.0 indicates no sorption (solute travels at the same velocity as the tracer) and a value >1.0 indicates sorption. With no additional assumption as to the sorption mechanism or rate, the solute retardation factor can be determined by integrating the area in front of the solute breakthrough curve. The sorption mass parameter K_d is defined by the mass of solute on the sediment surface (per gram of sediment) to the mass of solute in aqueous solution (per mL of solution), and can be calculated from R_f by:

$$R_f = 1 + \rho_b K_d / \theta \quad (2)$$

where ρ_b is the dry bulk density (g/cm^3) and θ is the total porosity. The dry bulk density was calculated from sediment dry weight and total column volume for each experiment. Column porosity was calculated from values of sediment wet weight and dry weight for each column experiment. In this study, the K_d values were determined by area integration, as described above, which accounts for all of the solute lag in retardation relative to a tracer in both homogeneous and heterogeneous sediments. Model fits to breakthrough data (equilibrium, first-order, and two-region models) also yielded K_d values, but these were not considered as accurate as area integration, because in some cases the model fit was poor. In general, a poor model fit meant that mathematical description of the physical and chemical processes was insufficient to describe the actual data.

4.2 Model Fitting of Breakthrough Curve Data

Reactive transport modeling was used to determine the solute desorption rate from the breakthrough data using the CXTFIT code (Toride et al. 1993, 1999). The CXTFIT code contains analytical solutions to an equilibrium model, a first-order model, and a two-region model. This equilibrium model is defined by the differential Equation (3):

$$\frac{\rho_b}{\theta} \frac{\partial S}{\partial t} = D_L \frac{\partial^2 C}{\partial z^2} - v \frac{\partial C}{\partial z} \quad (3)$$

The model includes three parameters; velocity (v), retardation factor, (R_f), and longitudinal dispersion (D_L) to describe the rate of change in the solute or tracer aqueous concentration (C) or surface concentration (S), as first described by Gleuckauf (1947). An analytical solution to this model with a nonlinear least squares parameter estimation routine was first described by van Genuchten and others (1974, 1979).

The first-order model, by incorporation of the kinetic reaction (4),

$$C \xrightleftharpoons[k_b]{k_f} S \quad (4)$$

accounts for reversible slow adsorption and slow desorption where k_f is the forward rate coefficient and k_b is the backward rate coefficient. An advective-dispersive transport with this single, reversible, linear adsorption/desorption reaction is defined by the solution to the differential Equations (5) and (6):

$$\frac{\partial C}{\partial t} + \frac{\rho_b}{\theta} \frac{\partial S}{\partial t} = D \frac{\partial^2 C}{\partial z^2} - v \frac{\partial C}{\partial z} \quad (5)$$

$$\frac{\rho_b}{\theta} \frac{\partial S}{\partial t} = k_f C - k_b S \quad (6)$$

with previously defined parameters. This first-order model contains four parameters; velocity, longitudinal dispersion, and the two reaction rate parameters (k_f , k_b). The equilibrium distribution coefficient is by definition k_f/k_b . The first-order kinetic model was first fit to 1-D solute transport data in sediments by Leenheer and Ahlrichs (1971).

The two-region model describes solute advective/dispersive transport through a porous media with both mobile (subscript “e”) and immobile (subscript “i”) pore regions and equilibrium sorption in both regions (van Genuchten et al. 1974), as defined by the differential Equations (7) and (8):

$$\theta_e \frac{\partial C_e}{\partial t} + \theta_i \frac{\partial C_i}{\partial t} + f \rho_b \frac{\partial S_e}{\partial t} + (1-f) \rho_b \frac{\partial S_i}{\partial t} = D_L \frac{\partial^2 C_e}{\partial z^2} - v \theta_e \frac{\partial C_e}{\partial z} \quad (7)$$

$$\theta_i \frac{\partial C_i}{\partial t} + (1-f) \rho_b \frac{\partial S_i}{\partial t} = \alpha_e (C_e - C_i) \quad (8)$$

where f is the fraction of sorbent in the mobile region, C_e and C_i are solute concentrations in the mobile and immobile regions, respectively; S_e and S_i are the respective sorbed concentrations, and θ_e and θ_i are the volume fractions of the mobile and immobile liquid regions. Mobile pore regions are those regions in where advective transport of pore fluid occurs. Immobile pore regions are those regions in a porous media where pore fluid is trapped (e.g., in media microfractures). This two-region model has five parameters: velocity (v), longitudinal dispersion (D_L), equilibrium sorption ($K_d = S/C$), diffusional mass transfer between mobile and immobile pore fluid (α_e), and the fraction of solute mass in the mobile region (f). The two-region model is mathematically equivalent to a fast and slow reaction in parallel or in series.

Model fits to breakthrough data (equilibrium, first-order, and two-region models) were evaluated in the context of three general scenarios:

- (i) equilibrium model shows a good fit to both tracer and solute breakthrough data. This indicates an equilibrium adsorption/desorption reaction ($K_d = S/C$) could explain the observed desorption behavior.
- (ii) equilibrium model fit to tracer data, but solute data has a poor fit with the equilibrium model and a good fit with the first-order model. This indicates equilibrium sorption was not the only process occurring, and that reversible first-order kinetics was required to represent observed desorption behavior.
- (iii) equilibrium model fit to tracer data, but solute data has a poor fit with the equilibrium and first-order model but a good fit with the two-region model. This indicates that a complex set of either two physical or two chemical processes (i.e., one kinetic and one at equilibrium) are required to represent observed desorption behavior).

Because both hydrodynamic dispersion and slow sorption/desorption will define the solute breakthrough curve shape, a systematic modeling method was used to accurately define these separate processes. In step 1, the equilibrium model was used to fit the tracer data in order to define the hydrodynamic dispersion (longitudinal) with a defined velocity and defined retardation factor (1.0; i.e., the velocity and retardation factor were not allowed to vary in the simulation, only the dispersion). In step 2, the solute data was fit with this equilibrium model with the velocity fixed and longitudinal dispersion fixed at the tracer value (i.e., allowing R_f to vary). If a good fit of the solute data was observed with equilibrium model simulation, then it was concluded that no significant slow release of the solute from the sediment was occurring in the desorption process. For those desorption profiles where additional breakthrough curve spreading was observed (i.e., greater than that defined by longitudinal dispersion), indicating the presence of slow desorption of the solute from the sediment surface, first-order and two-region models were used to fit the solute data to evaluate the nature of the slow desorption process. In these simulations, the longitudinal dispersion was fixed at the value determined by the equilibrium model fit to the tracer data. “Goodness of fit” was described by comparing breakthrough data to model fit (sum of the squares difference between data and model fit at each point) and comparison of the breakthrough area of the actual data to the simulation.

5.0 Results

5.1 Physical and Chemical Characteristics of Sediments

Table 1 summarizes the physical-chemical characteristics of three sediment samples taken adjacent to the sediment cores (i.e., within 15.24 cm [6 in.]) used in the desorption experiments. Properties of these sediments were assumed to be the same as those used in the desorption experiments based on visual comparison of the sediments' bulk-fraction features. Samples A and B, representing sediments co-located with PGRS-1 intact cores, had similar distributions with approximately 77 to 88% of the mass associated with the gravel and sand fractions. This is in contrast to sediment sample C, representing MRS-41 intact cores which contained a high sand content (87.9%) and virtually no gravel. Silt contents varied between 1.7% and 5.6%, and clay contents varied between 8.3% and 17.1%. The highest silt/clay content was found in sample A.

Table 1. Properties of Co-Located Sediment Samples

Property	Co-Located Sediment Sample		
	PGRS-1 A	PGRS-1 B	MRS-41 C
>9.5 mm (%)	0.6	1.1	0
4.75 to 9.5 mm (%)	1.3	8.0	0
2.00 to 4.75 mm (%)	10.4	15.2	1.9
Sand (%)	65.0	63.3	87.9
Silt (%)	5.64	4.15	1.71
Clay (%)	17.1	8.26	8.72
Total Carbon (%)	0.03	0.02	0.03
Inorganic Carbon (%)	0.00	0.00	0.00
Organic Carbon (%)	0.03	0.02	0.03
Surface Area (m ² /g)	8.53	3.52	8.85
Represented Corresponding Desorption Experiment	Experiment T-30	Experiment T-31	Experiments T-38 and T-39

The organic carbon content of the sediments was low (<0.05%), consistent with previous findings for SRS sediments (Denham et al. 2000). Inorganic carbon was not detected in samples A, B, and C. Samples A and C had comparable surface areas (8.5 m²/g vs. 8.8 m²/g). Sample B contained less than half the surface area (3.5 m²/g) of samples A and C consistent with more mass in the larger particle-size fractions.

