

***PNNL SFA: Role of Microenvironments and Transition Zones in
Subsurface Reactive Contaminant Transport***

Reactive Transport of U and Tc in Heterogeneous Sediment Systems

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<http://www.pnl.gov/biology/sfa/>

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Reactive Transport of U and Tc in Heterogeneous Sediment Systems

Objective

- ▶ Determine how heterogeneities in pore scale environments (mm-cm) and properties zonation (cm-m) influence the biogeochemical transport of U and Tc in field-relevant sediment systems.

Scope

- ▶ Integration experiments leading to field; coupled reaction and transport processes; biotic and abiotic oxygen consumption, Tc redox reactions, and Fe(II) minerals; U(VI) surface complexation and precipitation/dissolution in varied physicochemical domains. Heterogeneity effects, pragmatic scaling concepts, and reactive transport modeling.

Research Team and Expertise

John Zachara (*PNNL*) - Geochemistry, biogeochemistry, and mineralogy

Mart Oostrom (*PNNL-EMSL*) - Hydrology and reactive transport science

Jim McKinley (*EMSL*) - Electron microscopy and microprobe

Steve Heald (*ANL-APS*) - Tc x-ray spectroscopy, microscopy, and diffraction

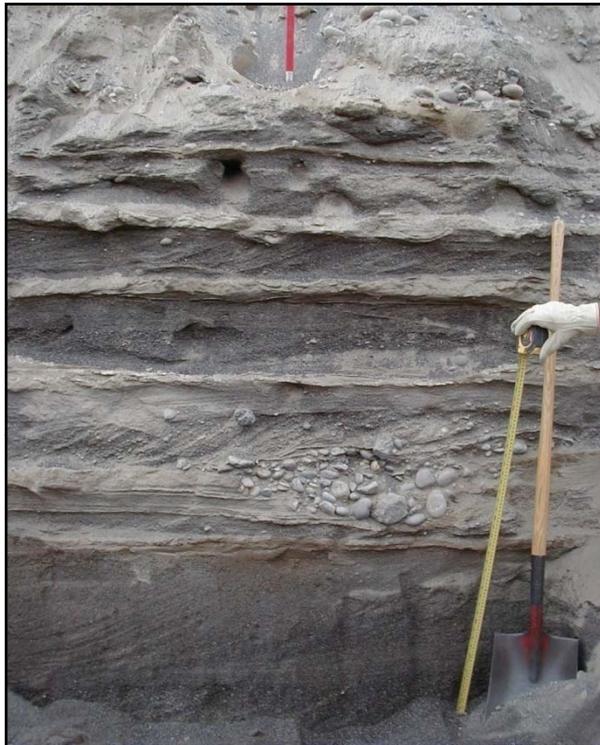
Jim Davis (*USGS*) - Uranium geochemistry and microscopic reactive transport

Phil Jardine (*ORNL*) - Mesoscale reactive transport experiments

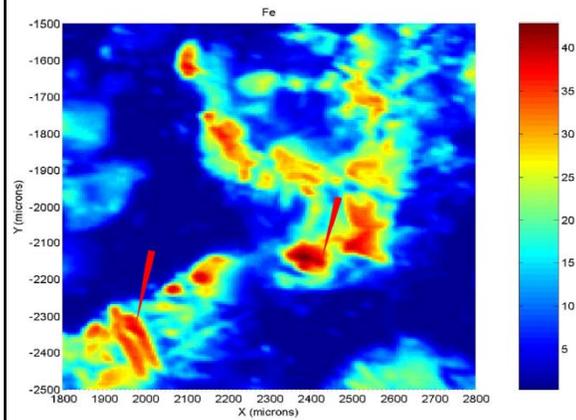
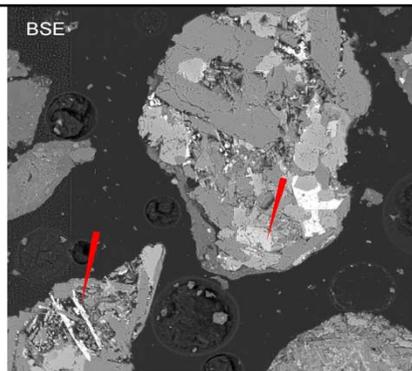
Chongxuan Liu and Mark Rockhold (*PNNL*) – Reactive transport modeling through SFA and IFRC collaboration



Hanford Science Motivation



- ▶ Reactive transport vectors run both normal and parallel to complex sediment stratigraphy
- ▶ Contaminants experience different reaction networks and flow velocities in different layers.
- ▶ In-layer reactivity is complex because of pore-scale variations.
- ▶ Predicting system level breakthrough requires understanding of integrated behavior.



- ▶ Hanford sediments contain complex Fe mineralogy that is not well understood.
- ▶ Over 50% is Fe(II) with high potential for reactivity with oxidants [O_2 , Tc(VII)].
- ▶ Hanford (~15 K bp) and Ringold (~120 K bp) formations display different weathering extents and lithic fragment distributions.
- ▶ Large repository of Fe(II/III) minerals in lithic fragment interiors with variable fractures and porosities.