5.2 Desorption Experiment Design

Electrical conductivity, TCE and PCE desorption profiles of the four sediment cores are depicted in Figures 5, 6, 7, and 8. Table 2 summarizes the conditions in which desorption experiments were conducted. Column residence time for all four experiments was approximately 10 to 11.5 hrs/pore volume (interstitial velocity was ~1.0 cm/h). For PGRS-1 sediments, experiments were run for approximately 10 days (255 hours) with a total of 22 to 23 pore volumes of water passing through the columns. Experiments conducted with MRS-41 sediments included a stop flow step to allow the potential for determination of TCE and PCE release rates at locations along the slow release portion of the desorption curve. Stop flow for both experiments occurred after 60 pore volumes and lasted for 237 hours before flow was continued for another 26 pore volumes of desorption.

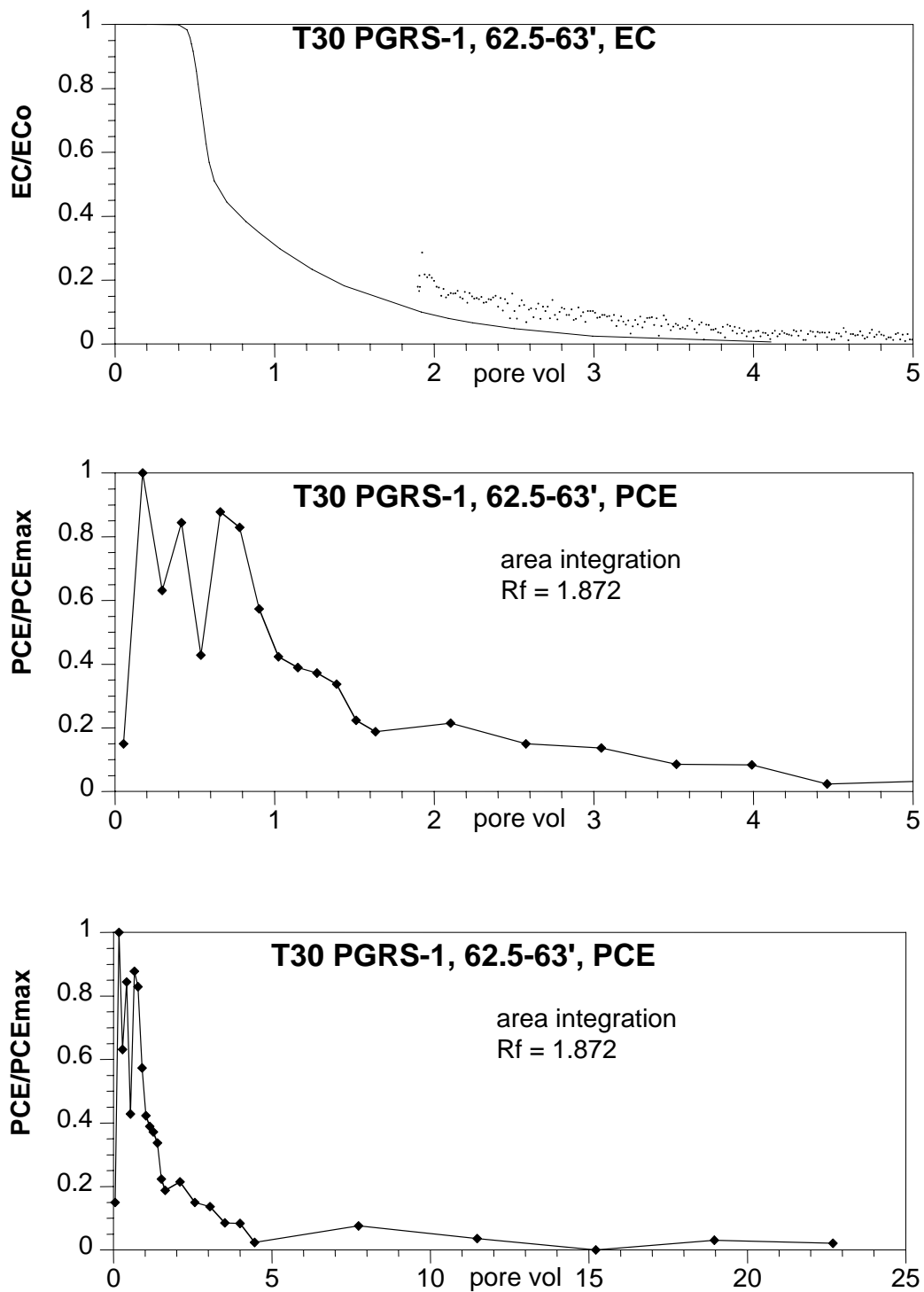


Figure 5. Column Experiment T30 Tracer (a), First Five Pore Volumes of PCE Data (b), and All PCE Data (c). (Note differences in pore volume scales and that there is no data for TCE for this experiment.)

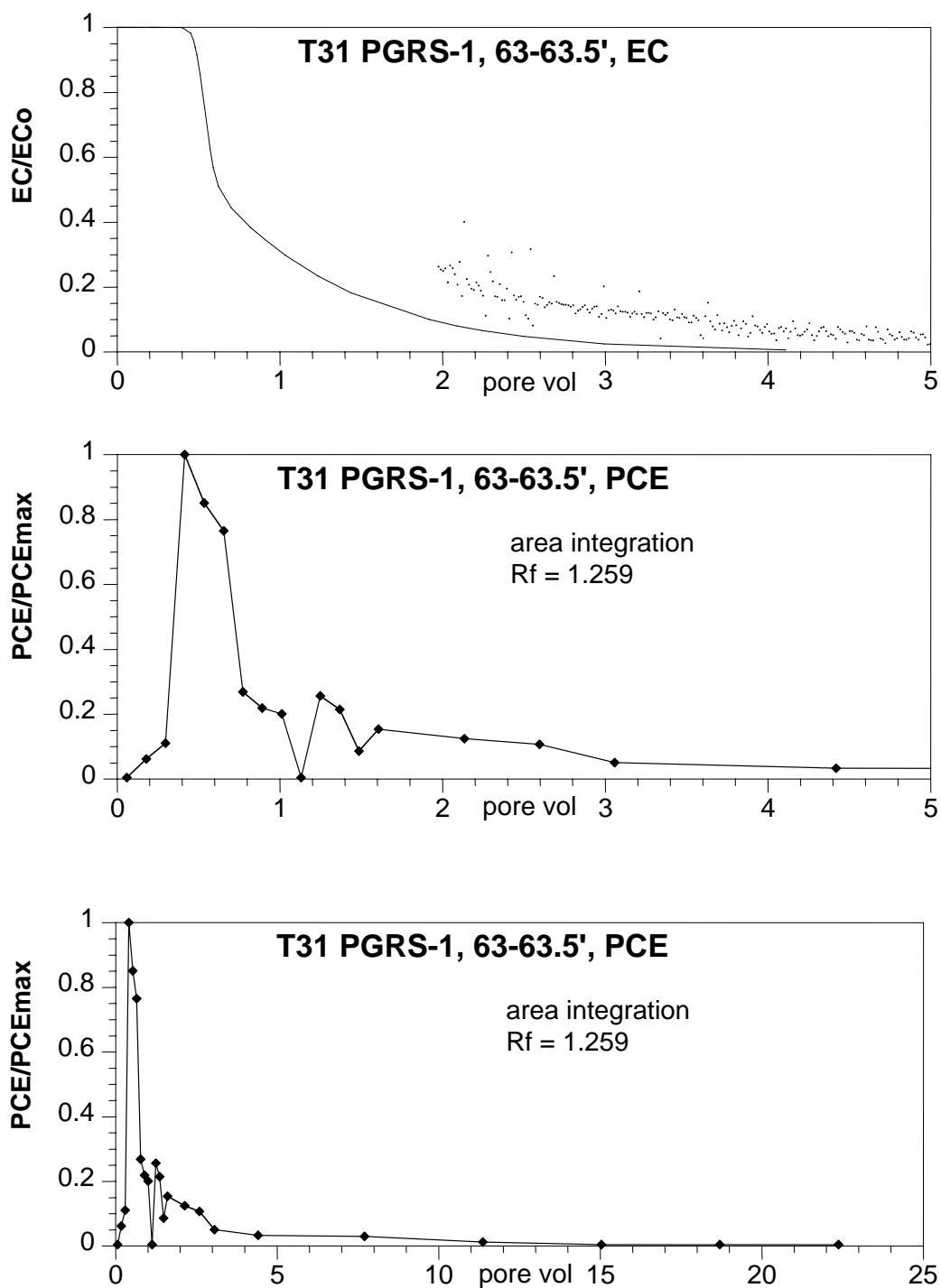


Figure 6. Column Experiment T31 Tracer (a), First Five Pore Volumes of PCE Data (b), and All PCE Data (c). (Note differences in pore volume scales and that there is no TCE data for this experiment.)