Research Structure and Logic

- ▶ Address various heterogeneities leading to field, with IFRC as emphasis
- ▶ Strong linkage with other projects is required
- ▶ Emphasis on macroscopic and mesoscopic experiments linking reaction and transport (batch, flow-cell, column, tank/block)
- ▶ Focus on select/key contaminant reaction scenarios (U, Tc)

Field	Intact Sediments • Uncontrolled heterogeneities	Model Structures • Controlled heterogeneities	Field Textured Materials • Pore scale heterogeneity	Particle Size Fractions • Particle heterogeneity
<ul style="list-style-type: none"> ■ 300 A IFRC ■ 200 A 	<ul style="list-style-type: none"> ■ Influence of depositional features ■ In-situ organism distributions ■ Complex advection flow paths ■ Impacts of in-situ associations of all type ■ Comprehensive data sets for understanding and model application 	<ul style="list-style-type: none"> ■ Function of complex physical, chemical, and microorganism structures or distribution ■ Multi-scale mass transfer between advective regions ■ Mass transfer controlled kinetic effects ■ Kinetic transport models 	<ul style="list-style-type: none"> ■ Basic parameter upscaling ■ Geochemical additivity effect ■ Biogeochemical processes ■ Hydraulic effects of gravel ■ Macroscopic transport model 	<ul style="list-style-type: none"> ■ Size/mineralogy effects on reactivity and mass transfer ■ Behavior of <2 mm fraction and controlling variables ■ Microscopic to macroscopic reaction model



Deconstructive Experimentation



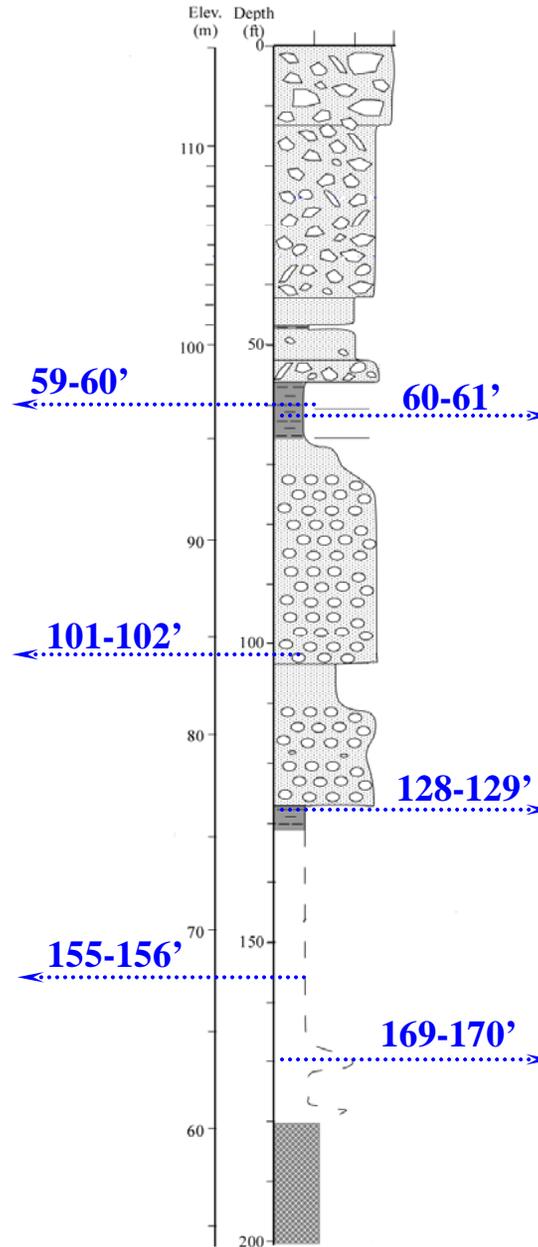
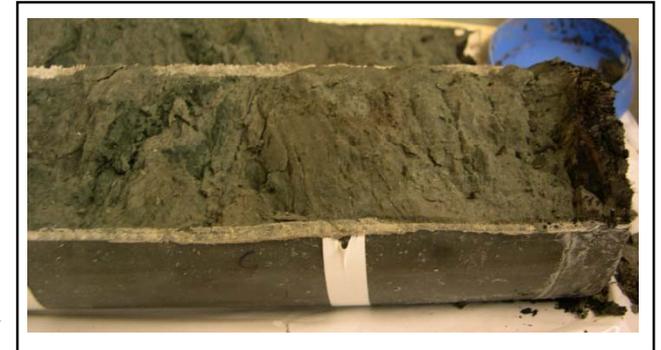
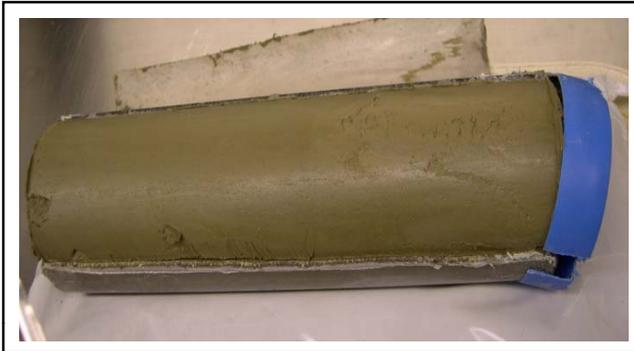
Reactive Transport in Heterogeneous Sediments

Two Primary Lines of FY09 Research

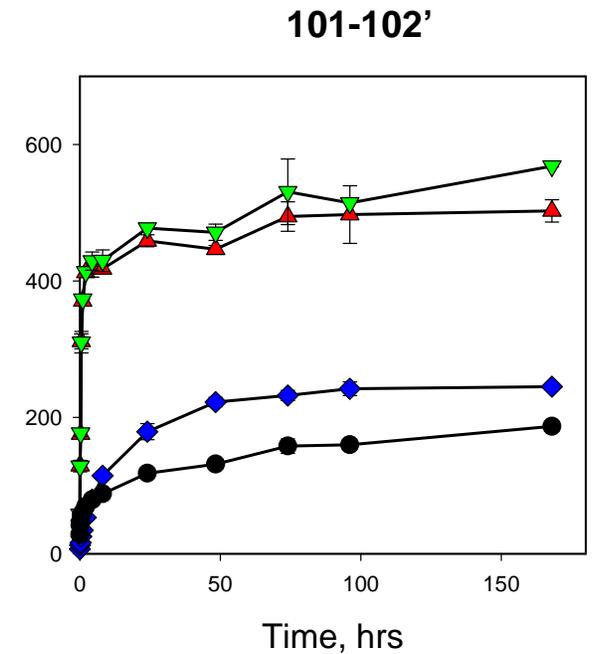
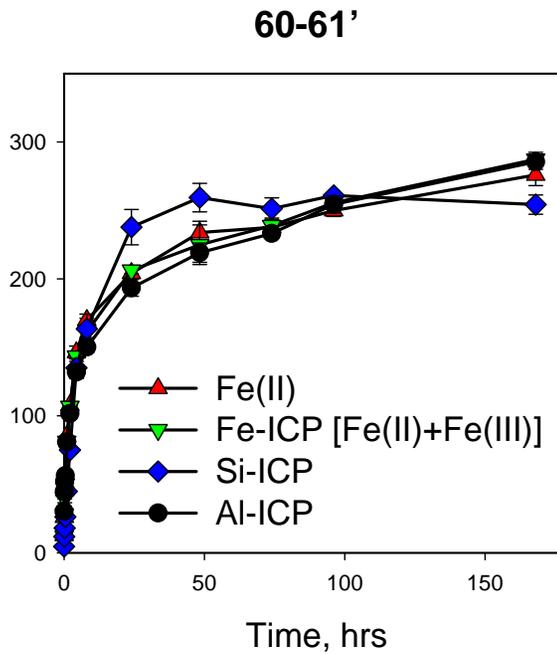
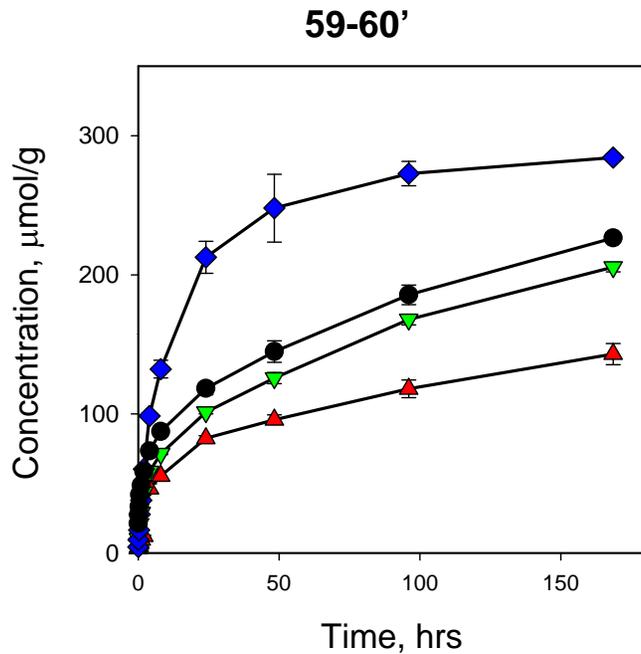
- ▶ Tc(VII) reaction kinetics (FY09, 40%)
 - 300 A Ringold/IFRC redox transition zone
 - Correlation of Tc(VII) and O₂ reduction rates (batch and flow cell)
 - XAS (ID, BM), SEM/TEM, and TMS for spatial distribution, location, molecular speciation, and Fe mineralogy
(Collaborations: Fredrickson, ISFA; Rosso, ISFA)

- ▶ Intact column experiments with 300 A IFRC site sediments in EMSL-SFTF (FY09, 45%)
 - 3 intact Hanford Formation cores, near WT and center of SZ
 - U(VI) desorption, tracer behavior, U(VI) adsorption/desorption
 - Core deconstruction with subsequent characterization and “mechanistic” study of < 2 mm fraction
 - Multi-scale reactive transport modeling
(Collaborations: Liu, Scheibe/Rockhold, Ward, Jardine, and Davis)

IFRC Transition Zone Samples (C6209)



Weak Acid Extraction of Fe, Al, Si



I. 59-60'

$[\text{Fe(II)}] < [\text{Fe-ICP}] < [\text{Al-ICP}] < [\text{Si-ICP}]$

II. 60-61', 128-129', 155-156'

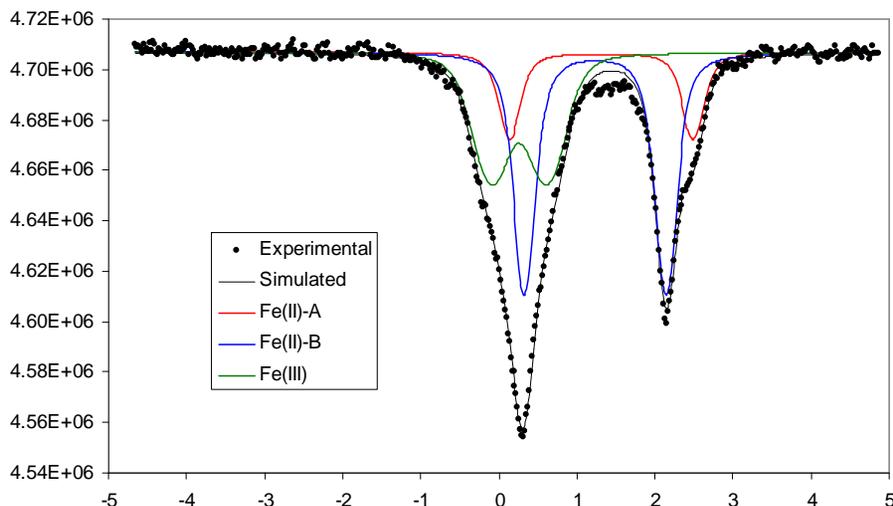
$[\text{Fe(II)}] = [\text{Fe-ICP}] = [\text{Al-ICP}] = [\text{Si-ICP}]$