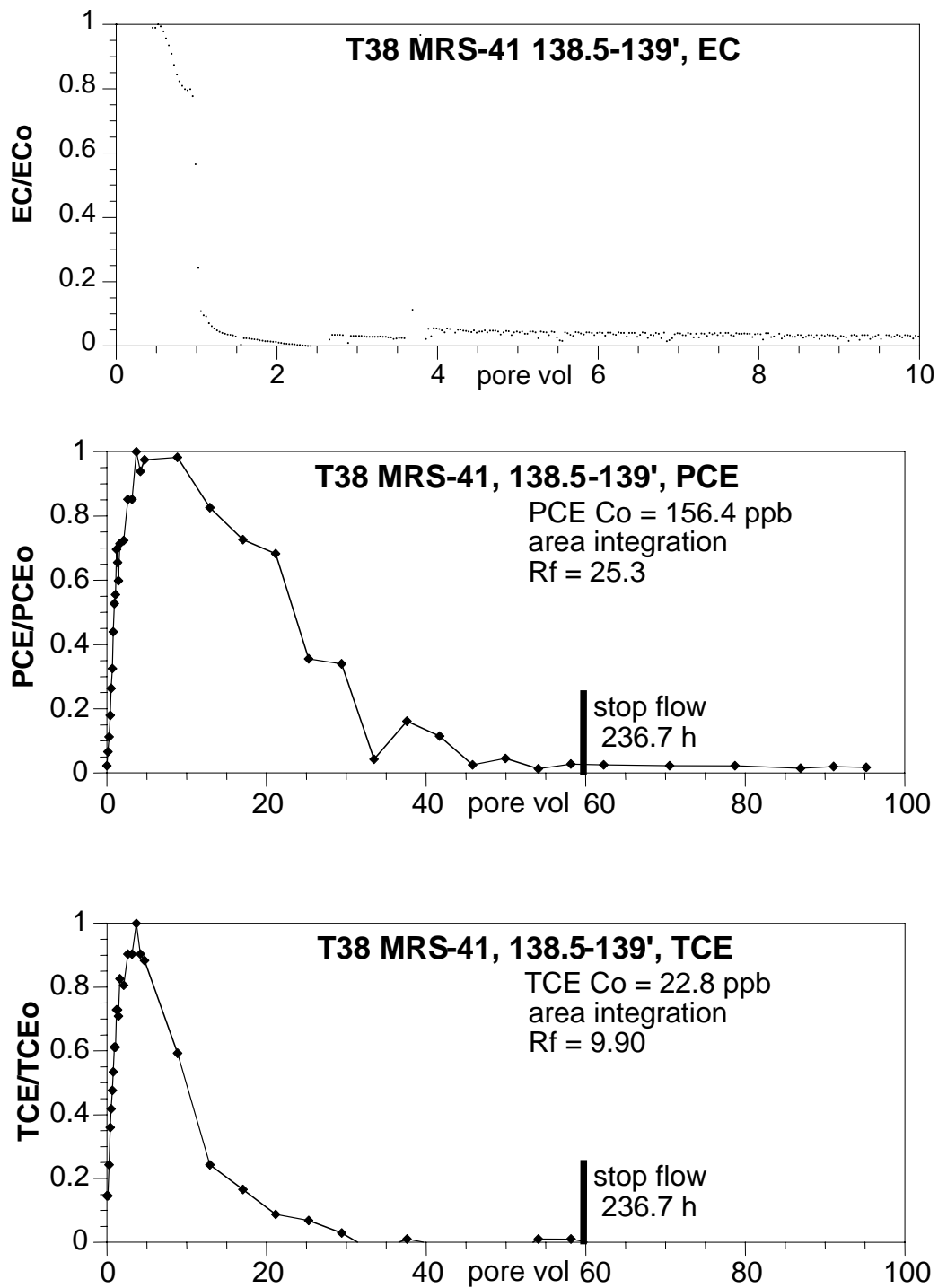


Figure 7. Column Experiment T38 Tracer Data (a), PCE Data (b), and TCE Data (c). (Note differences in pore volume scales.)

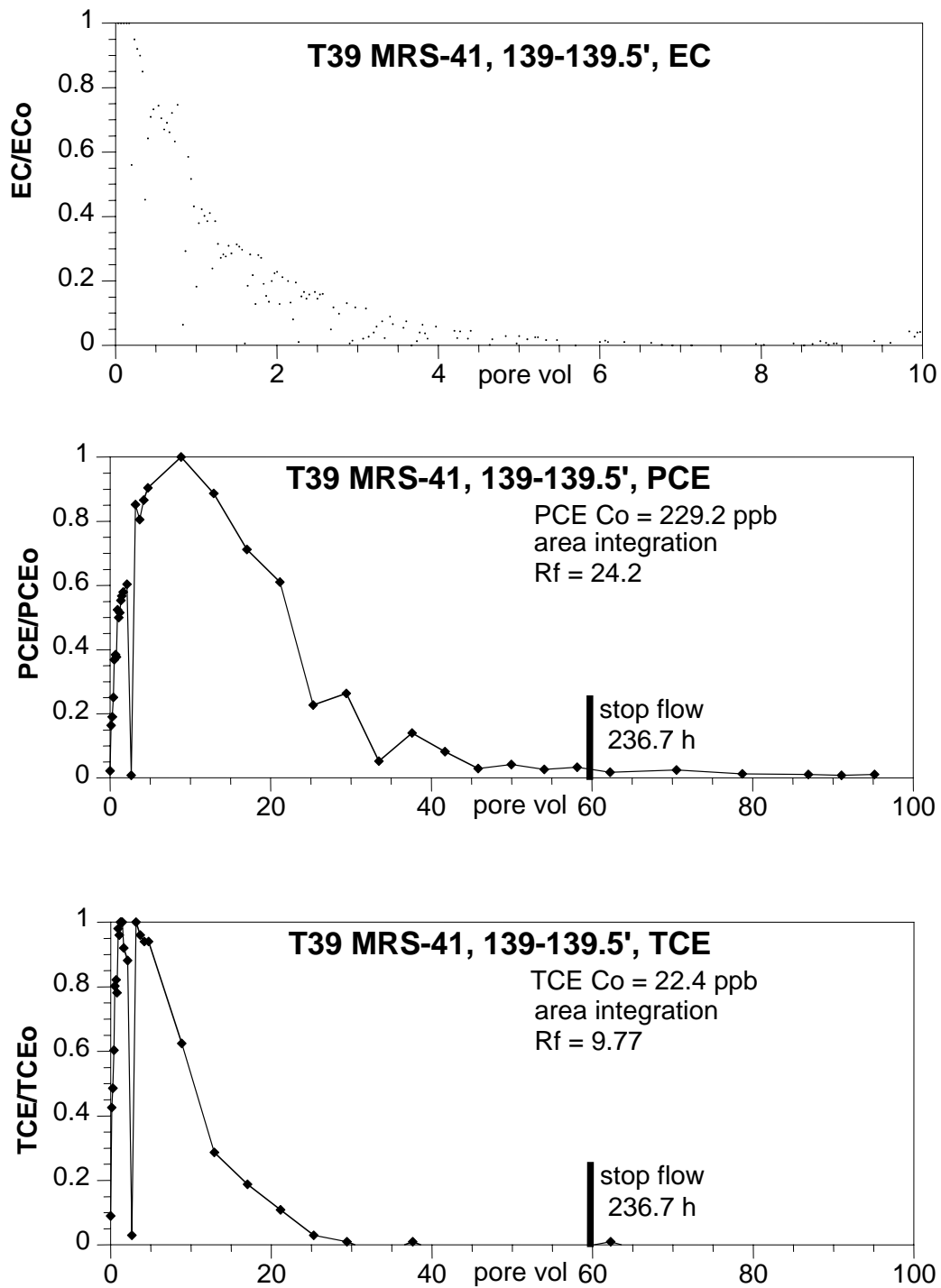


Figure 8. Column Experiment T39 Tracer Data (a), PCE Data (b), and TCE Data (c). (Note differences in pore volume scales.)

Table 2. Desorption Experiment Conditions

Sediment/Experiment	Flow Rate (hrs/pore volume)	Pore Volumes	Length of Desorption (hrs)
PGRS-1 (T-30)	11.24	23	255
PGRS-1 (T-31)	11.47	22	255
MRS-41 (T-38)	10.23	60 36	614 ^(a) 368
MRS-41 (T-39)	10.25	60 36	615 ^(a) 368
(a) After the first 60 pore volumes, flow was stopped for 237 hrs. Flow was then initiated for another 36 pore volumes.			

5.3 TCE and PCE Distribution Coefficients

Table 3 summarizes TCE and PCE K_d values calculated from measured values of retardation factor and sediment properties (i.e., dry bulk density and total porosity). TCE K_d values of 1.96 L/kg and 1.94 L/kg were determined for MRS-41 samples. Retardation factors measured from PCE desorption profiles of PGRS-1 sediment cores resulted in calculated K_d values of 0.192 L/kg and 0.055 L/kg. For MRS-41 sediment cores, PCE K_d values of 5.31 L/kg and 5.11 L/kg were calculated. PCE K_d values for MSR-41 cores were 10 to 100 times higher than K_d values calculated for PCE in PGRS-1 cores. TCE was not detected in the effluents from PGRS-1 sediment cores. Therefore, neither TCE desorption profiles nor TCE K_d values were generated for these samples.

Table 3. Calculation of K_d Values

Compound	Sediment	Density (ρ_b) g/cm ³	Porosity (θ) cm ³ /cm ³	Retardation Factor (R_f)	K_d (cm ³ /g or L/Kg) ^(a)
PCE	PGRS-1 (62.5 to 63 ft)	1.570	0.352	1.87	0.192
PCE	PGRS-1 (63.0 to 63.5 ft)	1.650	0.352	1.26	0.055
TCE	MRS-41 (138.5 to 139.0 ft)	1.595	0.352	9.90	1.96
TCE	MRS-41 (139 to 139.5 ft)	1.595	0.352	9.77	1.94
PCE	MRS-41 (138.5 to 139.0 ft)	1.595	0.352	25.3	5.36
PCE	MRS-41 (139 to 139.5 ft)	1.595	0.352	24.2	5.11
(a) $K_d = (R_f - 1)\theta/\rho_b$.					

Distribution coefficient values from this study (Table 3) were compared to those that we estimated from equations that calculate K_d values based on sediment organic fraction (f_{oc}) and estimates of the normalized sorption coefficient (K_{oc}) as well as from octanol-water distribution coefficient (K_{ow}) and surface area data (Table 4). Calculated PCE K_d values for PGRS-1 samples (this study) were similar to those estimated based on f_{oc} and K_{oc} data but were approximately a factor of five less than estimates made based on K_{ow} and sediment surface area data. Calculated TCE K_d values for MRS-41 samples (this study) were 60 to 150 times higher than those estimated based on f_{oc} and K_{oc} data and approximately a factor of five higher than estimates made based on K_{ow} and sediment surface area data. Calculated PCE K_d values for MRS-41 samples (this study) were about 50 times higher than those estimated based on f_{oc} and K_{oc} data

Table 4. Calculated versus Estimated and Recommended Values of K_d

Sediment (depth-ft)	Distribution Coefficient (K_d) L/Kg					
	TCE Calculated	TCE (Est) ^(a)	TCE (Est) ^(b)	PCE Calculated	PCE (Est) ^(a)	PCE (Est) ^(b)
PGRS-1 (62.5 to 63)	ND ^(c)	ND	NA	0.192	0.106	0.454
PGRS-1 (63.0 to 63.5)	ND	ND	NA	0.055	0.071	0.452
MRS-41 (138.5 to 139.0)	1.96	0.032	0.408	5.35	0.106	0.454
MRS-41 (139.0 to 139.5)	1.94	0.013	0.408	5.11	0.106	0.454
<p>(a) K_d estimated based on the equation: $K_d = f_{oc} \times K_{oc}$ where: $f_{oc} = 0.0003$ for sediment samples PGRS-1 (62 to 62.5 ft) and MRS-41 (138.5 to 139 ft and 139.0 ft to 139.5 ft) and 0.0002 for sediment sample PGRS-1 (63.0 to 63.5 ft) and K_{oc} for TCE (109.65 L/kg) and PCE (354.8 L/kg) is the average of literature-cited values for K_{oc} in soils (Mackay et al., 1993).</p> <p>(b) K_d estimated based on equation: $\log K_d = 0.16 \log K_{ow} + \log S_a/200$ (McCarty et al. 1981) where $\log K_{ow}$ for TCE (2.52) and PCE (2.81) is the average of literature cited values (Mackay et al. 1993) and S_a is the sediment surface areas reported in Table 1.</p> <p>(c) TCE not detected in the column effluent or sediment of the desorption experiment.</p>						

and approximately a factor of five higher than estimates made based on K_{ow} and sediment surface area data. Previously, values of TCE and PCE K_d recommended for SRS fate and transport studies were 0.013 and 0.038, respectively (Looney et al. 1987). The average TCE K_d value from MRS-41 sediments was over 100 times higher (0.013 L/kg versus 1.95 L/kg) than those previously estimated. The average PCE K_d value from PGRS-1 sediments was a factor of 3 higher (0.038 L/kg versus 0.012 L/kg) than previously estimated, and the average PCE K_d value from MRS-41 sediments was over 100 times higher (0.038 L/kg versus 5.24 L/kg) than previously estimated.

5.4 Comparison of SRS TCE and PCE K_d Values to those of Other Contaminated Sites

TCE and PCE K_d values calculated in this study were compared to values obtained for aquifer sediments of low organic carbon content (i.e., <0.1%) from other United States (Lawrence Livermore National Laboratory, Livermore California, Lula, Oklahoma) and Canadian (Borden Air Force base, Ontario, Canada) sites containing subsurface volatile chlorinated hydrocarbon contamination (Table 5). K_d values for PCE in SRS P-Area sediments (0.055 L/kg to 0.192 L/kg) were similar to the range in values (0.01 L/kg to 0.81 L/kg) at the other sites. In contrast, TCE and PCE K_d values for sediments from the SRS A/M area were approximately a factor of ten or more higher than those from the other sites, including those values (PCE only) obtained for sediments at the SRS P-Area.

Table 5. Site Comparisons of TCE and PCE K_d Values

Site/Reference	K_d (L/kg) Sorption (S), Desorption (D)		Organic Carbon (%)	Surface Area (m ² /g)
	TCE	PCE		
Savannah River-PGRS-1 (62.5 to 63 ft)		0.192 (D)	0.02	8.53
Savannah River-PGRS-1 (63 to 63.5 ft)		0.055 (D)	0.03	3.52
Savannah River-MRS-41 (138.5 to 139 ft)	1.96 (D)	5.36 (D)	0.03	8.85
Savannah River-MRS-41 (139 to 139.5 ft)	1.94 (D)	5.11 (D)	0.03	8.85
Lawrence Livermore National Lab (LLNL)-aquifer sediments (Hoffman 1995)	0.19-0.27 (D)	0.27-0.81 (D)	<0.1	
Lula OK-aquifer sediment (Piwoni and Banerjee 1989)	0.032 (S)	0.19 (S)	0.021	11±1.2
Borden Air Force Base, Ontario, Canada-aquifer sediment (Piwoni and Banerjee 1989)	0.088 (S)	0.35 (S)	0.031	<10
Borden Air Force Base, Ontario, Canada-aquifer sediment (Curtis et al. 1986)		0.45±0.02 (S) 0.31-0.53 ^(a) 0.31-0.89 ^(a)	0.02	0.8
LLNL, Arroyo Seco PCE plume-aquifer sediments (Bishop et al. 1991)		0.34-0.35 ^(b) 0.01-0.05 ^(b) 0.04-0.05 ^(b) 0.11-0.13 ^(b)	<0.1	
Borden Air Force Base, Ontario, Canada-aquifer sediment (Ball and Roberts 1991a)		0.76±0.21 (S)	0.021	0.42-1.7
(a) K_d estimated from contaminant travel time or velocity in groundwater.				
(b) K_d values were determined from measurement of PCE pore water and PCE sediment concentrations.				

5.5 TCE and PCE Retention on Sediments

The distribution of TCE and PCE between the column sediment and effluent was determined for each desorption experiment (Table 6). All or almost all of the mass (98.8% to 100%) of TCE and PCE resided with the effluent water, indicating the absence of very slow releasing TCE and PCE mass fractions resistant to migration.

Table 6. TCE and PCE Mass Distributions Post Desorption

Sediment (Depth-ft)	Compound	Mass-Effluent (µg)	Mass-Solvent Extraction (µg)	Total Mass (µg)	Mass Fraction in Effluent (%)	Mass Fraction on Sediment (%)
PGRS-1 (62.5 to 63 ft)	PCE	567.1	6.84	573.91	98.8	1.2
PGRS-1 (63.0 to 63.5 ft)	PCE	305.3	6.77	312.07	97.8	2.2
MRS-41 (138.5 to 139.0 ft)	TCE	139.4	ND ^(a)	139.4	100.0	0.0
MRS-41 (139 to 139.5 ft)	TCE	138.0	ND ^(a)	138.0	100.0	0.0
MRS-41 (138.5 to 139.0 ft)	PCE	1,332.5	1.00	1,333.5	99.9	0.1
MRS-41 (139 to 139.5 ft)	PCE	1,785.2	1.00	1,786.2	99.9	0.1
(a) Not detected.						