III. 101-102', 169-170'

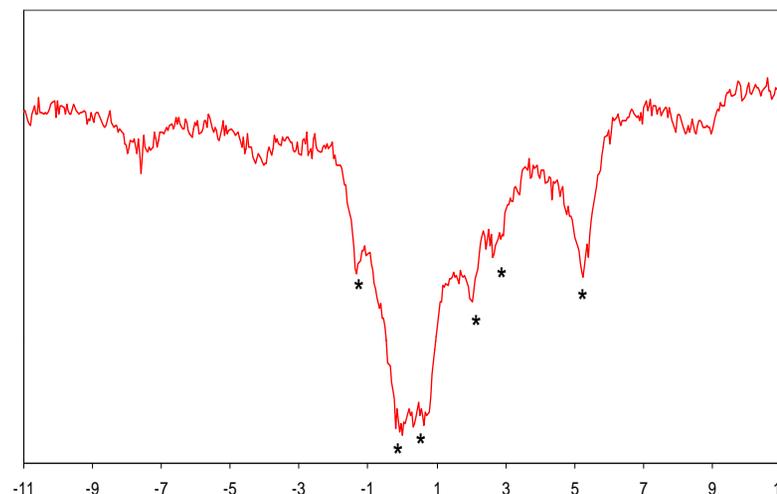
$[\text{Fe(II)}] = [\text{Fe-ICP}] > [\text{Si-ICP}] > [\text{Al-ICP}]$

Variable Temperature Mössbauer Spectroscopy for Fe Mineral Forms (101-102')

Room temperature (RT) *Preliminary fit*



4.2 K



RT

Fe(III) and two Fe(II) species

77 K and 45 K
virtually identical

no Fe(III) ordering
no Fe(III) oxides

no Fe(II) ordering
no siderite (FeCO₃)

12 K

no Fe(III) ordering

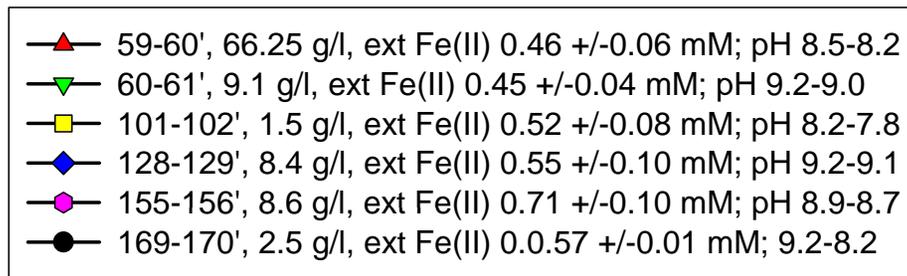
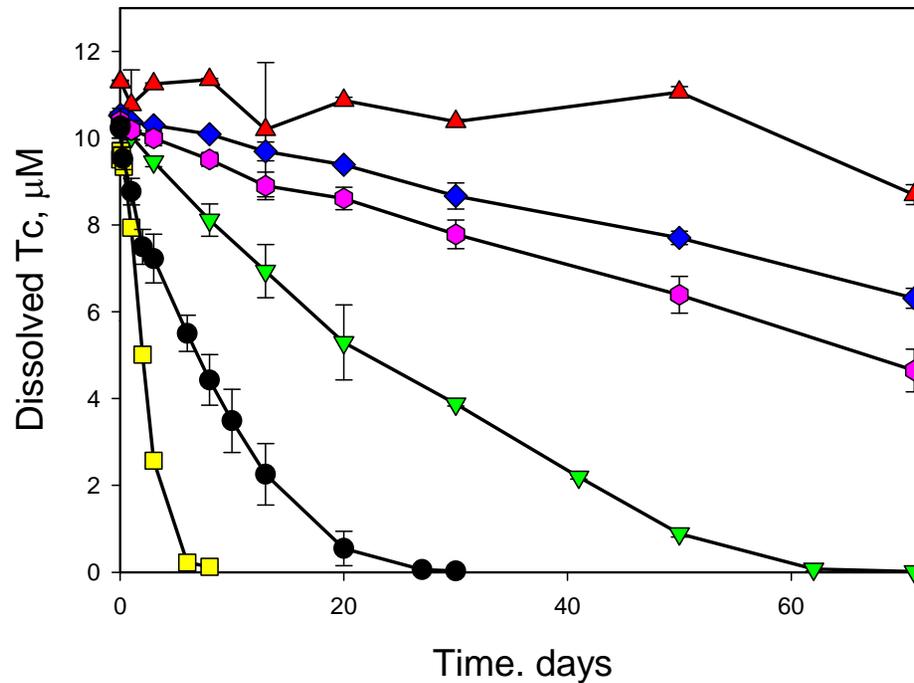
ordering of one of Fe(II) species
Fe(II) separate phase (?)