5.6 Sediment DNAPL and Porewater Concentration Assessments

Total TCE and PCE masses in the column desorption experiments (Table 6, column 5) were used to calculate maximum TCE and PCE sediment concentrations present in each sediment core prior to conducting the desorption experiments. Calculated concentrations showed good agreement with PCE concentration in co-located samples collected and analyzed independently by staff at the Savannah River National Laboratory (Table 7). Sediment concentrations were used to calculate sediment porewater concentrations as a basis for assessing whether a TCE or PCE DNAPL phase may have been present in the sediment cores prior to conduct of the sediment desorption experiments (Feenstra et al. 1991). The presence of DNAPL would confound the interpretation of the desorption K_d results if dissolution from a DNAPL phase was occurring during aqueous desorption. In all cases, porewater concentrations were significantly below TCE and PCE concentrations necessary to exceed TCE and PCE solubility in water, indicating the absence of TCE and/or PCE DNAPL in the sediments (Table 7). Estimated porewater concentration of TCE and PCE in MRS-41 sediments was compared to the concentration of TCE and PCE measured in groundwater from well MSB 31C located proximate to sediment sample collection. Groundwater data were from samples collected at about the same time of sediment sample collection (i.e., groundwater samples were collected in September 2005 and March 2006 and the sediment samples were collected in January 2006). There was good agreement between estimated porewater concentration and groundwater concentration for both TCE and PCE (i.e., concentrations were within a factor of approximately 2 to 3 of each other). Note that the high values of K_d for TCE and PCE are necessary for reasonable comparability with measured groundwater concentrations.

5.7 Breakthrough Curve Analysis

Table 8 summarizes parameter data for equilibrium, first-order, and two-region model simulations of the TCE and PCE desorption profiles from the four sediments from the two SRS sampling sites. Simulated profiles from application of equilibrium, first-order, and two-region models to tracer, TCE, and PCE data were used to assess TCE and PCE desorption behavior in PGRS-1 and MRS-41 sediments. One possibility is that TCE and PCE desorption from sediments can be explained with only the mechanisms of advective-dispersive transport (i.e., the application of a simple equilibrium model), indicating reversible sorption as the process governing their migration in SRS sediment. Another possibility is that desorption is kinetically-controlled. The simplest kinetic reaction is first-order reversible sorption (sorption to a single site), and its application in a transport model uses the sorption and desorption rate. A model that incorporates equilibrium sorption parallel to first-order kinetic sorption (two-region model) was also used to describe breakthrough curve shape. An improved fit with the first-order or second-order models over the equilibrium model suggests (but does not prove) that TCE and PCE migration is a function of some combination of reversible and irreversible sorption processes.

For the purposes of our analyses, irreversible processes are of two types. In type 1, TCE and PCE absorption into the sediment has been moderate in which case desorption behavior may be explained with a first-order model or two-region model with reversible sorption remaining the dominant process. In type 2, a significant source of TCE and/or PCE mass as a migration resistant fraction (MRF) is found remaining in the sediment after desorption. This MRF releases TCE or PCE very slowly into the migrating pore water. We assume the release rate of TCE or PCE from MRF is orders of magnitude slower than the release rate for moderately absorbed TCE and/or PCE. In this case, we assume that

Table 7. Comparison of Estimated TCE/PCE Porewater Concentrations with Measured Groundwater Concentrations

Sediment (depth-ft)	Compound	Total Solute Mass (μg)	Compound Concentration on Sediment ($\mu\text{g/kg}$) ^(a)	K_d (L/kg)	Estimated Porewater Concentration (C_w) ($\mu\text{g/L}$) ^(b)	Measured Groundwater Concentration ($\mu\text{g/L}$)
PGRS-1 (62.5 to 63 ft)	PCE	574	356 (337) ^(c)	0.192	856	
PGRS-1 (63.0 to 63.5 ft)	PCE	312	187 (113) ^(d)	0.055	697	
MRS-41 (138.5 to 139.0 ft)	TCE	139	89	1.96	41	8.7 ^(e) , 16 ^(e)
MRS-41 (139 to 139.5 ft)	TCE	138	91	1.94	42	8.7 ^(e) , 16 ^(e)
MRS-41 (138.5 to 139.0 ft)	PCE	1,334	857 (1,086) ^(f)	5.36	154	584 ^(e) , 320 ^(e)
MRS-41 (139 to 139.5 ft)	PCE	1,786	1,174 (1,220 ^(g) , 1,553 ^(g))	5.11	220	584 ^(e) , 320 ^(e)
<p>(a) Total mass \div dry weight of sediment in each of the experimental sediment desorption columns. Dry weight of sediments is: T-30, 1.611 kg; T-31 1.667 kg; T-38, 1.556 kg; T-39, 1.522 kg. This calculation assumes all TCE and PCE mass was associated with the sediment.</p> <p>(b) $C_w = C_t \times \rho_b \div (K_d \times \rho_b + \theta)$ (Feenstra et al 1991) for saturated sediment where C_t is PCE or TCE sediment concentration (column 4, this table), K_d are values in column 5 of this table, ρ_b (bulk density) and θ (porosity) are those values found in Table 3. TCE and PCE solubilities in water of 1,100 mg/L and 150 mg/L were used in the comparison (Horvath 1982).</p> <p>(c) Data from sample PGRS-1BOTTOM-DUP provided by Margaret Millings (Savannah River National Laboratory).</p> <p>(d) Data from sample PGRS-1TOP-DUP provided by Margaret Millings (Savannah River National Laboratory).</p> <p>(e) TCE and PCE data from analysis of groundwater samples from well MSB 31C located proximate to MRS-41 borehole. Groundwater measurements were made several months before and several months after sediment samples were collected from MRS-41 borehole. Data provided by Margaret Millings (Savannah River National Laboratory).</p> <p>(f) Data from sample MRS-41-138-DUP provided by Margaret Millings (Savannah River National Laboratory).</p> <p>(g) Data from sample MRS-41-140 and MRS-41-140 DUP provided by Margaret Millings (Savannah River National Laboratory).</p>						

neither the first-order or two-region model is sufficiently sensitive to discern the presence of MRF release in the presence of moderately absorbed TCE and/or PCE. Analysis of individual experiment simulations are discussed in Sections 5.7.1 through 5.7.4, taking into account previously discussed data (i.e., Table 6 and the results of the stop-flow step in the MRS-41 experiments). In the case of Figure 6a, where the electrical conductivity data set was incomplete, the model was asked to provide a “best fit” profile based on the available data.

Table 8. Column Experiment Simulation Results

v			equilibrium model					first-order model							two-region model							
			R _f	K _d	D	ssq	model/	R _f	Kd	D	kb	half-life	ssq	model/	R _f	Kd	D	α	half-life	f, mobile	ssq	model/
exp.	solute	(cm/h)		(L/Kg)	(cm2/s)		data	(L/Kg)	(cm2/s)		(1/h)	(h)		data	(L/Kg)	(cm2/s)		(1/h)	(h)	fraction		data
T30	tracer	1.11	1.0F	0.00	1.72E-01	0.061	0.860	1.1	0.021	0.172F	1E-08	7E+07	0.101	0.586	1.90	0.135	0.172F	1.98E-03	349.233	0.822	0.055	1.034
	PCE	1.11	1.05	0.008	0.172F	0.093	0.615															
T31	tracer	1.09	1.0F	0.00	3.03E-01	0.016	0.806	1.2	0.029	0.303F	37.7424	0.0184	0.016	0.806	1.33	0.050	0.303F	6.51E-04	1064.39	0.132	0.006	1.06
	PCE	1.09	0.75	0.00	.303F	0.173	0.664															
T38	tracer	1.22	0.99		1.10E-02	0.043	0.932	23	3.255	0.011F	0.3631	1.9089	0.053	0.964	24.1	3.481	0.011F	2.80E-01	2.47525	0.790	0.053	0.994
	PCE	1.22	22.6	3.25	0.011F	0.057	0.987															
	TCE	1.22	9.39	1.26	0.011F	0.084	0.982															
T39	tracer	1.23	1.0F	0.00	6.38E-01	0.022	0.964	22	3.107	0.638F	8.37819	0.0827	0.030	0.968	23.00	3.302	0.638F	2.04E-03	340.252	0.186	0.027	1.02
	PCE	1.23	21.6	3.09	0.638F	0.030	0.946															
	TCE	1.23	10.5	1.43	0.639F	0.026	1.150															

v = average interstitial velocity

D = hydrodynamic dispersion

model/data = model btc area integration/data btc area integration

 ρ_b = dry bulk density θ = porosity

F = parameter fixed in simulation

*model could not converse on solution

5.7.1 Experiment T30 (PGRS-1, 62.5 ft to 63 ft) Simulations

PCE elution in column experiment T30 (Figure 5c) showed a considerable amount of tailing (to 12 pore volumes) beyond the average PCE breakthrough of 1.87 pore volumes. Some of this tailing is accounted for in hydrodynamic dispersion as indicated by the breakthrough curve tailing to 5+ pore volumes (Figure 9a). An equilibrium model was able to account for most of the tailing observed in the tracer data (line in Figure 9a). The tracer dispersivity was adopted as the PCE dispersivity (i.e., the dispersivity value was fixed for PCE simulations so that the PCE desorption processes contribution within the total tailing in the PCE breakthrough curve could be identified). The equilibrium model when applied to the PCE data (solid line, Figure 9b) accounted for only 65% of PCE breakthrough mass (missing the 35% in the tailing), indicating that the simplified process of equilibrium adsorption with dispersion does not account for all of the observed data. The additional tailing could be matched with the two-region model with a small fraction (18%) of PCE slow release sites (desorption half life 350 hours, Table 8). This type of simulation does not prove the existence of a slow release process – the modeling only indicates the breakthrough data can be matched with most solute being transported with reversible adsorption with a small PCE fraction migrating according to a slow release process.