4.2 K

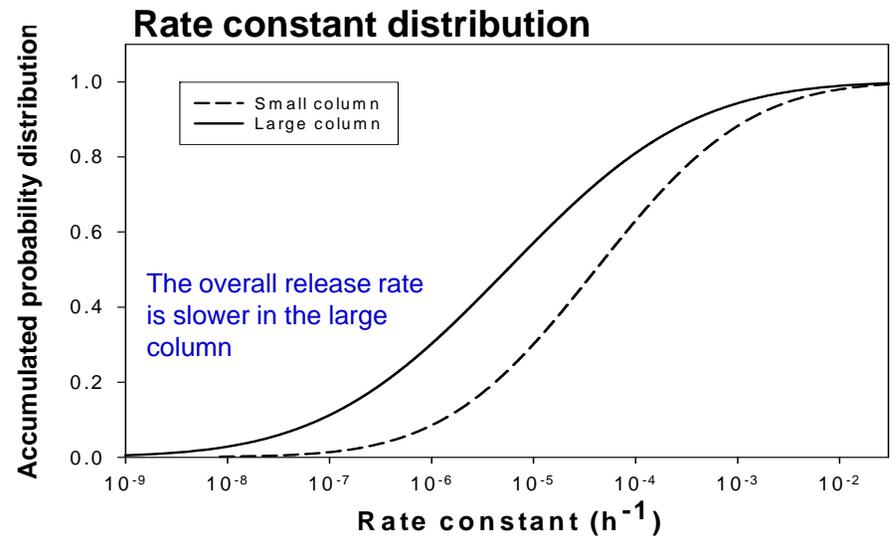
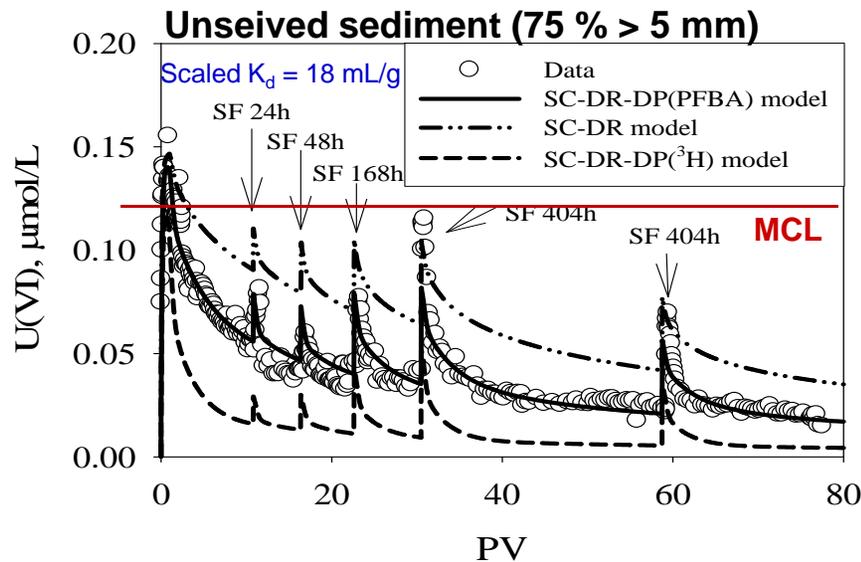
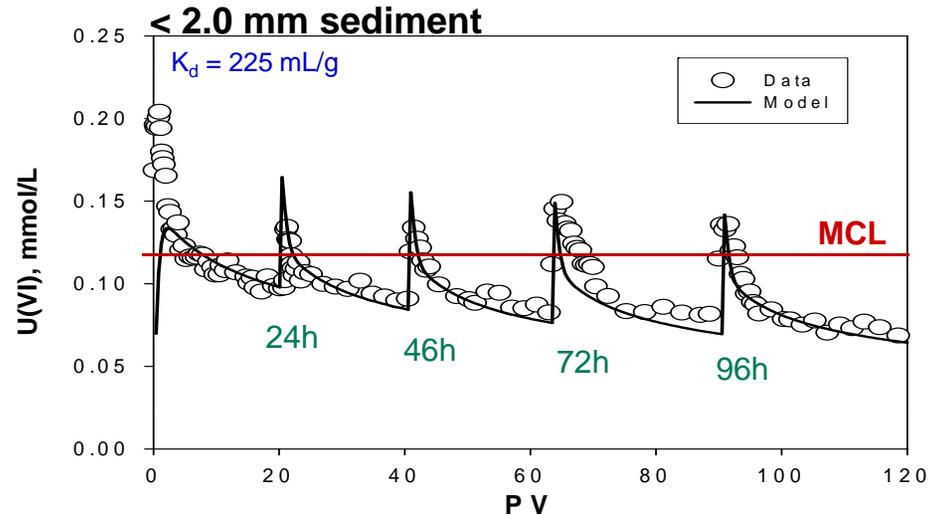
Fe(III) ordering

Fe(II) ordering

Tc(VII) Reduction: Constant Fe(II) [all C6209 samples]