5.7.2 Experiment T31 (PGRS-1, 63 ft to 63.5 ft) Simulations

The PCE elution profile in column experiment T31 (Figure 6c) showed some tailing (up to 8 pore volumes) relative to the average tracer breakthrough of 1.26 pore volumes. Again, the equilibrium model could only account for 66% of the PCE breakthrough mass. Similar to experiment T30 results, the two-region model was best able to fit the PCE data (i.e., it could account for the PCE breakthrough curve tailing) with a small fraction of (21%) of slow PCE release sites (Figure 10a, 10b) (desorption half life 311 hours, Table 8).

5.7.3 Experiment T38 (MRS-41, 138.5 ft to 139 ft) Simulations

PCE and TCE breakthrough data in experiment T38 showed significantly different release behavior as compared to observed desorption behavior in PGRS-1 sediment cores. The tracer data (Figure 11a) showed very little tailing and was well simulated with the equilibrium model. The PCE (Figure 11b) and TCE (Figure 11c) data also were well described by the equilibrium model, in which TCE and PCE concentrations decreased within another 20 pore volumes of the average breakthrough. Application of the first-order and two-region models did not significantly improve the fit. The 237 hours stop flow (Figure 7b, 7c) event at 60 pore volumes yielded no increase in PCE or TCE concentration in the effluent, indicating the absence of measurable TCE and PCE slow release from any TCE and PCE remaining on the sediment. This was affirmed from the fact that no residual TCE or PCE mass was detected on the sediment after desorption. These results indicate that the irreversible sorption process was not significant in this sediment.