Mass Transfer: Importance, Modeling, and Scaling



Determination of Ex-Situ/In-Situ Reaction Parameters

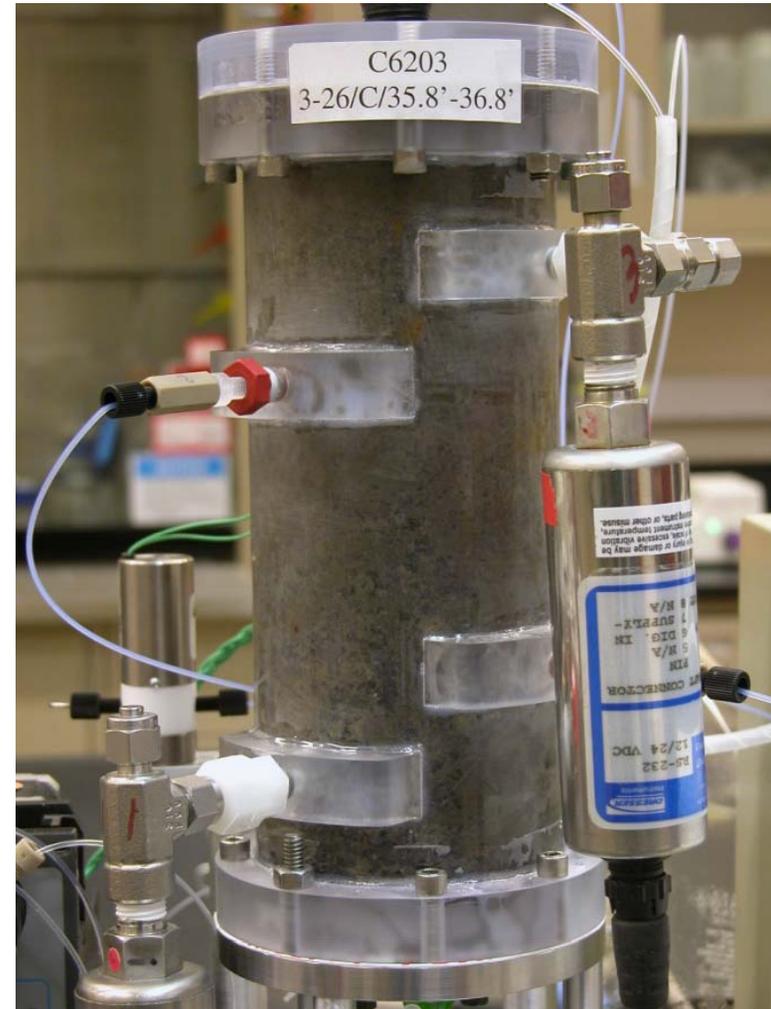
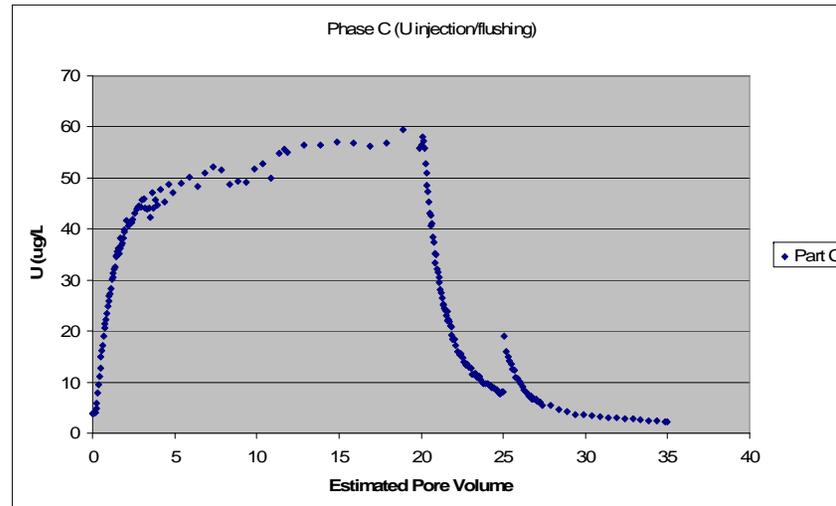
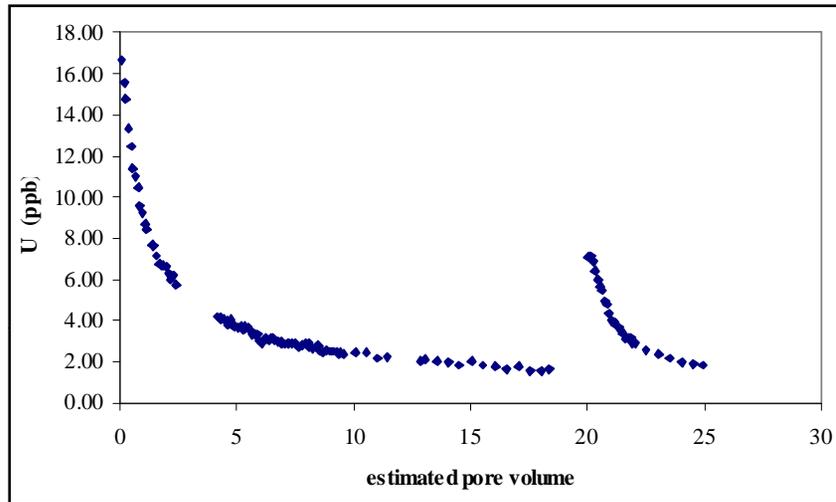
Intact Column Studies

- ▶ RT experiments at 10'/d, 3 cores from different aquifer depths
 - A-desorption, B-tracer, C-adsorption/desorption SGW-1, D-hydrologic properties
 - Contaminant versus spiked U(VI)
 - Unique structured behavior