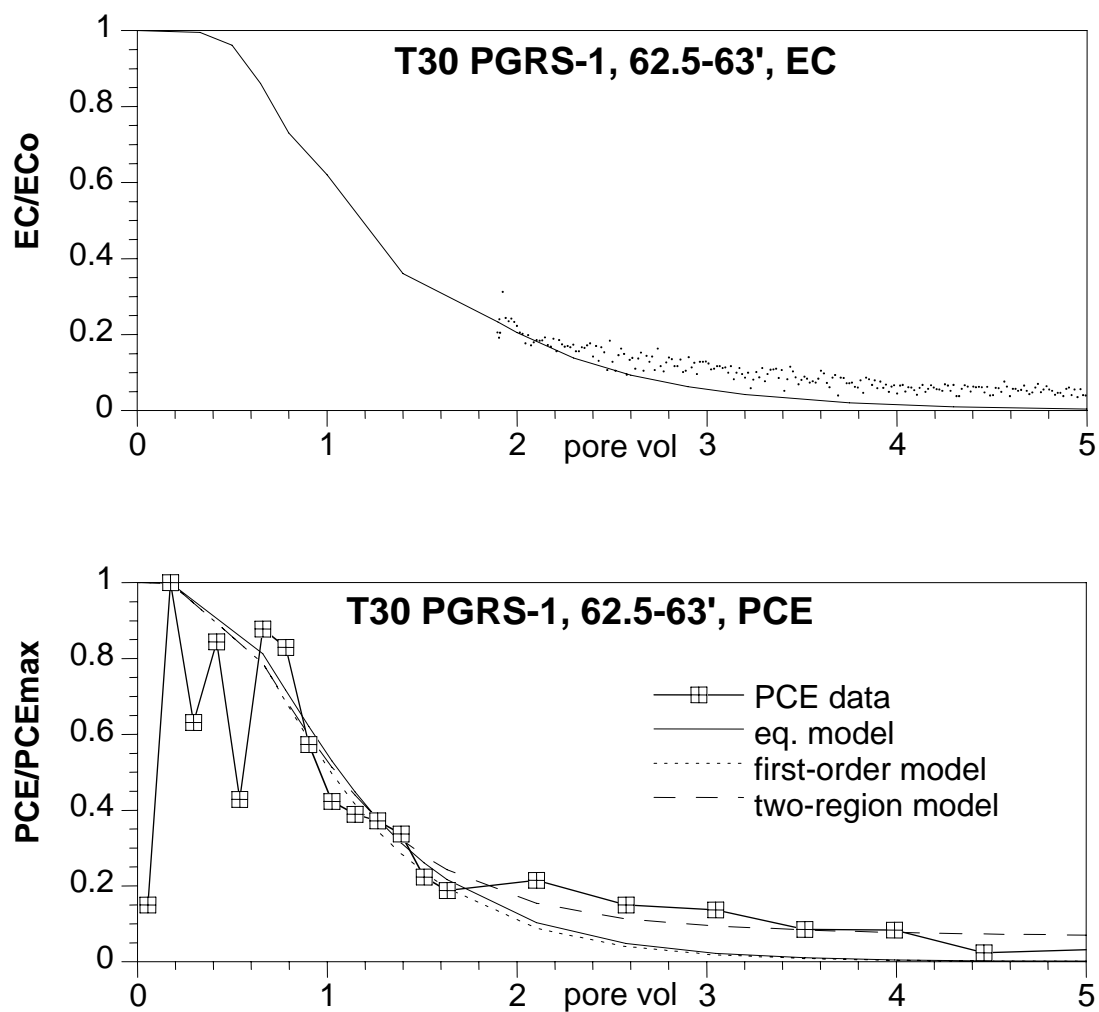


Figure 9. Simulation of Tracer (a) and PCE (b) for Experiment T30

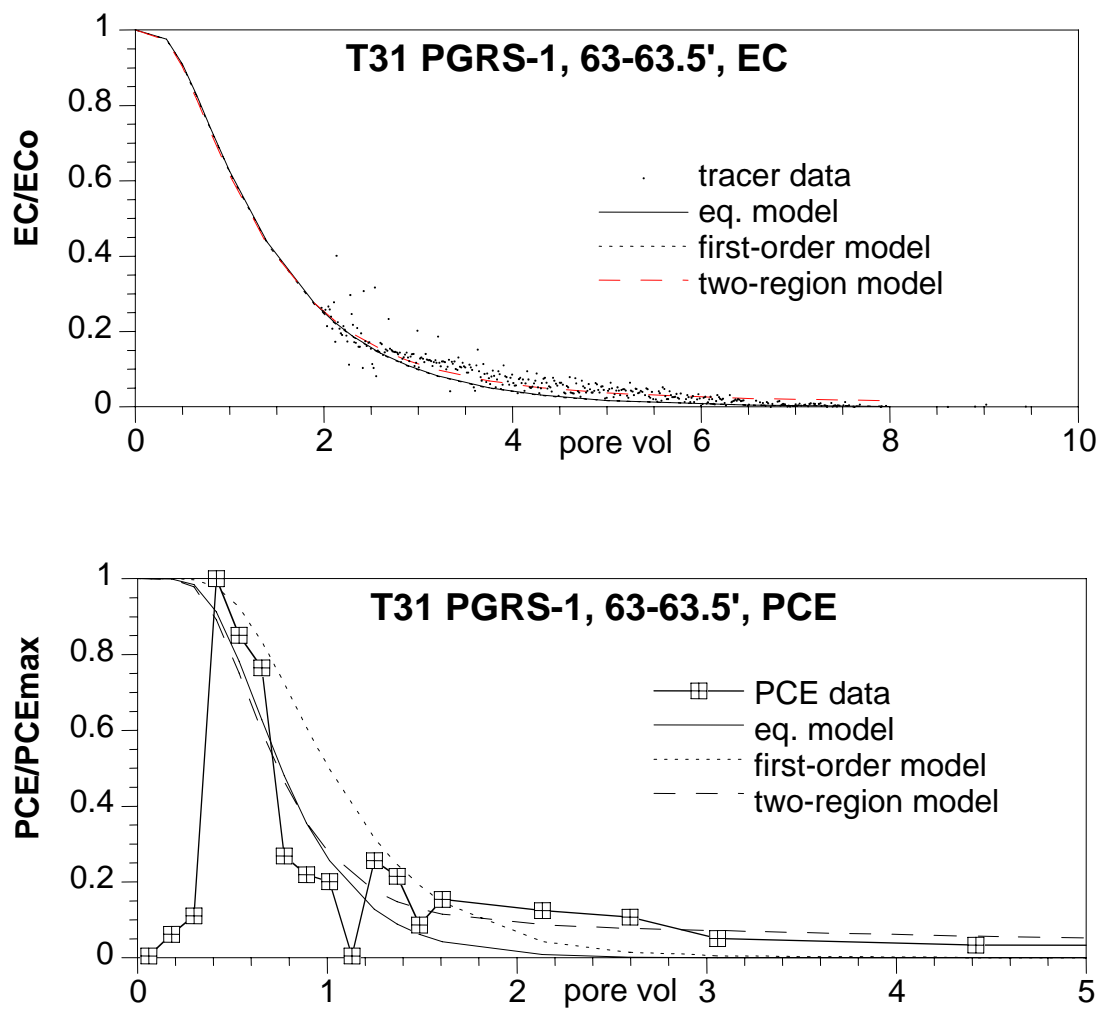


Figure 10. Simulation of Tracer (a) and PCE (b) for Experiment T31

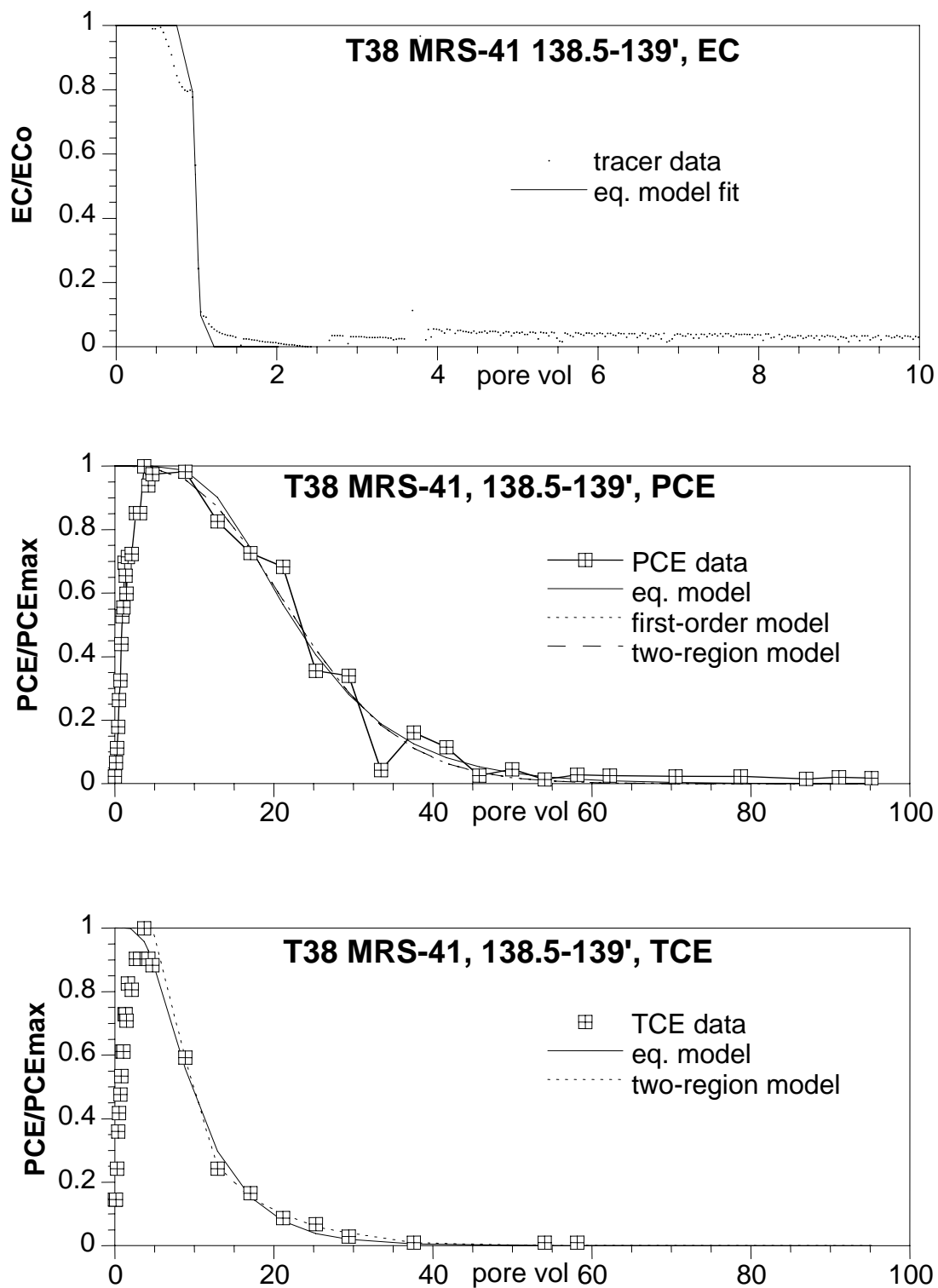


Figure 11. Simulation of Tracer (a) and PCE (b) for Experiment T31 (Note differences in pore volume scales.)

5.7.4 Experiment 39 (MRS-41, 139 ft to 139.5 ft) Simulations

PCE and TCE breakthrough data in experiment T39 showed essentially the same behavior as the MRS-41 sediment core from experiment T-38. The tracer data (Figure 12a) showed very little tailing and was well simulated with the equilibrium model. The PCE (Figure 12b) and TCE (Figure 12c) data also were well described by the equilibrium model, in which TCE and PCE concentrations decreased within another 20 pore volumes of the average breakthrough. Application of the first-order and two-region models did not significantly improve the fit. The 267 hours stop flow (Figure 8b, 8c) event at 60 pore volumes yielded no increase in PCE or TCE concentration in the effluent, indicating the absence of measurable TCE and PCE slow release from any TCE and PCE remaining on the sediment. This was affirmed from the fact that only trace levels TCE or PCE mass (e.g., PCE mass on sediment was 1 mg on the sediment versus 1,785.2 mg released in the effluent) was detected on the sediment after desorption. These results suggest that the irreversible sorption process was not significant in this sediment.

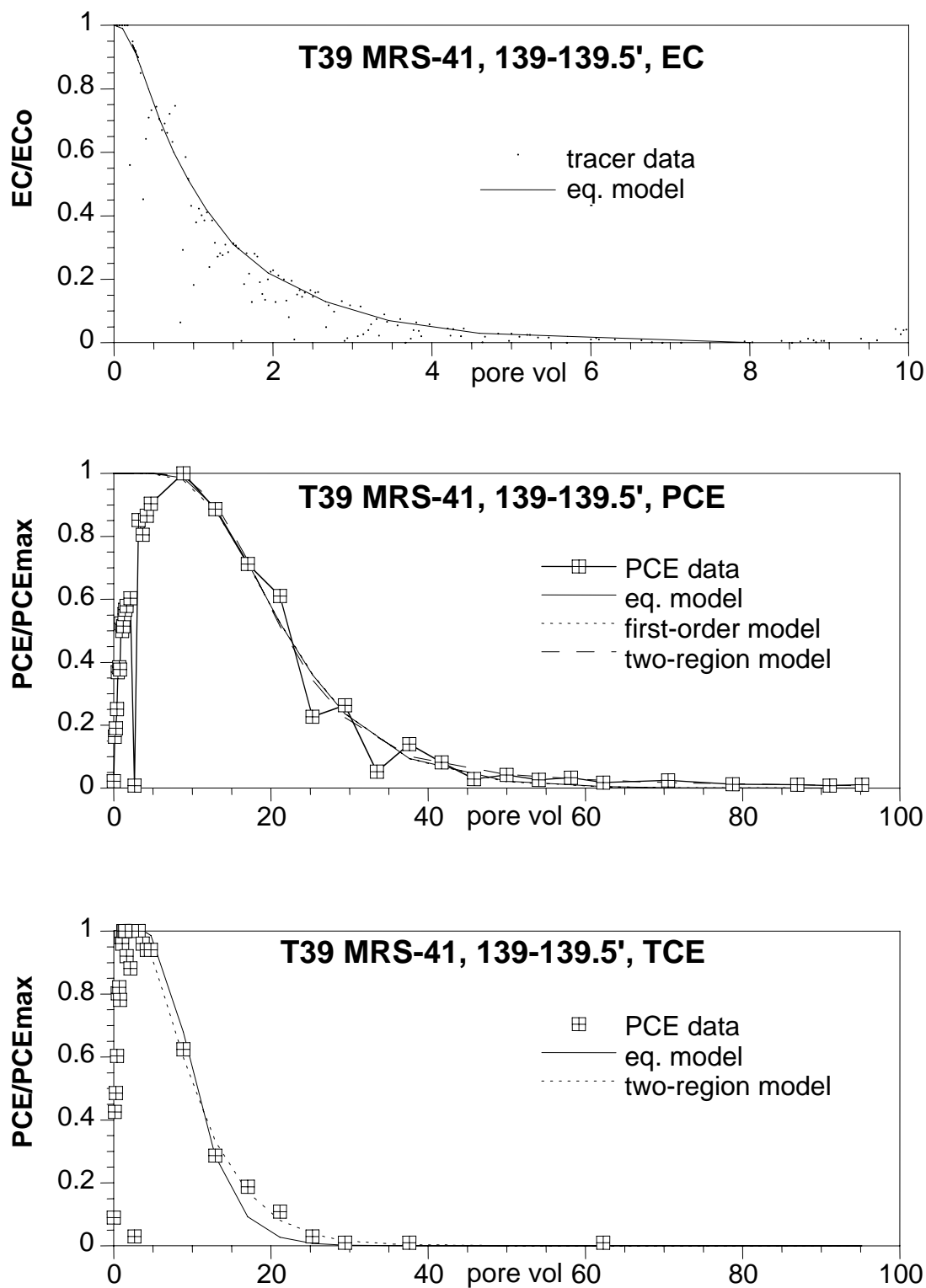


Figure 12. Simulation of Tracer (a), PCE (b), and TCE (c) for Experiment T39 (Note difference in pore volume scales.)