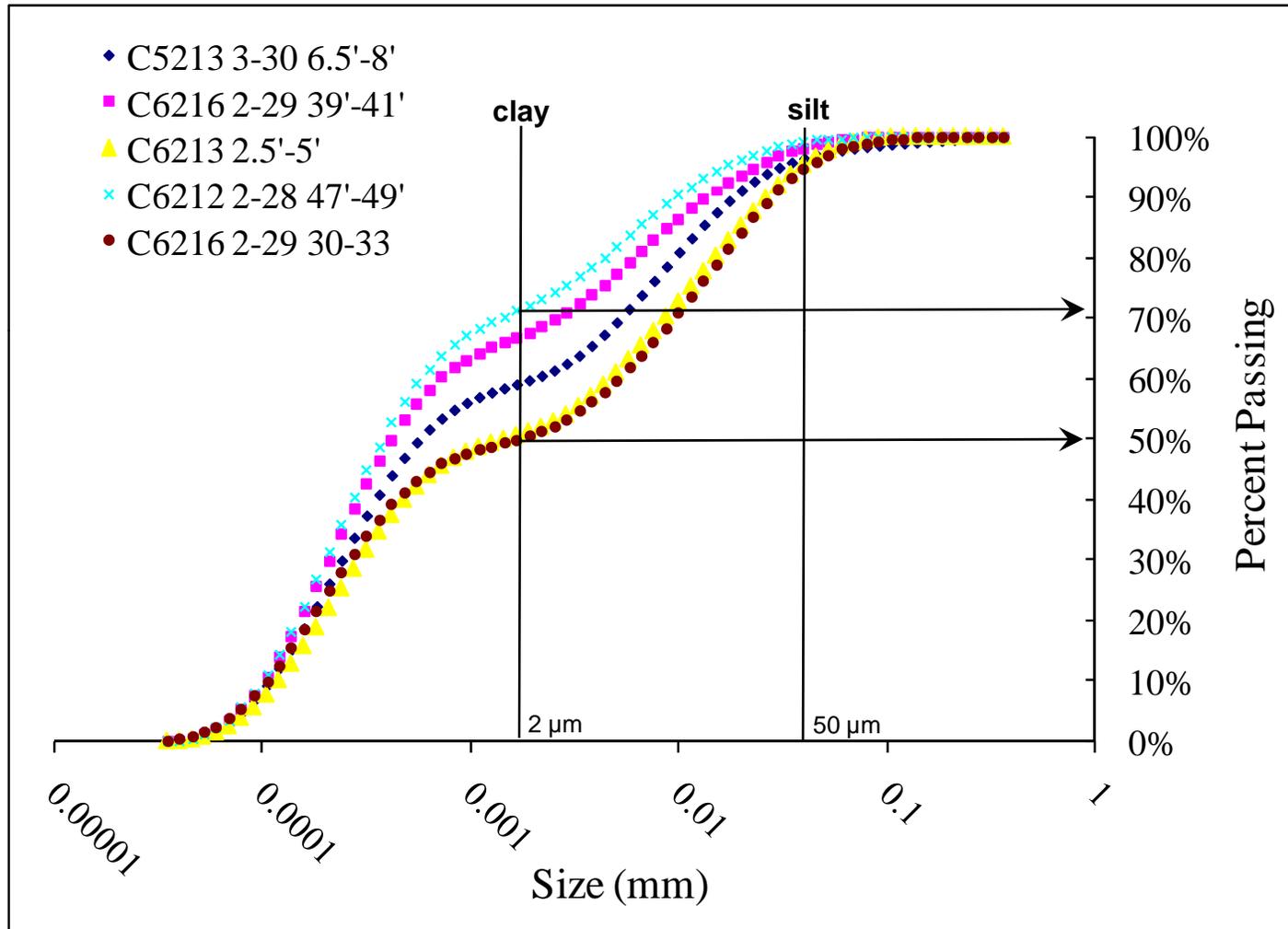
- ▶ Deconstruction
 - PSD, SA (< 2 mm), < 2 μ m mineralogy
 - Low [U(VI)] isotherm (< 2 mm) with two SGW
 - Mass transfer parameters by flow cell (< 2 mm)
 - AAO Fe(III) (< 2 mm)
 - CLIFS spectroscopy

- ▶ Upscaled modeling
 - SC-DR-DP (PFBA)
 - Other

U(VI) Desorption and Adsorption/Desorption from Intact IFRC Core 3-26/35.8-36.8'



Particle Size of Coatings on Gravel (32 mm)

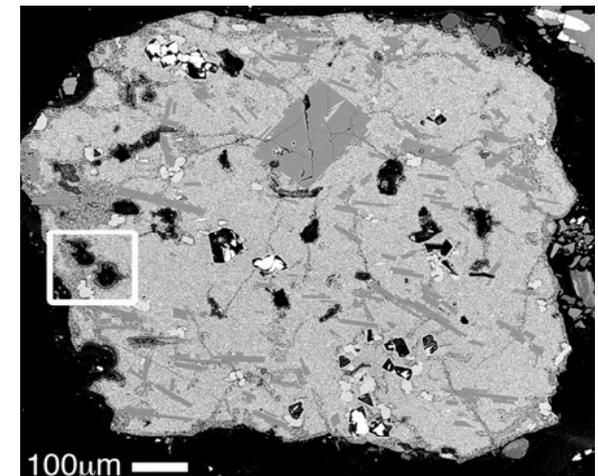
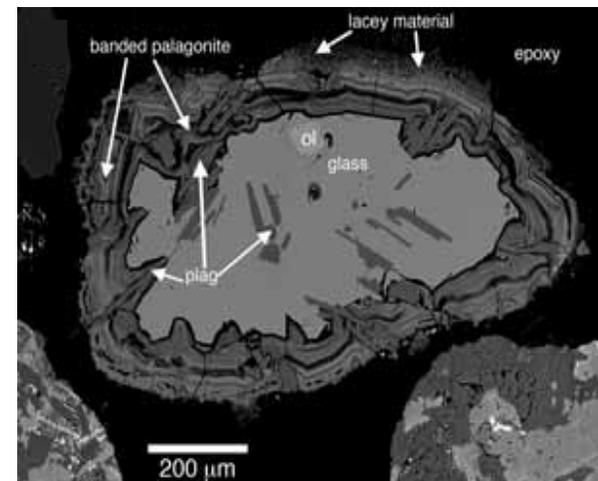


ESFA Collaborations

Davis – Particle heterogeneity

Objective: Quantify impact of intergranular porosity (IGP) on rates of U(VI) adsorption/desorption

- ▶ IGP of different sediment fractions and responsible phases w CL ISFA
- ▶ XRM studies of tracer in-diffusion w KK ESFA
- ▶ Species specific diffusion rates, cross diffusion, and effects in diffusion w CL ISFA
- ▶ Microscopic reactive transport modeling w CL ISFA and MR IFRC
- ▶ Focus on Hf (300 A), RF (300 A), RF (200 A)
- ▶ Paper from FY08 in progress



ESFA Collaborations

Jardine – Macroscopic heterogeneities and unsaturated conditions (mm to dm scale)

Objectives: Determine how dual porosity develops with gravel content and provide experimental basis for upscaling < 2 mm reaction parameters

- ▶ Reactive transport experimentation under unsaturated and saturated conditions
- ▶ Initial target is the role of gravel on U adsorption/desorption at the 300 A IFRC
 - Quantify reactivity of different < 2 mm fractions with varying volume fractions of gravel
 - Vary flow rate, water composition, and U(VI) concentration
 - Investigate transition between equilibrium and kinetically dominated reactions and causes
- ▶ Requires modeling support; other issues



FY09 Science Deliverables

- ▶ Relationship of oxygen consumption and Tc(VII) reduction rates to Fe(II) mineralogy of Hanford sediments.
- ▶ Spectroscopically based surface complex reaction model for U(VI) in 300 A saturated zone sediments.
- ▶ U(VI) reactive transport behavior in naturally structured 300 A sediments, role of kinetic and thermodynamic processes.
- ▶ Modeling approach to scale < 2 mm U(VI) reaction parameters (rate constants, site concentrations, diffusivities) to whole sediment (with others).