6.0 Discussion

This study produced the first values of TCE and PCE desorption K_d s based on measurements of TCE and PCE retardation factors in SRS contaminated aquifer sediments. TCE and PCE had been in contact with these sediments for decades. Such contact times are impracticable to reproduce in laboratory experiments. Values of TCE and PCE K_d calculated from measured retardation factors were 1.95 ± 0.01 L/kg ($n=2$) and 5.24 ± 0.13 L/kg ($n=2$), respectively for sediments from the A/M Area site. A K_d value of 0.12 ± 0.07 L/kg ($n=2$) was calculated for PCE in sediments from the P-Area site. Qualitatively, there was a positive correlation between the magnitude of K_d and clay content (i.e., the sediment with higher clay content showed the higher value of K_d). Concentrations of PCE and TCE in the sediments were well below the concentration that would indicate the potential presence of DNAPL, eliminating the presence of DNAPL dissolution as a process that could confound the K_d results. PCE K_d values differed significantly between the two sites, with the PCE K_d value at the A/M Area more than a factor of ten higher than that calculated at the P-Area site. Likewise, values of PCE and TCE K_d at the A/M site were significantly higher than have been reported for other low organic carbon and low surface area aquifer sediments at other contaminated sites in the United States and Canada. An explanation for these differences is unknown at this time (i.e., the lack of significant differences in organic carbon content and clay content of A/M versus P-Area sediments cannot account for these differences in K_d values). However, the results while limited, suggest the need to consider that sorption capacity may play a more important role in mass balance at SRS and that, the A/M Area may take longer to remediate. Estimated porewater concentrations of TCE and PCE in MRS-41 intact core sediments based on the high values of K_d compared reasonably well with concentrations of TCE and PCE measured in samples of groundwater from a well proximate to the site of sediment collection. This helps validate the higher than expected values of K_d determined for TCE and PCE from the desorption experiments.

K_d values of TCE and PCE estimated from sediment organic carbon content and organic carbon distribution coefficients (K_{oc}) or surface area and octanol-water distribution coefficients (K_{ow}) tended to underestimate those that were derived from experimental measurements of retardation factors from this study. Values of TCE and PCE K_d 's were also underestimated when a f_{oc} value of 0.01 and literature values of K_{oc} was used in calculating TCE (0.013 L/kg) and PCE (0.038 L/kg) K_d s (Looney et al. 1987). This suggests the need to consider, where possible, the acquisition and use of site-specific experimentally-derived data as opposed to data derived from estimation methods as a source of data for tools used in site assessments. Caution should be exercised in applying K_d s derived from estimation methods in software tools addressing questions in the context of natural attenuation (e.g., time to remediate) (Chapelle et al. 2003), monitored natural attenuation assessment (e.g., *MNAtoolbox*- Brady et al. 1999) or numerical models (e.g., RT3D) that incorporate key monitored natural attenuation/enhanced attenuation processes (Clement 1997; Clement et al. 1998).

Values of TCE and PCE K_d in aquifer sediments from previously reported studies have been shown to be greater than those estimated from key parameters (i.e., f_{oc} , K_{oc}), consistent with the findings of this study (Ball and Roberts 1991; Curtis et al. 1986; Piwoni and Banerjee 1989). Most significantly, the high TCE and PCE K_d values calculated based on experimental measurement of retardation factors for A/M Area sediments suggests a need to be aware of the possibility of underestimation of the time required for remediation of these sediments based on use of estimated values of K_d or values that have been historically recommended (Looney et al. 1987). For example, previous work has shown that cleanup time (i.e., using pump-and-treat) is approximately 10 times longer for $K_d = 1$ than the non-reactive case (i.e.,

where $K_d = 0$), and has additional impact for K_d increases up to a value of 5 (Berglund and Cvetkovic 1995). A/M sediments may become a secondary source of TCE and PCE release to groundwater in adjacent sediments of higher permeability (Bear et al. 1994).

All of the TCE and PCE mass (100%) from desorption of sediments from the A/M Area site (MRS-41) was found to be present in the effluent, providing evidence for the absence of TCE or PCE migration resistant fractions in these sediments. Simulation of the experimental tracer and PCE breakthrough curve data showed a good fit with an equilibrium model, consistent with reversible sorption influencing TCE and PCE transport in these sediments. Coupled with the mass distribution data, this suggests the absence of influence of a significant irreversible sorption process. Subsequent to the stop flow component of the MRS-41 experiments, neither TCE or PCE was detected in effluent following resumption of desorption providing further evidence for the absence of migration resistant fractions in these sediments.

In contrast, solute desorption from the two PGRS-1 sediment cores showed some profile tailing in both experiments, indicative of slow release of PCE. Simulation of PCE breakthrough curve data with an equilibrium model could not match the profile tailing observed, and accounted for only 66% of PCE breakthrough mass. Also, a first-order model could not match the tailing behavior observed in the profile. The two region model was able to fit the observed breakthrough data, consistent with the hypothesis that some PCE mass was being released at a moderately slow rate from sediments to solution (irreversible sorption) but not from a significant source of mass remaining in the sediment as a migration resistant fraction. One interpretation of the two-region model results is that most PCE (80%) is being rapidly released from the sediment and a small fraction of PCE (20%) is being slowly released (desorption half-life 320 h). This two sorption site representation could account for the fast desorption of the bulk of PCE mass (66%) and also the slow desorption of the remaining (34%) PCE mass in the tailing portion of the profile. Slow desorption could be chemical in nature (i.e., rate controlled by an actual solute-surface phase chemical reaction) or physical (i.e., slow diffusion out of sediment microfractures, for example).

Sediment samples from the SRS P-Area contained significantly higher percentages of particles in the greater than 2-mm-size range as compared to SRS sediment from the A/M area (12.3% and 24.3% versus 1.9%). Previous work has shown the coarse grain fractions of low organic carbon aquifer sediments sorb PCE (Ball and Roberts 1991) and other chlorinated hydrocarbon compounds (Zhao et al. 2005) to an extent that was disproportionate to its mass fraction in direct contrast to the commonly held preconception that sorption capacity is inversely related to particle size (Bishop et al. 1989). The coarse grain fractions contained a higher percentage of the organic carbon content of the sediments and showed higher values of K_d than the finer grained fractions of the same sediments. Assuming organic carbon and PCE of P-Area sediment samples may be disproportionately concentrated in the larger particle fraction of these sediments may explain the observed irreversible sorption behavior in P-Area sediments but not in the A/M-Area sediments.

The observed behavior of TCE and PCE in SRS sediments is in contrast to mass fractions of chloroform (CHCl_3) and to a limited extent carbon tetrachloride (CCl_4) found resistant to migration (MRF) in a U.S., DOE Hanford Site aquifer sediment (Riley et al. 2005). The experimental environment was the same for both CCl_4 and CHCl_3 in the Hanford study. Therefore, differences in the observed retention behavior of CCl_4 and CHCl_3 (i.e., MRF observed more often and in higher amounts than CCl_4 MRF) have to reside in the structural (one chlorine atom replaced by a hydrogen atom) and chemical (non-polar CCl_4 versus relatively polar CHCl_3) differences in these molecules and the way they interact with sediments of low organic carbon content. Single solute isotherm studies have shown a number of soils to adsorb

appreciable quantities of CHCl_3 in comparison to CCl_4 suggesting that inorganic (mineral) constituents of the soil are more selective to polar compounds, providing support to the findings of the Hanford study (Peng and Dural 1998).

The physical/chemical properties of Hanford and Savannah River site sediments are similar (i.e., low organic carbon content, low surface areas, particle size distributions). The absence of mass fractions of TCE and PCE resistant to migration in SRS sediments suggest that we need to look at the differences in the structural and chemical properties of these compounds (i.e., CHCl_3 versus CCl_4 , TCE, and PCE) in explaining why CHCl_3 and to a small extent CCl_4 can exist as an irreversible mass fraction in sediment but TCE and PCE do not.

7.0 Recommendations

We recommend the conduct of additional intact core desorption experiments incorporating a stop-flow step that is implemented earlier in the experiments to obtain kinetics data on irreversible sorption behavior. Such experimentation could also be extended to key TCE/PCE degradation products. Such data would be important in populating the expanded parameter capability of numerical models (e.g., RT3D). To improve the understanding of K_d variability among SRS sites and associated uncertainty, we recommend determination of TCE and PCE K_d s on additional sediments. A cost-effective approach might be short contact time batch experiments as opposed to intact core desorption experiments.

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