FY10 Research Scope

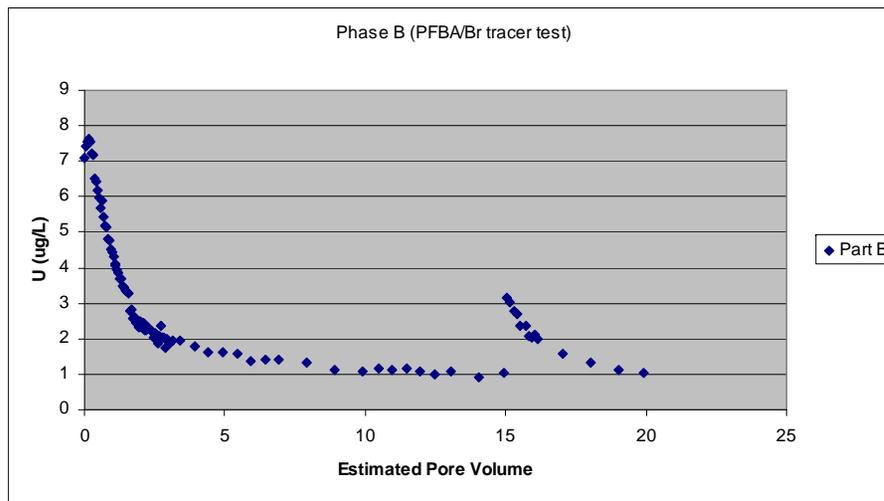
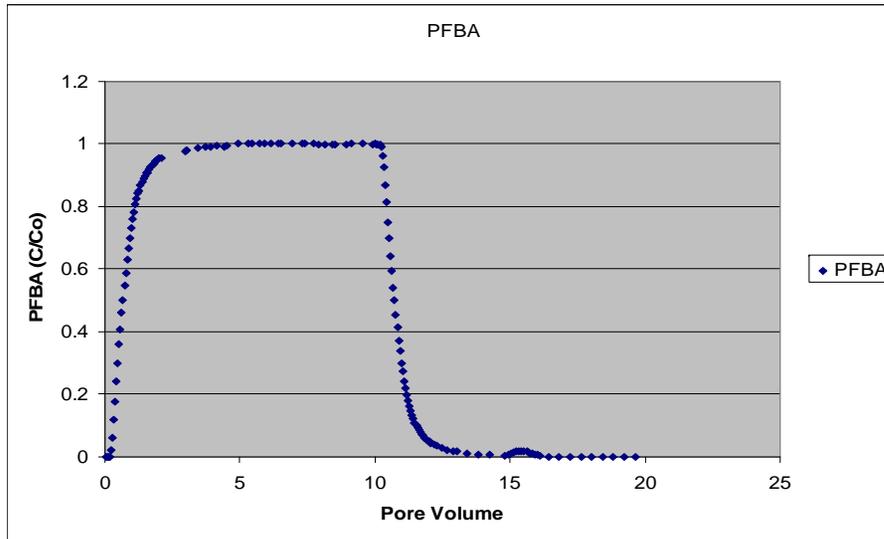
- ▶ Continued studies on Hanford Fe(II) mineralogy and heterogeneous Fe(II)-promoted reactions
- ▶ Mineral-microbe studies on Fe(II) mineralogy, oxygen consumption and bacterial Fe(II) oxidation, Tc(VII) reduction, and U(VI) adsorption
- ▶ Influence of gravel on adsorption-desorption or dissolution rates with U(VI) in known speciation states/concentration [Fe(III) oxides, chlorite, and weathered mica; or metatorbernite]
- ▶ Reactive transport experiments with model transition zones and interfaces (Hanford-Ringold contact, Ringold redox transition)

Expected Science Accomplishment and Impact

- ▶ Cross scale reactive transport experimentation that identifies critical features, properties, and parameters controlling Tc and U migration behavior from microscopic to mesoscopic scale for important Hanford scenarios.
- ▶ Improved understanding of key factors controlling process-coupling at different scales.
- ▶ Experimentally based modeling approaches to describe U and Tc migration in naturally complex materials with heterogeneities of different size, type, and orientation.

Impact: Risk reduction, insights for remediation, improved transport simulations that properly describe processes, and account for heterogeneities at different scales.

Tracer Behavior and U(VI) Desorption During Phase B



Science Issues and SFA Hypotheses

Science Problems

- ▶ Reactivity of different ferrous iron mineral types in varied structural/pore environments. Locations, scales, and effects of O₂ consumption on Tc and U transport.
- ▶ Importance of in-situ depositional features, structures, and natural mineral associations on redox, complexation, and precipitation/dissolution processes.
 - Reaction issues - biotic versus abiotic effects and their locations
 - Spatial issues - locations of water flow versus reaction
 - Transport issues - exchange between zones of different velocities
- ▶ Scaling process understanding to the field.
 - Critical process-level information, parameters, or abstractions that must be preserved
 - Effects of different heterogeneity types
 - Characterization/imaging of essential in-situ features and properties

SFA Hypotheses

- ▶ Reduction reactions will occur where oxygen consumption occurs by ferrous iron mineral reaction or microbiologic activity.
- ▶ Contaminant adsorption and precipitation will occur in grain interiors, lithic fragments, and fine-grained sediment bodies. Mass transfer will regulate contaminant exchanges between these zones and advective conduits.
- ▶ Understanding effects of in-situ features and geologic heterogeneities critical for lab-field transfer.

